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Characterization of Portuguese gypsums as raw materials for dermocosmetics

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Abstract

Portuguese gypsum deposits utilized by the cement industry were characterized mineralogically, chemically and technologically for possible application in dermocosmetics. The deposits studied (Loulé, Óbidos and Soure) correspond to small outcrops in diapiric anticline areas. In principle, they represent gypsites which are white, and generally of higher quality for traditional applications (*e.g.* white cement), or greyish, and generally not adequate for cements and mortars. The analytical methods used to characterize the materials were wet sieving and X-ray sedimentation, X-ray diffraction, X-ray fluorescence spectrometry and assessment of abrasiveness, plasticity, texturometrics (adhesivity and firmness), oil absorption and cooling rate. The Óbidos gypsum displayed greater mineralogical and chemical quality (almost pure calcium sulfate) and had a finer grain size (<63 μ m), whereas Loulé and Soure gypsums contain mineralogical impurities (mainly quartz). The Óbidos gypsum shows good characteristics in general for application in dermocosmetics because of its absorption, plasticity, adhesivity, firmness and low abrasiveness.

Keywords: dermocosmetics, gypsum, plaster, Portugal

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The term 'gypsum' derives from the Latin terms 'gypsum' and Greek 'gypsos', meaning 'plaster'. It is a material of great economic potential that has been utilized since prehistoric times. Being abundant and widespread in the Earth's crust, gypsum has been used continuously since the times of Ancient Egypt, and its field of application has increased progressively (Olson, 2002; Scott, 2011). Today, plaster is used in medicine, pharmaceuticals, paper, civil construction and agriculture, among other areas (Velho & Campos, 2006; Schaefer, 2013; Moura *et al.*, 2015).

Gypsum is used in large volumes in Europe, the USA and Asia, with total production of 80 million tons per year (Olson, 2002; Scott, 2011). The construction sector consumes \sim 95% of total plaster production (Scott, 2011).

In Portugal, the known gypsum deposits are of evaporitic origin and occur mainly in the Lusitanian and Algarve basins in the marls and grey clays in the forms of hyaline crystals or fibrous white masses that are silky to granular in appearance (Soure and Loulé) or having a saccharoid aspect (Óbidos). CIMPOR is the main Portuguese cement company; cement manufacture consumes most of the national production of gypsum (~550,000 tons per year) distributed in four centres: Soure, Óbidos, Souto de Carpalhosa and Loulé (Velho & Campos, 2006).

The present work has as its main objective the preliminary analysis of the potential of Portuguese gypsum as a raw material for the future preparation of relevant dermocosmetic

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formulations. A comparative study was focused on three of the main Portuguese gypsum deposits in Loulé, Óbidos and Soure, through examination of chemical, mineralogical, physical and technological properties, with particular emphasis on those recommended for assessment of the degree of quality needed for dermocosmetic formulations (i.e. Olejnik, 1999; Kanouni et al., 2005; Aguzzi et al., 2007; López-Galindo et al., 2007; Viseras et al., 2007; El Karakaya et al., 2017). For dermocosmetic applications, gypsum should be alkaline (pH \approx 10), fine grained (>90% fine fraction) and containing a small amount of potentially hazardous elements such as As, Fe, Cl, F, Cr, Cd and Pb. Similarly to other minerals (e.g. clay minerals) used for dermocosmetics, specific surface area, sorption, cooling rate and textural (adhesiveness, firmness) properties should show medium to high values, whereas abrasiveness should be low (Olejnik, 1999; Baltar et al., 2005; Rowe et al., 2009; Thomas & Puleo, 2009).

Materials and methods

Samples from the Loulé (GPS coordinates: 37.190222, -8.067639), Óbidos (GPS coordinates: 39.361293, -9.175657) and Soure (GPS coordinates: 40.059335, -8.627096) deposits were collected. These deposits correspond to outcrops on small areas of diapiric anticlines, associated with larger saline deposits. In general, the gypsum present is white and usually of sufficiently high quality to be adequate for its traditional applications (cements and mortars), whereas lower grades with greyish (sometimes dark) colourations also occur, which are considered to be of insufficient grade for these applications.

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Fig. 1. Flow chart illustrating the analytical procedure used in the present study. BET = Brunauer-Emmett-Teller; XRD = X-ray diffraction; XRF = X-ray fluorescence.

