# Article



# Nipalarsite, Ni<sub>8</sub>Pd<sub>3</sub>As<sub>4</sub>, a new platinum-group mineral from the Monchetundra Intrusion, Kola Peninsula, Russia

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# Abstract

Nipalarsite, Ni<sub>8</sub>Pd<sub>3</sub>As<sub>4</sub>, is a new platinum-group mineral discovered in the sulfide-bearing orthopyroxenite of the Monchetundra layered intrusion, Kola Peninsula, Russia (67°52′22″N, 32°47′60″E). Nipalarsite forms anhedral grains (5–80 µm in size) in intergrowths with sperrylite, kotulskite, hollingworthite, isomertieite, menshikovite, palarstanide, nielsenite and monchetundtraite enclosed in pentlandite, anthophyllite, actinolite and chlorite. Nipalarsite is brittle, has a metallic lustre and a grey streak. In plane-polarised light, nipalarsite is light grey with a blue tinge. Reflectance values in air (in %) are: 46.06 at 470 nm, 48.74 at 546 nm, 50.64 at 589 nm and 54.12 at 650 nm. Values of VHN<sub>20</sub> fall between 400.5 and 449.2 kg.mm<sup>-2</sup>, with a mean value of 429.9 kg.mm<sup>-2</sup>, corresponding to a Mohs hardness of ~4. The average result of 27 electron microprobe wavelength dispersive spectroscopy analyses of nipalarsite is (wt.%): Ni 44.011, Pd 28.74, Fe0.32, Cu 0.85, Pt 0.01, Au 0.05, As 25.42, Sb 0.05, Te 0.39, total 99.85. The empirical formula (normalised to 15 atoms per formula unit) is: (Ni<sub>8.10</sub>Fe<sub>0.06</sub>)<sub>2.8.16</sub>(Pd<sub>2.94</sub>Cu<sub>0.18</sub>)<sub>2.3.12</sub>(As<sub>3.68</sub>Te<sub>0.03</sub>)<sub>2.3.71</sub> or, ideally, Ni<sub>8</sub>Pd<sub>3</sub>As<sub>4</sub>. Nipalarsite is cubic, space group  $Fm\overline{3}m$ , with a = 11.4428(9) Å, V = 1498.3(4) Å<sup>3</sup> and Z = 8. The strongest lines in the powder X-ray diffraction pattern of synthetic Ni<sub>8</sub>Pd<sub>3</sub>As<sub>4</sub> [*d*, Å (*I*) (*hkl*)] are: 2.859(10)(004), 2.623(6)(313), 2.557(6)(024), 2.334(11)(224), 2.201(35)(115,333), 2.021(100)(044), 1.906(8)(006,244) and 1.429(7)(008). The crystal structure was solved and refined from the single-crystal X-ray diffraction study of natural nipalarsite. The density calculated on the basis of the empirical formula of nipalarsite is 9.60 g.cm<sup>-3</sup>. The mineral name corresponds to the three main elements: Ni, Pd and As.

**Keywords:** nipalarsite, platinum-group mineral, Ni<sub>8</sub>Pd<sub>3</sub>As<sub>4</sub> phase, electron-microprobe data, reflectance data, X-ray diffraction data, crystal structure, Monchetundra layered intrusion, Russia

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# Introduction

Nipalarsite, Ni<sub>8</sub>Pd<sub>3</sub>As<sub>4</sub>, is a new platinum-group mineral (PGM) species discovered in specimen from the Monchetundra layered Intrusion, Kola Peninsula, Russia. The mineral was found in a drill core taken from the borehole 1819 ( $67^{\circ}52'22''$ N,  $32^{\circ}47'60''$ E) at the depth of 101.3–101.4 m. The mineral was discovered initially as anhedral grains (5–80 µm in size) in polished sections from the bulk sample 1819074 (Grokhovskaya *et al.*, 2009, fig. 3a, c–e). It was described preliminarily as an unnamed phase, Ni<sub>6</sub>Pd<sub>2</sub>As<sub>3</sub>, intergrown with other PGM (UM2003-40-As: NiPd, UM2009-02-As:NiPd, Grokhovskaya *et al.*, 2003, 2009).

A mineral with corresponding chemical composition [(wt.%: Pd 26, Ni 44.6, As 28, total 98.6, with an empirical formula (based on 15 atoms) of Ni<sub>8.27</sub>Pd<sub>2.66</sub>As<sub>4.07</sub>)] and XRD data [d, Å, I (%) (2.28 (3); 2.20 (4); 2.02 (10); 1.931 (3)] was also listed as an unnamed phase Pd<sub>2</sub>Ni<sub>6</sub>As<sub>3</sub> (UM1986-As:NiPd) from massive cubanite–talnakhite ores of the Talnakh and Norilsk deposits, Russia (Genkin *et al.*, 1981; Genkin and Evstigneeva, 1986). Previously, Gervilla *et al.* (1994) reported, that it could represent a Ni-rich extension of the (Pd,Ni)<sub>8</sub>As<sub>3</sub> solid-solution series, endmember of which is a mineral known as stillwaterite Pd<sub>8</sub>As<sub>3</sub>.

