Powder diffraction data for methylenedioxymethylamphetamine hydrochloride monohydrate (MDMA.HCI.H₂O, Ecstasy hydrate)

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Powder diffraction data are presented for 3,4-methylenedioxymethylamphetamine hydrochloride monohydrate, the hydrated form of MDMA.HCl, the drug of abuse commonly known as Ecstasy. Samples of pure MDMA.HCl were recrystallized from a variety of solvents, and powder diffraction patterns of the resulting products recorded. Data were collected at room temperature using iron-filtered Co*K* α radiation. © 2012 International Centre for Diffraction Data. [doi:10.1017/S0885715612000693]

Key words: methylenedioxymethylamphetamine hydrochloride monohydrate, Ecstasy, MDMA.HCl. H₂O, forensic, drug, XRD

I. INTRODUCTION

Methylenedioxymethylamphetamine hydrochloride, commonly known as MDMA.HCl or Ecstasy, is a well-known drug of abuse and is available in powdered and tablet form as anhydrous salt (Figure 1). A number of hydrated crystalline forms also exist, and Shulgin (1986) has referred to the melting points of five individual phases: anhydrous salt, quarter hydrate, hemihydrate, three-quarter hydrate and monohydrate. The final form assumed by the salt during crystallization depends on temperature and concentration in the initial stages of crystallization (Shulgin and Shulgin, 1992).

X-ray powder diffraction (XRD) is used in many forensic science laboratories for the analysis of drugs of abuse, together with their adulterants and excipients. Analysis of these mixtures is occasionally complicated by the presence of seemingly unidentified material. Powders containing Ecstasy are no exception, and if their analyses show them to be free from precursors and common adulterants or excipients, then consideration should be given to the possibility that the contaminant is either a polymorph or a hydrated form of the drug itself.

Single-crystal data have been collected for both the anhydrous (Morimoto *et al.*, 1998), PDF 02-070-4055 (ICDD, 2012) and monohydrate (Zapata-Torres *et al.*, 2008), PDF 02-098-9765 (ICDD, 2012), forms of MDMA.HCl. The only experimental pattern in the Powder Diffraction File is that of the anhydrous form, PDF 00-039-1560 (ICDD, 2012). Consequently, attempts to produce crystals of the monohydrate of MDMA.HCl have been undertaken to generate high-quality X-ray powder data.

II. EXPERIMENTAL

A. Preparation

Attempts to grow crystals of the monohydrate were made using four solvents: (a) deionized water, (b) methanol, (c) acetone, and (d) chloroform. Approximately 200 mg pure MDMA.HCl was dissolved in the minimum volume of solvent in small porcelain crucibles. A lid was placed on each crucible to protect the solutions from airborne contamination, and the solutions were allowed to evaporate at room temperature $(20 \pm 1 \text{ °C})$. When the solvents had completely evaporated, the resulting crystals were hand-ground for 5 min using an agate mortar and pestle. The finely powdered samples were side-loaded for powder diffractometry using the technique of McMurdie *et al.* (1986).

B. Data collection

X-ray data were collected using a Philips PW1050/37 vertical diffractometer in $\theta/2\theta$ mode with Fe-filtered CoK α radiation ($\lambda = 1.78897$ Å) from a Philips long fine focus tube powered at 35 kV and 42 mA. Fixed slits (1° divergence, 1° scatter, and 0.2 mm receiving) and a stationary sample holder were used. A scan range of 3–63° 2θ was used with a step size of 0.05° 2θ and a count time of 1 s at each step. Data were collected at 20 °C.

III. RESULTS AND DISCUSSION

The X-ray powder patterns obtained are shown in Figure 2. The powder pattern for the sample of MDMA.HCl used as starting material (Figure 2a) was, as expected, identical to that reported in PDF 00-039-1560 (ICDD, 2012). This will be referred to as Form I.

The XRD patterns of the samples recrystallized from methanol (Figure 2b) and acetone (Figure 2c) were identical in terms of *d*-spacing, but with slightly differing relative intensities, probably owing to a degree of preferred orientation. The overall diffraction pattern is, however, quite different from that of the starting material, Form I. This compound will be referred to as Form II. The patterns of the samples recrystallized from water (Figure 2d) and chloroform (Figure 2e) are clearly owing to mixtures of Forms I and II.

The powder data from Form II were indexed and unit cell parameters determined using DICVOL04 (Boultif and Louer, 2004). The Smith–Snyder (Smith and Snyder, 1979) index F_N

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Figure 1. Structural diagram of anhydrous MDMA.HCl (Ecstasy).

for Form II is $F_{20} = 70.0(0.0092, 31)$. These experimental unit cell parameters agree closely with those determined from the single-crystal study of MDMA.HCl.H₂O (Zapata-Torres

et al., 2008) suggesting that Form II is indeed the monohydrate of MDMA.HCl.

Crystal data are as follows:

3,4-methylenedioxymethylamphetamine hydrochloride monohydrate

 $C_{11}H_{15}NO_2.HCl.H_2O$ F.W. = 247.71

Monoclinic, space group $P2_1/n$ (No. 14)

a = 7.2457 (12), b = 20.8132 (38), c = 9.1857 (17) Å, β = 108.257 (19)°

 $V = 1315.53 \text{ Å}^3$, Z = 4, $D_x = 1.251 \text{ (g cm}^{-3})$.