A flow chart detailing the analytical procedure used in this study is illustrated in Fig. 1. The samples were ground down to 110 μ m and wet sieved at 63 μ m. The grain-size distribution of the <63 μ m fractions was assessed using an X-ray Grain Size Analyser (Micromeritics SediGraph).

The mineralogical composition was determined by X-ray diffraction (XRD) with a Panalytical X'Pert-Pro MPD diffractometer using Cu-K α radiation ($\lambda = 1.5405$ Å) on randomly oriented powders. To assess the final presence of phyllosilicates, suspensions of the samples were heated at 70°C and the insoluble minerals were then analysed. The identification of the different mineral phases followed the criteria recommended by Brindley & Brown (1980) and the Joint Committee for Powder Diffraction Standards. The mineralogical semi-quantification of the identified minerals was performed through peak-area determination of specific reflections and was calculated following the 'reflective powers method' (Galhano *et al.*, 1999; Oliveira *et al.*, 2002).

The chemical composition of the samples was assessed using X-ray fluorescence (XRF) spectrometry with a Panalytical AXIOS PW4400/40 XRF spectrometer. Loss on ignition (LOI) was determined by heating 1 g of the sample at 1000°C for 1 h in a MF20 Cassel furnace (series 83215).

The technological properties, namely abrasiveness, plasticity, texturometrics (adhesivity and firmness), Brunauer–Emmett–Teller (BET) specific surface area, oil absorption and cooling rate, were determined in accordance with the protocols and norms followed in the Departments of Geosciences and of Material Engineering of the University of Aveiro and in the Faculty of Pharmacy of the University of Porto (Quintela *et al.*, 2010, 2014, 2015; Rebelo *et al.*, 2011a, 2011b; Pena-Ferreira *et al.*, 2011).

Results and discussion

The samples studied originally had white (Óbidos) to greyish (Soure and Loulé) colours, which turned to white after grinding. The Óbidos sample is the finest, with 68% of the material at <63 μ m, followed by Loulé (54%) and Soure (48%). Grain-size analysis of the fine fractions yielded a similar size classification, with the Óbidos material showing the lowest value of ϕ 50 (24.8 μ m), followed by Loulé (28.5 μ m) and Soure (30.2 μ m).

Table 1. Mineralogical compositions of the Portuguese gypsum.

	Gypsum	Quartz	Phyllosilicates	Pyrite	Dolomite ^{<i>a</i>}	Silhydrite
Loulé (bulk)	92	2	2	Trace	4	Trace
Loulé (<63 µm)	87	3	3	1	6	2
Óbidos (bulk)	100	-	-	-	-	-
Óbidos (<63 µm)	96	Trace	1	Trace	4	Trace
Soure (bulk)	85	6	3	3	3	Trace
Soure (<63 µm)	77	6	5	4	5	3

^aDolomite and Mg-calcite.

The three samples showed similar pH values, namely 9.8 for Óbidos and 9.7 for Loulé and Soure.

The mineralogical compositions of the bulk samples and the fine fractions are reported in Table 1.

The mineralogical composition is dominated by gypsum, with the Óbidos material being almost monomineralic. Quartz and dolomite are the main accessory minerals; other accessory minerals are phyllosilicates, pyrite and silhydrite. All of the accessory minerals are slightly more abundant in the fine fractions of all samples studied. This quartz increase in fine fractions (<63 µm), indicative of the very fine grain size, may be adverse for common gypsum applications (Velho & Campos, 2006), although for some dermocosmetic applications, the SiO₂, if present in a reactive form (e.g. silhydrite), may have a positive contribution, being a skin regenerator and strengthener and having anti-inflammatory properties (Gomes & Silva, 2007; Schleier et al., 2014). The insoluble residue analysis after gypsum dissolution allowed for more accurate assessment of the phyllosilicates; illite and smectite are present in all of the samples, with illite being more abundant in the Soure material and smectite in the Loulé material.

The chemical compositions (major elements) of the fine fractions and bulk samples are reported in Table 2.

The theoretical composition of gypsum is 32.5% CaO, 46.6% SO_3 and 20.9% H₂O. Thus, Óbidos (GO) and Loulé (GL) gypsums are sufficiently pure, whereas Soure (GS) gypsum contains significant amounts of Al_2O_3 and SiO_2 , related to the presence of quartz and, probably, phyllosilicates. The lack of peaks which are characteristic of phyllosilicates indicate that they may be poorly ordered. The fine-fraction Óbidos gypsum exhibits a greater CaO content and lower LOI than the bulk sample (and

Table 2. Chemical compositions (major elements) of the Portuguese gypsum.