The name of the mineral corresponds to its chemical composition: Ni, Pd and As. Both the mineral and mineral name were approved by the Commission on New Minerals and Mineral Names, IMA (IMA2018-075, Grokhovskaya *et al.*, 2018). The type material is deposited at the Fersman Mineralogical Museum, Russian Academy of Sciences, Leninsky Prospekt 18–2, Moscow 119071, Russia, registration number 5236/1.

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Fig. 1. Simplified geological map of the Monchegorsk Igneous Complex showing the sample location (modified from Grokhovskaya et al., 2003).

The article describes the properties of the mineral and its geological occurrence, along with comments on its potential conditions of formation.

# Geological setting, occurrence and associated minerals

The Early Palaeoproterozoic  $(2504.4 \pm 1.5 \text{ to } 2453 \pm 4 \text{ Ma})$  layered intrusions of the Monchegorsk Igneous Complex are located within

the southern inner corner of the Pasvic–Pechenga–Imandra– Varzuga rift-transform system (Amelin *et al.*, 1995; Grokhovskaya *et al.*, 2003; Bayanova *et al.*, 2010). The Monchegorsk Igneous Complex intrudes into gneisses of the Archaean Kola Group and is overlain by volcanics and clastics of Palaeoproterozoic Pechenga–Varzuga Series. The Monchegorsk Igneous Complex consists of the 'Main Gabbro Ridge' massif (including the Monche-, Chuna-, Volche- and Losevy-Tundra intrusions) striking

Table 1 Electron-microprobe	analyses of PGM	associated with nipalarsite	. Monchetundra lavered intrusion
			, <b>,</b>

Mineral	Mer	Kot	Mon	Mtt	Tel	Lvt	lsm	Mnv	Pls	Pls	Cu₃Pt	Nlt	Hlt	Irt	Spt
wt%															
Pd	28.70	45.85	n.d	39.89	42.97	50.99	70.67	48.72	65.55	67.04	4.04	32.35	n.d	n.d	n.d
Pt	1.36	0.81	43.85	n.d	n.d	n.d	0.01	n.d	n.d	n.d	40.92	n.d	22.82	12.48	52.4
Rh	n.d	n.d	n.d	-	-	-	n.d	-	-	-	n.d	-	20.75	5.22	2.01
lr	-	-	-	-	-	-	-	-	-	-	-	-	n.d	30.55	n.d
Ru	-	-	-	-	-	-	-	-	-	-	-	-	9.94	4.84	n.d
Os	-	-	-	-	-	-	-	-	-	-	-	-	n.d	3.88	n.d
Ag	-	-	-	-	26.62	7.19	n.d	-	-	-	-	-	-	n.d	n.d
Ni	0.02	-	0.50	11.06	n.d	n.d	0.27	17.3	0.53		3.25	0.12	-	0.40	0.10
Cu	n.d	-	n.d	0.12	-	0.69	4.05	n.d	0.11	0.59	47.64	64.5	-	n.d	n.d
Au	-	-	-	n.d	0.32	n.d	n.d	-	1.21	5.12	n.d	n.d	-	-	-
Fe	-	-	-	0.23	n.d	-	-	-	0.63	0.81	1.79	1.32		0.54	0.12
Sn	-	-	-	n.d	_	-	0.58	-	21.29	11.7	n.d	n.d	-	n.d	n.d
Sb	-	-	-	-	-	-	15.61	-	0.56	n.d	-	-	-	-	-
Те	57.80	53.26	54.84	48.4	24.73	42.06	n.d	-	n.d	-	-	-	-	-	-
Bi	12.12	n.d	n.d	n.d	5.22	n.d	-	-	-	-	-	-	-	-	-
As	n.d	-	-	-	0.06	-	9.52	33.64	7.86	13.87	1.54	n.d	31.95	24.84	43.7
S	-	-	-	-	n.d	-	n.d	n.d	n.d	n.d	-	-	14.69	11.14	0.54
Total	100.00	99.92	99.19	99.70	99.92	100.93	100.71	99.66	97.77	99.13	99.18	98.29	100.15	93.89	98.88
apfu	3	2	3	5	4	25	15	8	7	7	4	4	3	3	3
Pd	1.03	1.01	-	1.98	1.85	13.52	10.04	3.05	4.62	4.58	0.14	0.9	-	-	-
Pt	0.03	0.01	1.02	-	-	-	-	-	-	-	0.76	-	0.27	0.19	0.9
Rh	-	-	-	-	-	-	-	-	-	-	-	-	0.47	0.15	0.07
lr	-	-	-	-	-	-	-	-	-	-	-	-	-	0.46	-
Ru	-	-	-	-	-	-	-	-	-	-	-	-	0.23	0.14	-
Os	-	-	-	-	-	-	-	-	-	-	-	-	-	0.06	-
Ag	-	-	-	-	1.13	1.88	-	-	-	-	-	-	-	-	-
Ni	-	-	0.04	0.99	-	-	0.07	1.96	0.07	-	0.2	0.01	-	0.02	0.01
Cu	-	-	-	0.01	-	0.31	0.96	-	0.01	0.07	2.71	3.02	-	-	-
Au	-	-	-	-	0.01	-	-	-	0.05	0.19	-	-	-	-	-
Fe	-	-	-	0.02	-	-	-	-	0.09	0.11	0.12	0.07	-	0.03	0.01
Sn	-	-	-	-	-	-	0.07	-	1.35	0.72	-	-	-	-	-
Sb	-	-	-	-	-	-	1.94	-	0.03	-	-	-	-	-	-
Те	1.72	0.98	1.94	2	0.89	9.3	-	-	-	-	-	-	-	-	-
Bi	0.22	-	-	-	0.11	-	-	-	-	-	-	-	-	-	-
As	-	-	-	-	-	-	1.92	2.99	0.79	1.35	0.07	-	0.98	0.96	1.96
S	-	-	-	-	-	-	-	-	-	-	-	-	1.06	1.01	0.06