Table I shows $2\theta_{obs}$, d_{obs} , I_{obs} , hkl, $2\theta_{cal}$, and $\Delta 2\theta$ for 3,4-methylenedioxymethylamphetamine hydrochloride monohydrate.



Figure 2. Powder diffraction patterns of: (a) pure anhydrous MDMA.HCl (Form I), (b) MDMA.HCl recrystallized from methanol (Form II), (c) MDMA.HCl recrystallized from acetone (Form II), (d) MDMA.HCl recrystallized from water (mixture of Forms I and II), and (e) MDMA.HCl recrystallized from chloroform (mixture of Forms I and II).

TABLE I. XRD data for MDMA.HCl.H₂O (CoKα).

$2\theta_{\rm obs}$	$d_{\rm obs}$	Iobs	h k l	$2\theta_{\rm cal}$	$d_{\rm cal}$	$\Delta 2\theta$
9.879	10.388	14	020	9.898	10.369	-0.019
12.819	8.013	19	011	12.803	8.023	0.016
15.409	6.672	33	021	15.414	6.670	-0.005
15.789	6.512	290	1 1 0	15.775	6.518	0.014
15.909	6.464	206	$-1 \ 0 \ 1$	15.907	6.464	0.002
16.659	6.174	97	$-1\ 1\ 1$	16.664	6.173	-0.005
17.969	5.728	112	120	17.967	5.728	0.002
18.749	5.491	426	-121	18.755	5.490	-0.006
18.989	5.423	225	031	18.999	5.420	-0.010
19.829	5.195	20	040	19.833	5.194	-0.004
21.119	4.881	26	130	21.135	4.877	-0.016
21.809	4.728	638	101	21.838	4.722	-0.029
22.399	4.605	403	111	22.401	4.605	-0.002
23.129	4.462	306	041	23.129	4.462	0.000
23.719	4.352	999	002	23.704	4.355	0.015
24.239	4.260	684	-112	24.280	4.253	-0.041
24.939	4.143	16	140	24.928	4.144	0.011
25.759	4.013	317	-122	25.783	4.009	-0.024
26.489	3.904	502	131	26.499	3.903	-0.010
27.609	3.749	41	051	27.584	3.752	0.025
28.129	3.681	269	-132	28.121	3.682	0.008
29.169	3.552	82	$-2\ 1\ 1$	29.168	3.552	0.001
29.649	3.496	197	141	29.651	3.496	-0.002
30.149	3.439	94	200	30.175	3.436	-0.026
30.569	3.393	217	210	30.594	3.390	-0.025
31.129	3.334	170	-142	31.122	3.334	0.007
31.829	3.262	282	220	31.820	3.263	0.009
32.369	3.209	832	112	32.365	3.209	0.004
32.449	3.201	509	-231	32.477	3.199	-0.028
33.299	3.122	63	151	33.303	3.121	-0.004
33.559	3.098	263	160	33.605	3.094	-0.046
34.639	3.005	200	-152	34.635	3.005	0.004
35.539	2.931	47	-232	35.521	2.932	0.018
35.719	2.917	58	-123	35.717	2.917	0.002
36.269	2.874	71	211	36.263	2.874	0.006
37.109	2.811	51	071	37.100	2.812	0.009
37.309	2.796	91	221	37.324	2.795	-0.015
37.899	2.754	62	142	37.890	2.755	0.009
38.999	2.680	18	231	39.037	2.677	-0.038
39.489	2.648	72	-213	39.495	2.647	-0.006
39.889	2.622	16	-143	39.865	2.624	0.023
40.269	2.598	23	080	40.255	2,599	0.014
41.299	2.536	33	241	41.335	2.534	-0.036
42.089	2.491	75	0.8.1	42.086	2.491	0.003
42.989	2.441	16	260	42.996	2.441	-0.007
43.889	2.393	18	-311	43.881	2.394	0.008
44.119	2.382	54	2.5.1	44.144	2.380	-0.025
44.669	2.354	42	123	44.633	2.356	0.036
44.799	2.347	47	-321	44,790	2.348	0.009
46.249	2.278	25	181	46 243	2.278	0.006
	2.270	20			2.270	Continued

TABLE I. Continued

$2\theta_{\rm obs}$	$d_{\rm obs}$	I _{obs}	h k l	$2\theta_{\rm cal}$	$d_{\rm cal}$	$\Delta 2\theta$
46.899	2.248	31	270	46.894	2.248	0.005
47.289	2.230	54	-332	47.295	2.230	-0.006
48.149	2.193	60	172	48.148	2.193	0.001
48.859	2.163	41	014	48.866	2.163	-0.007
49.259	2.146	16	-342	49.281	2.145	-0.022
50.149	2.111	20	-281	50.186	2.109	-0.037
50.519	2.096	25	340	50.488	2.097	0.031
50.989	2.078	37	0 10 0	50.942	2.080	0.047
51.749	2.050	23	-352	51.750	2.050	-0.001
52.279	2.030	32	182	52.282	2.030	-0.003
52.919	2.007	14	350	52.917	2.008	0.002
53.559	1.985	22	-273	53.540	1.986	0.019
56.029	1.904	27	1 10 1	56.044	1.904	-0.015
56.809	1.880	19	-292	56.786	1.881	0.023

IV. CONCLUSION

Crystallization of MDMA.HCl from a variety of solvents may result in either a monohydrate salt or a mixture of anhydrous salt and monohydrate. This work has resulted in an ambient temperature experimental powder pattern of pure monohydrate of the drug, which will be of use to forensic diffractionists.

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