Parameter (analytical		<63 µm		Bulk sample			
error)	GL (%)	GO (%)	GS (%)	GL (%)	GO (%)	GS (%)	
Na ₂ O (0.001)	0.03	0.02	0.06	0.03	0.02	0.04	
MgO (0.005)	2.15	0.56	1.91	1.37	0.33	1.23	
Al ₂ O ₃ (0.021)	1.77	0.28	4.65	1.22	0.12	2.90	
SiO ₂ (0.014)	4.86	0.68	10.03	3.05	0.27	6.32	
P ₂ O ₅ (0.001)	0.02	0.00	0.04	0.01	0.00	0.02	
SO ₃ (0.018)	44.29	46.98	36.87	41.90	44.57	39.55	
K ₂ O (0.003)	0.26	0.06	0.79	0.15	0.02	0.50	
CaO (0.022)	32.17	35.21	25.87	29.64	32.73	27.62	
TiO ₂ (0.003)	0.08	0.00	0.14	0.05	0.00	0.11	
MnO (0.001)	0.01	0.00	0.01	0.00	0.00	0.00	
Fe ₂ O ₃ (0.006)	0.56	0.10	1.21	0.27	0.04	0.82	
Sr (0.001)	0.07	0.20	0.10	0.04	0.09	0.08	
LOI	13.74	15.87	18.31	22.26	21.81	20.79	
Total	100.09	99.96	99.99	99.99	100.00	99.98	

GL = Loulé; GO = Óbidos; GS = Soure.

Table 3. Chemical specifications for health applications (Harben, 2002).

Chemical parameters	Mandatory specifications
CaSO ₄ .2H ₂ O	>96.75%
As	<3 ppm
Se	<30 ppm
F	<30 ppm
Fe	<100 ppm
Pb	<10 ppm
Other heavy metals	<10 ppm

Table 4. Chemical compositions (minor elements) of the samples studied.

		<63 µm		Whole sample				
	$\begin{array}{ccc} {\sf GL} & {\sf GO} & {\sf GS} \\ {({\sf mg}\ {\sf kg}^{-1})} & {({\sf mg}\ {\sf kg}^{-1})} & {({\sf mg}\ {\sf kg}^{-1})} \end{array}$		GL (mg kg ⁻¹)	GO (mg kg ⁻¹)	GS (mg kg ⁻¹)			
As	<4.4	<4.4	11.7	<4.4	<4.4	8.1		
Se	<0.8	<0.8	<0.8	<0.8	<0.8	<0.8		
Pb	6.9	3.4	14.8	4.8	4.5	10.1		
Cr	11.3	4.1	11.5	8.3	3.0	12.0		
Cd	<2.0	<2.0	<2.0	<2.0	<2.0	<2.0		
Ni	3.1	<1.0	4.9	2.2	<1.0	5.0		
Cu	4.7	4.0	5.5	3.1	2.0	5.6		
Zn	3.7	<0.6	9.3	1.4	<0.6	11.3		
Мо	2.8	3.1	1.6	2.1	2.0	1.7		

GL = Loulé; GO = Óbidos; GS = Soure.

the gypsum theoretical composition), in accordance with the mineralogical composition (decrease in gypsum and increase in the Mg-calcite and dolomite contents). The LOI decrease in the fine fractions is due to the decrease in the gypsum contents compared to the bulk sample. In the Soure and Loulé samples, the SiO₂ is present in significant amounts (~5%), which is important for dermocosmetic applications (Gomes & Silva, 2007; Schleier *et al.*, 2014).

Taking into account both the mineralogical and chemical analysis, the samples from Loulé and Soure have lower gypsum contents and greater iron contents than is recommended for health applications (Olejnik, 1999; Baltar *et al.*, 2005; Rowe *et al.*, 2009; Thomas & Puleo, 2009).

Table 5. Abrasiveness and Abrasivity Index values of the samples studied.

Samples		Abrasiveness	Abrasivity Index
Loulé	<63 μm	0.0031	10.1639
	Bulk sample	0.0049	16.0656
Óbidos	<63 μm	0.0022	7.2131
	Bulk sample	0.0043	14.0984
Soure	<63 μm	0.0030	9.8361
	Bulk sample	0.0054	17.7049

Table 6. Atterberg limits and Plasticity Index values of the studied samples.