Abbreviations: Mer – merenskyite, Kot – kotulskite, Mon – moncheite, Mtt – monchetundraite, Tel – telargpalite, Lvt – lukkulaisvaaraite, Ism – isomertieite, Mnv – menshikovite, Pls - palarstanide, Nlt – nielsenite, Hlt – hollingworthite, Irt – irarsite, Spt – sperrylite; n.d – not detected; '-' not analysed; apfu – atoms per formula unit.

NW to SE, and the Monchegorsk Pluton (Fig. 1*a*,*b*). Ultramafic cumulates prevail in the Monchegorsk Pluton whereas intrusions of Main Gabbro Ridge are composed mainly of mafic, and to a lesser extent of ultramafic cumulates.

The Monchetundra layered intrusion consists of a lower ultrabasic zone of dunite, peridotite and pyroxenite (100–300 m thick); a middle zone of rhythmically layered pyroxenite, olivine pyroxenite, norite and gabbronorite (300–400 m thick), and an upper zone of gabbronorite and gabbro–anorthosite (800–1000 m thick). The reef-type platinum-group element (PGE) mineralisation (up to 3–5 ppm Pt + Pd) was discovered within a rhythmically layered zone (Grokhovskaya *et al.*, 2003; 2009).

The Monchetundra intrusion and Monchegorsk Pluton join along the Monchetundra Regional Fault expressed by an extremely permeable shear-zone striking SE–NW. The Monchetundra Regional Fault consists of a lot of steeply dipping faulted slices represented by more- or less-altered rocks of the Monchetundra intrusion, ranging from dunite and peridotite, to pyroxenite, norite and gabbronorite. Low-sulfide PGE mineralisation is located in tectonised chromite-bearing peridotite, pyroxenite and norite, which features highly variable thicknesses (from 1 to 100 m) and PGE contents (from 1–3 to 20 ppm Pt + Pd). More than fifty PGM species have been found in this shear zone including nipalarsite (Grokhovskaya *et al.*, 2009).

The host rocks of the samples investigated are represented by orthopyroxene cumulates and consist of orthopyroxene, olivine, augite and plagioclase, sometimes totally replaced by anthophyllite, actinolite, Mg-hornblende, biotite-phlogopite-group and chlorite-group minerals with minor chromite, ilmenite and magnetite. Disseminated sulfides include pentlandite, violarite, heazlewoodite, chalcopyrite, chalcocite and pyrrhotite with some accessory minerals. The most common PGM are sperrylite, kotulskite, moncheite, merenskyite, isomertieite, telluropalladinite, keithconnite, hollingworthite, irarsite, stibiopalladinite, lukkulaisvaaraite, nipalarsite, menshikovite, telargpalite, monchetundraite, palarstanide, nielsenite and Cu<sub>3</sub>Pt, listed in approximate order of abundance. The electron-microprobe data associated with nipalarsite are given in Table 1.

Nipalarsite typically occurs as a part of complex polymineralic PGM intergrowths attached to pentlandite and enclosed in amphiboles and chlorite (Fig. 2a-f). Nipalarsite was found in polished sections and heavy-mineral concentrates ranging in size from 3.0 to 80 µm. It forms anhedral and subhedral grains intergrown with sperrylite, kotulskite, hollingworthite, isomertieite, palarstanide, monchetundraite and pentlandite (Fig. 2a,b).



**Fig. 2.** Reflected light (*a*) and back-scattered electron (b-f) images of nipalarsite (Npt) and associated minerals from the Monchetundra intrusion. (*a*, *b*) Association of nipalarsite, isomertieite (Ism), sperrylite (Spt), palarstanide (Pls) and kotulskite (Kot), replaced by monchetundraite (Mtt) and pentlandite (Pn); (*c*) two small grains of nipalarsite (used for EBSD analysis) intergrown with palarstanide embedded in anthophyllite (Anth); (*d*) nipalarsite (used for EBSD analysis), isomertieite and pentlandite replaced by chalcocite (Cc); (*e*) polyphase PGMs represented by nipalarsite, isomertieite, sperrylite and hollingworthite, attached to pentlandite enclosed in anthophyllite and chlorite (Chl); (*f*) association of nipalarsite with isomertieite, menshikovite (Mnv) and pentlandite (Pn); inset in the right corner – general view of polyphase intergrowth of nipalarsite with other PGMs.

The genetic link between nipalarsite and other PGM is not apparent: in some cases nipalarsite forms its own grains attached to pentlandite (Fig. 2a,e), after pentlandite (Fig. 2d), or in one

case possibly menshikovite and pentlandite formed symplectites then replaced by nipalarsite, or menshikovite occurred after nipalarsite in a pentlandite matrix (Fig. 2f). An advanced deposition

Table 2. Reflectance data (%) for nipalarsite in air.

λ (nm)	R (%)	λ (nm)	R (%)	λ (nm)	R (%)
400	42.72	520	47.81	620	52.33
420	43.82	540	48.52	640	53.52
440	44.81	546	48.74	650	54.12
460	45.66	560	49.3	660	54.7
470	46.06	580	50.2	680	55.74
480	46.43	589	50.64	700	56.46
500	47.12	600	51.21		

The values required by the Commission on Ore Mineralogy are given in bold.



**Fig. 3.** Reflectance data for nipalarsite in air. The reflectance values (R, %) are plotted versus wavelength ( $\lambda$ , nm).

of pentlandite in relation to the PGM is indicated in Fig. 2*e*, as the pentlandite is fractured and its cracks do not propagate into the adjacent nipalarsite.

#### Physical and optical properties

Nipalarsite is opaque, with metallic lustre, and brittle. Its powder has a grey streak. In plane-polarised reflected light the mineral is light grey with a blue tinge; in intergrowths associated with sperrylite it has a lilac tint. Reflectance values of nipalarsite were measured in air using an MSFU-312 microspectrophotometer (LOMO, Russia) with a WTiC standard (R589 in air = 48.7%). Reflectance data for natural nipalarsite are given in Table 2 and illustrated in Fig. 3. Values of VHN<sub>20</sub> fall between 400.5 and 449.2 kg.mm<sup>-2</sup>, with a mean value of 429.9 kg.mm<sup>-2</sup>, corresponding to a Mohs hardness of ~4. The density calculated on the basis of the empirical formula is 9.60 g.cm<sup>-3</sup>.

# **Chemical composition**

Chemical analyses were performed with a Jeol 8200 electron probe microanalyser in wavelength dispersive spectroscopy (WDS) mode using an electron beam diameter focused to 1–2  $\mu$ m, with an accelerating voltage of 20 kV and a beam current of 20 nA on the Faraday cup. Pure Ni, Pd, Pt, Au metals, and FeS<sub>2</sub>, CuFeS<sub>2</sub>, PbS, GaAs, Sb<sub>2</sub>S<sub>3</sub> and AuTe<sub>2</sub> were used as standards. Concentrations were quantified on the K $\alpha$  lines for Cu, Fe, Ni;  $L\alpha$  lines for Au, Pd, Pt, As, Te, Sb; and M $\alpha$  line for Pb.

Nipalarsite has a fairly constant composition, WDS analyses of representative grains are given in Table 3. The average result of 27 electron microprobe WDS analyses of nipalarsite is (wt.%): Ni 44.01, Pd 28.74, Fe0.32, Cu 0.85, Pt 0.01, Au 0.05, As 25.42, Sb 0.05, Te 0.39, total 99.85. The empirical formula (normalised to 15 atoms per formula unit) for nipalarsite is:  $(Ni_{8.10}Fe_{0.06})_{\Sigma 8.16}(Pd_{2.94}Cu_{0.18})_{\Sigma 3.12}(As_{3.68}Te_{0.03})_{\Sigma 3.71}$ , ideally, Ni<sub>8</sub>Pd<sub>3</sub>As<sub>4</sub>, and for the synthetic analogue (average of nine analyses) is Ni<sub>8.20</sub>Pd<sub>2.80</sub>As<sub>4.00</sub>.

Table 3. Electron-microprobe analyses of nipalarsite and its synthetic analogue\*.

Grain No.	gr1-1	gr1-4	gr1-6	gr1-8	gr1-10	gr1-12	gr2-2	gr2-5	gr2-8	gr3-4	gr3-5	gr5-18	Synth. <i>n</i> = 9
wt.%													
Ni	45.22	43.26	44.70	45.59	43.45	43.28	44.32	43.15	43.70	42.67	43.47	44.58	44.78
Pd	27.62	29.35	28.33	27.52	29.42	29.23	28.20	30.26	30.08	29.58	28.92	26.98	27.73
Pt	0.00	0.05	0.01	n.d.	n.d.	0.08	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	
Au	0.07	n.d.	0.15	0.19	n.d.	0.09	0.03	0.21	n.d.	n.d.	n.d.	n.d.	
Fe	0.26	0.04	0.19	0.11	0.10	0.24	0.07	0.07	0.13	0.87	0.81	0.54	
Cu	0.34	1.81	0.39	0.29	1.93	1.10	0.47	0.53	0.21	1.05	0.50	1.68	
Pb	0.02	0.05	n.d.	0.03	n.d.	n.d.	n.d.	0.03	n.d.	n.d.	0.03	0.07	
As	25.94	25.06	25.86	25.76	24.74	25.13	25.16	25.50	25.43	25.19	25.10	25.05	27.90
Sb	0.04	0.04	0.08	0.06	0.07	0.07	0.04	0.04	0.09	n.d.	n.d.	n.d.	
Te	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	1.08	1.44	0.85	0.05	0.88	0.41	
Total	99.51	99.66	99.70	99.56	99.69	99.23	99.37	101.23	100.48	99.4	99.72	99.30	100.41
Atoms per f	ormula uni	t (based on	15 atoms)										
Ni	8.33	8.03	8.26	8.40	8.05	8.07	8.25	7.98	8.10	7.94	8.07	8.21	8.20
Pd	2.81	3.00	2.89	2.80	3.01	3.01	2.89	3.09	3.07	3.04	2.96	2.74	2.80
Pt	0.00	0.00	0.00			0.00							
Au	0.00		0.01	0.01		0.00	0.00	0.01					
Fe	0.05	0.01	0.04	0.02	0.02	0.05	0.01	0.01	0.03	0.17	0.16	0.11	
Cu	0.06	0.31	0.07	0.05	0.33	0.19	0.08	0.09	0.04	0.18	0.09	0.29	
Pb	0.00	0.00		0.00				0.00			0.00	0.00	
As	3.74	3.64	3.74	3.72	3.59	3.67	3.67	3.69	3.69	3.67	3.65	3.62	4.00
Sb	0.00	0.00	0.01	0.01	0.01	0.01	0.00	0.00	0.01				
Те							0.09	0.12	0.07	0.00	0.08	0.03	

\*Notes: Prefix 'gr.1' denotes grain 1, the largest grain of nipalarsite intergrown with isomertieite, sperrylite, hollingworthite and pentlandite (see Fig. 2e); 'gr.2' – grain 2 used for XRD analysis Fig 2a,b; 'gr.3' – grain 3, nipalarsite studied with EBSD (Fig. 2c); 'gr.5' – grain 5, nipalarsite (studied with EBSD) intergrown with penlandite (Fig 2d); 'Synth.' – synthetic analogue Ni<sub>8</sub>Pd<sub>3</sub>As<sub>4</sub>, average of nine analyses on seven different grains. n.d. – not detected.

Table 4. Crystal	data and	structure	refinement	for the	synthetic	analogue	of
nipalarsite.							

Crystal data	
Structural formula	Ni <sub>8</sub> Pd <sub>3</sub> As <sub>4</sub>
Crystal system, space group	cubic, Fm3m
a (Å)	11.4428(9)
V (Å <sup>3</sup> )	1498.3(4)
Ζ	8
D <sub>calc</sub> g/cm <sup>3</sup>	9.65
μ (mm <sup>-1</sup> )	44.06
Data collection	
Diffractometer	Bruker APEX-II CCD
Radiation, wavelength (Å)	Μο <i>Κ</i> α (λ = 0.71073)
Temperature (K)	293(2)
2θ range for data collection (°)	3.08 to 32.48
R <sub>int</sub> , R <sub>sigma</sub> (%)	3.11, 0.61
Indices range of h, k, l	$-17 \le h \le 17, -17 \le k \le 17,$
	$-17 \leq l \leq 17$
Refinement	
No. of measured, independent and observed [I >2 $\sigma$ (I)] reflections	9957, 179, 175
Data/restraints/parameters	179/0/16
S (Goodness-of-fit on F <sup>2</sup> )	1.439
R [/>=4o (/)]	$R_1 = 0.0144$ , w $R_2 = 0.0357$
R [all data]	$R_1 = 0.0155, wR_2 = 0.0384$
δρ (max)/(min) (e <sup>-</sup> /ų)	1.177/-1.015

 $w = 1/[\sigma^2(F_o^2) + (0.0092P)^2 + 41.4800P]$  where  $P = (F_o^2 + 2F_c^2)/3$ 

#### Synthetic analogue

Several grains of nipalarsite were extracted from the polished sections and examined using single-crystal X-ray diffraction. However, they behaved like aggregates of several grains, showed Debye-Scherrer rings and proved to be unsuitable for structural analysis. Therefore, a fragment of a single crystal of synthetic Ni<sub>8</sub>Pd<sub>3</sub>As<sub>4</sub> was used for the structural study. The synthetic Ni<sub>8</sub>Pd<sub>3</sub>As<sub>4</sub> phase was prepared using the evacuated silica glass tube method. Pure elements: palladium (99.95%), nickel (99.999%) and arsenic (99.999%) were used as starting materials for the synthesis. The evacuated tube with its charge was sealed and then annealed at 900°C for 7 days. After cooling in a coldwater bath, the charge was ground into powder in acetone using an agate mortar, and mixed thoroughly to homogenise. The pulverised charge was sealed in an evacuated silica-glass tube again and reheated at 350°C for 60 days. The experimental product was rapidly quenched in cold water.

#### X-ray crystallography

#### Single-crystal X-ray diffraction

A fragment of the synthetic analogue of  $Ni_8Pd_3As_4$  was examined using a Bruker Apex–II diffractometer equipped with a CCD detector (MoK $\alpha$  radiation). A total of 9957 reflections were collected. The data were integrated and corrected for Lorenz and polarisation effects by means of the Bruker *SAINT* program. A semi-empirical absorption correction was applied on the basis of intensities of equivalent reflections using *SADABS* (Sheldrick, 2008).

The crystal structure was solved *via* direct methods in the space group  $Fm\bar{3}m$  and refined to the final *R* factor of 0.0144 (for 175 unique observed reflections with  $|F| \ge 2\sigma F$ ). All calculations were performed with *SHELX* programs (Sheldrick, 2015*a*,*b*) in the framework of a *WinGX* software package (Farrugia, 2012).

All atomic positions were located by direct methods and subsequently refined anisotropically. Inspection of the difference-Fourier map revealed that maximum positive and negative peaks were 1.177 and  $-1.015 \text{ e}^{-}/\text{Å}^3$ , respectively. The refinement yielded a relatively short Pd-As3 distance of 2.3698 Å within the [As3Pd<sub>6</sub>] octahedra (see below). Refinement allowing the occupancy of As3 atom at the 4*b* position (½, ½, ½) to vary showed no tendency of this atom to depart from the full occupancy. Refinement with the disordered As3 atom at the adjacent 24*e* position (*x*, ½, ½) resulted in unacceptably short Pd-As3 contacts and is considered as less probable than the model with As3 at the 4*b* position. Details of data collection, crystallographic data and refinement are in Table 4. Atomic coordinates and anisotropic displacement parameters are presented in Table 5, respectively and interatomic distances are given in Table 6.

The crystallographic information files have been deposited with the Principal Editor of *Mineralogical Magazine* and are available as Supplementary material (see below).

#### Powder X-ray diffraction

Powder X-ray diffraction data of nipalarsite (Table 7) were obtained from an isolated grain (Fig. 2*b*) using a Rigaku R-AXIS Rapid II diffractometer equipped with a cylindrical image plate detector, in Debye–Scherrer geometry (d = 127.4 mm; CoK $\alpha$  radiation). The data were integrated using the software package *OSC2XRD* (Britvin *et al.*, 2017). As only a few reflections were detected the unit-cell refinement was performed on the synthetic material.

Powder X-ray diffraction data of the synthetic analogue of nipalarsite were collected on a Bruker D8 Advance diffractometer using CuK $\alpha$  radiation and the LynxEye XE detector in Bragg-Brentano geometry. The data were indexed by means of the structure model derived from the single-crystal diffraction data (see above). The unit-cell refinement was performed by the *CelRef* program (Laugier and Bochu, 2003), yielding refined unit-cell parameters of a = 11.4352(6) Å and V = 1495.3 Å<sup>3</sup>.

# Description of the crystal structure

The crystal structure of nipalarsite (Fig. 4) contains two Ni atom positions (both at Wyckoff position 32f), one Pd atom position

Table 5. Atomic coordinates and equivalent isotropic and anisotropic displacement parameters (Å<sup>2</sup>) for the synthetic analogue of nipalarsite.

Atom	Wyck. symbol	X	у	Ζ	Occ.	$U_{\rm eq}$	U <sup>11</sup>	U <sup>22</sup>	U <sup>33</sup>	U <sup>23</sup>	U <sup>13</sup>	$U^{12}$
Pd1	24 <i>e</i>	0.29290(7)	0	0	1	0.0167(2)	0.0198(4)	0.0151(2)	0.0151(2)	0	0	0
As1	4 <i>a</i>	0	0	0	1	0.0059(3)	0.0059(3)	0.0059(3)	0.0059(3)	0	0	0
As2	24 <i>d</i>	0	1/4	1/4	1	0.0079(2)	0.0092(2)	0.0092(2)	0.0054(3)	0	0	0.0001(3)
Ni1	32f	0.38240(5)	0.11760(5)	0.38240(5)	1	0.0094(2)	0.0094(2)	0.0094(2)	0.0094(2)	0.0016(2)	-0.0016(2)	0.0016(2)
Ni2	32f	0.33429(5)	0.33429(5)	0.66571(5)	1	0.0123(2)	0.0123(2)	0.0123(2)	0.0123(2)	-0.0047(2)	-0.0047(2)	0.0047(2)
As3	4 <i>b</i>	1/2	1/2	1/2	1	0.0376(8)	0.0376(8)	0.0376(8)	0.0376(8)	0	0	0

 $\mbox{Table 6.}$  Selected interatomic distances (Å) in the structure of the synthetic analogue of nipalarsite.

Pd1–As3	2.3698(8)	Ni1-As1	2.3309(9)
Pd1-Ni2	2.7231(8) ×4	Ni1–As2	2.5301(4) ×2
Pd1-Ni1	2.7650(6) ×4	Ni1-Ni2	2.5988(8) ×3
Pd1–As2	2.9025(3) ×3	Ni1-Ni1	2.6915(11) ×3
		Ni1-Pd1	2.7650(6) ×2
As1-Ni1	2.3309(9) ×8		
As2–Ni2	2.3358(2) ×4	Ni2–As2	2.3358(2) ×3
As2-Ni1	2.5301(4) ×4	Ni2-Ni1	2.5988(8) ×3
As2-Pd1	2.9025(3) ×4	Ni2-Pd1	2.7231(8) ×3
		Ni2-Ni2	2.7280(18) ×3
As3-Pd1	2.3698(8) ×6		

**Table 7.** Powder X-ray diffraction data (d in Å) obtained from nipalarsite and its synthetic analogue.

Nipal	arsite –	natural	material <sup>1</sup>	Synthetic analogue <sup>2</sup>					
I <sub>obs</sub>	$d_{\rm obs}$	$d_{\rm calc}$	h k l	I <sub>obs</sub>	$d_{\rm obs}$	$d_{\rm calc}$	hkl		
				2	5.7175	5.7175	002		
				10	2.8590	2.8588	004		
				6	2.6234	2.6234	313		
				6	2.5572	2.557	024		
				11	2.3343	2.3342	224		
48	2.21	2.2	115,333	35	2.2007	2.2007	115,333		
100	2.03	2.02	044	100	2.0215	2.0215	044		
				6	1.9327	1.9329	135		
6	1.91	1.91	006,244	8	1.9058	1.9058	006,244		
				3	1.808	1.808	206		
				3	1.7438	1.7438	335		
				6	1.724	1.7239	226		
				1	1.6012	1.6012	117		
				6	1.4887	1.4887	137		
				7	1.4293	1.4294	008		
				4	1.3477	1.3476	606		
				1	1.3204	1.3204	517		

<sup>1</sup>Obtained by Rigaku R-AXIS Rapid II diffractometer equipped with a cylindrical image plate detector, Debye-Scherrer geometry (d = 127.4 mm; CoK $\alpha$  radiation).

<sup>2</sup>Powder diffractometer in Bragg-Brentano geometry, CuKα radiation.

(Wyckoff 24*e*), and three As atom positions (Wyckoff 4*a*, 24*d* and 4*b*). This yields 120 atoms within a unit-cell and the formula  $Ni_8Pd_3As_4$  for Z = 8.

As is typical for intermetallic phases, the nipalarsite crystal structure is characterised by atoms with complex coordination environments and high coordination numbers. Ni1 is coordinated by 13 atoms (4 As, 7 Ni and 3 Pd), while the Ni2 atom has a coordination number of 12 (3 As, 6 Ni and 3 Pd). Ni1–As distances are in the range 2.3309–2.5301 Å. Ni2–As distances are equal to 2.3358 Å. Ni–Ni distances vary between 2.5988 Å and 2.7280 Å. Pd atoms show 4+4 Ni contacts at distances of 2.7231 Å and 2.7650 Å, respectively. The Pd coordination is completed by three As2 atoms at 2.9025 Å and a short distance to the As3 site (2.3698 Å).

The crystal structure of nipalarsite can be considered as a stuffed derivative of the Mg<sub>6</sub>Cu<sub>16</sub>Si<sub>7</sub> structure type (Bergmann and Waugh, 1956), which is also referred to as the Th<sub>6</sub>Mn<sub>23</sub> structure type in the Inorganic Crystal Structure Database (Fachinformationszentrum Karlsruhe, 2018). Considering this relation, the nipalarsite chemical composition can be expressed as Pd<sub>6</sub>Ni<sub>16</sub>As<sub>7</sub>As<sub>1</sub> (for Z = 4), which is directly comparable to that of Mg<sub>6</sub>Cu<sub>16</sub>Si<sub>7</sub>( $\Box$ ), where  $\Box =$  vacancy. In nipalarsite, As3 atoms occupy the 4*b* position of the space group *Fm*3*m*. This position is empty in the Mg<sub>6</sub>Cu<sub>16</sub>Si<sub>7</sub> structure type.



Fig. 4. Crystal structure of nipalarsite showing the edge-sharing framework of  $[\rm Ni_8As_{10}]$  supertetrahedra. The  $[\rm As3Pd_6]$  octahedra are emphasised.



**Fig. 5.** Detailed view of (*a*) the Ni<sub>8</sub> cluster and (*b*) the [Ni<sub>8</sub>As<sub>10</sub>] supertetrahedra in the nipalarsite crystal structure.

Nipalarsite is an intermetallic phase and its structural arrangement is not easy to visualise. As was mentioned by Holman et al. (2008) for synthetic phases with the Mg<sub>6</sub>Ni<sub>16</sub>Si<sub>7</sub> structure type, such structures can be described conveniently in terms of large intermetallic polyhedra. The nipalarsite structure contains a cluster of eight Ni atoms formed from two interpenetrating Ni4 tetrahedra (Fig. 5a). As is indicated in Fig. 5b, this cluster is subsequently surrounded by 10 As atoms forming the [Ni<sub>8</sub>As<sub>10</sub>] supertetrahedra, which can be selected as a basic structural motive used for a structural description. The nipalarsite structure can be then viewed as an edge-sharing framework of the [Ni<sub>8</sub>As<sub>10</sub>] supertetrahedra (Fig. 4). An important feature of this structural arrangement is the presence of large structural cavities. These cavities contain the Pd<sub>6</sub> octahedra with As3 atoms at their centres. The Pd–As3 distance of 2.3698 Å in this regular  $[As(3)Pd_6]$  octahedra is slightly shorter than similar Pd-As contacts in Pd-bearing arsenides (e.g. 2.425 Å in PdAs<sub>2</sub>, Furuseth et al., 1967), however it seems to be a feature of the nipalarsite structure. Similar short bonding distances were observed in crystal structures of polar intermetallic phases (e.g. Ca-Ca in CaAu<sub>3</sub>Ga, Lin and Corbett, 2008; Ag-Ag in AgPd<sub>3</sub>Se, Laufek et al., 2011), very likely to be because of the matrix effect of surrounding atoms. The filler atoms do not have enough space to expand the network and hence short bonds occur.

Nipalarsite does not have a structural analogue in the mineralogical system. From a chemical point of view, the most related minerals are majakite, PdNiAs, (Evstigneeva *et al.*, 2000) and menshikovite,  $Pd_3Ni_2As_3$  (Barkov *et al.*, 2002). The majakite structure is based on  $[PdAs_5]$  square pyramids and  $[NiAs_4]$  distorted tetrahedra, the nipalarsite structure is typical for



Fig. 6. EBSD image of natural nipalarsite; in the right pane, the Kikuchi bands are indexed.

intermetallic compounds with high coordination numbers and hence is very different from that of majakite. The crystal structure of menshikovite is not known.

# Relation of nipalarsite to synthetic Ni<sub>8</sub>Pd<sub>3</sub>As<sub>4</sub>

The structural identity of the natural and synthetic phase was confirmed by electron back-scattered diffraction (EBSD) measurements on the natural sample (grains in Fig. 2c,d) and comparison of X-ray diffraction data obtained from natural and synthetic Ni<sub>8</sub>Pd<sub>3</sub>As<sub>4</sub>.

The collected diffraction data from natural nipalarsite fit very well to those of synthetic Ni<sub>8</sub>Pd<sub>3</sub>As<sub>4</sub> (Table 7). Moreover, the EBSD patterns (also known as Kikuchi patterns) derived from the crystal structure of Ni<sub>8</sub>Pd<sub>3</sub>As<sub>4</sub> phase, fit very well to the measured patterns obtained from the natural grains of nipalarsite. The natural sample was prepared for investigation by etching the mechanically polished surface with colloidal silica (OP-U) for 15 min. to reduce the surface damage. The EBSD patterns were collected and processed using the proprietary computer program AZtec HKL (Oxford Instruments). The values of the mean angular deviation (MAD, i.e. goodness of fit of the solution) between the calculated and measured Kikuchi bands range between 0.53° and 0.39°. These values reveal a very good match; as long as values of mean angular deviation are <1, they are considered as indicators of an acceptable fit. A TESCAN Mira 3GMU scanning electron microscope combined with an EBSD system (Nordlys Nano detector, Oxford Instruments) was used for the measurements. The solid angles calculated from the patterns were compared with a structural

model proposed match containing 90 reflectors to index the patterns. The EBSD patterns obtained from the natural material (18 measurements on different spots on natural nipalarsite) were found to match the patterns generated from our structural model (Fig. 6).

#### Genetic implications and conclusions

Nipalarsite was found in association with other PGMs, disseminated sulfides and hydrous silicates in the intercumulus of orthopyroxene and plagioclase. It crystallised in intergrowths with isomertieite, hollingworthite, other PGMs and pentlandite. In some cases, nipalarsite replaces symplectites of menshikovite and pentlandite.

The Pd–Ni–As system was studied experimentally at 450°C and 790°C by Gervilla *et al.* (1994). The existence of many binary and only two ternary compounds was established at 450°C, PdNiAs (majakite) and Pd<sub>3</sub>Ni<sub>2</sub>As<sub>3</sub> (menshikovite). Nipalarsite was not observed at 450°C or at higher temperatures investigated by Gervilla *et al.* (1994). It is likely that the solid solution of still-waterite (Pd<sub>8</sub>As<sub>3</sub>) dissolving Ni (observed by Gervilla *et al.*, 1994) breaks down at lower temperatures and nipalarsite is formed. The synthetic analogue of nipalarsite was synthesised for this study at 350°C. Apparently its temperature of formation is lower than that of majakite and menshikovite, which is confirmed by the mineral association, where menshikovite and pentlandite form symplectites replaced by nipalarsite. Consequently, the proposed new mineral was probably formed in post magmatic, hydrothermal conditions at lower temperatures, equal or below 350°C.

However, its occurrence at higher temperatures under different thermodynamic conditions is not excluded.

**Supplementary material.** To view supplementary material for this article, please visit https://doi.org/10.1180/mgm.2019.70

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