Sampling site	Liquid limit (%)	Plastic limit (%)	Plasticity Index (%)	Classification
Loulé	26.164	21.00153	5.16247	Weakly plastic
Óbidos	31.628	28.83701	2.79099	Weakly plastic
Soure	28.120	18.90927	9.21073	Medium plastic

Table 7. Technological tests of the studied samples.

BET specific surface area $(m^2 g^{-1})$ 28Cumulative pore volume $(m^2 g^{-1})$ 26Absorption of linseed oil (g) 27Adhesiveness $(N mm^{-1})$ 4 days12 days-0Firmness (N) 4 days12 days012 days012 days0	3 22 3 22 32 32 0.4 -0.3 0.5 -1.4 0.30 0.50 0.60 0.33	31 29 25 -0.3 -0.5 0.30 0.43

Table 3 shows some of the chemical specifications for health applications (according to Harben, 2002).

The concentrations of the minor elements considered important for health applications in the samples studied are listed in Table 4.

The Soure gypsum contains more As, Pb and Cr in both the <63 μ m fractions and the bulk sample than the amount allowed for health applications. The Loulé and Óbidos samples show generally adequate values, with the Loulé sample having slightly greater Pb and Cr contents.

Thus, according to the mineralogical and chemical composition, only Óbidos gypsum is adequate for dermocosmetic applications, but only with a finer granulometry (*i.e.* after more extensive milling; Olejnik, 1999; Baltar *et al.*, 2005; Rowe *et al.*, 2009; Thomas & Puleo, 2009).

All studied samples show low values for abrasiveness and on the Abrasivity Index (AI) (Table 5). Abrasiveness is the net mass loss during testing. The AI is the loss in weight (in g m⁻²) of a standard bronze net when in contact with a suspension of the material after a certain number of rotations (Quintela *et al.*, 2014). The Óbidos gypsum displays lower abrasiveness and AI values in both the bulk samples and the <63 μ m fractions.

The low abrasiveness values allow classification of these samples as 'not abrasive' (Quintela *et al.*, 2014). The AI is considerably lower than the recommended values for aesthetic and pharmaceutical formulations (Baltar *et al.*, 2005; Rebelo *et al.*, 2011a, 2011b; Quintela *et al.*, 2014). The Atterberg limits and the Plasticity Index (PI) of the samples studied are indicative of materials with low plasticity (Table 6).

The results of technological tests to assess the potential use of the studied gypsums for dermocosmetics (BET specific surface

Table 8. Summary of the technological properties of the gypsum samples.

						Properties					
Sample	Mineralogy	Chemistry	SSA	CPV	OA	А	F	PI	AI	рН	СТ
Loulé Óbidos Soure	$\sqrt[n]{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt$	$\sqrt[n]{\sqrt{\sqrt{x^2}}}$	$\sqrt[]{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{$	$\sqrt[]{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{$	$\sqrt[]{\sqrt[]{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{$	$\begin{array}{c} \sqrt{} \\ \sqrt{} \\ \sqrt{} \\ \sqrt{} \end{array}$	$\begin{array}{c} \sqrt{} \\ \sqrt{} \\ \sqrt{} \\ \sqrt{} \end{array}$	$\sqrt[n]{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt$	$\begin{array}{c} \sqrt{} \\ \sqrt{} \sqrt{} \\ \sqrt{} \sqrt{} \end{array}$	$\begin{array}{c} \sqrt[4]{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt{\sqrt$	$\sqrt[]{}$

X = unsatisfactory; $\sqrt{=}$ satisfactory; $\sqrt{\sqrt{=}}$ good; $\sqrt{\sqrt{\sqrt{=}}}$ every good. SSA = specific surface area; CPV = cumulative pore volume; OA = oil absorption; A = adhesiveness; F = firmness; PI = Plasticity Index; AI = Abrasivity Index; CT = cooling time.

area, cumulative pore volume, oil absorption, adhesiveness and firmness) are listed in Table 7.

All of the samples studied show moderate BET specific surface area, cumulative pore volume and oil absorption, which are acceptable for the desired applications (Olejnik, 1999; Baltar et al., 2005; Aguzzi et al., 2007; López-Galindo et al., 2007; Viseras et al., 2007; Rowe et al., 2009; Thomas & Puleo, 2009; Karakaya et al., 2010). In addition, the adhesiveness and firmness values may be considered as adequate for dermocosmetic applications (Pena-Ferreira et al., 2011). All of the samples show an increase with time for both properties which is almost linear, and which is more pronounced in the Óbidos gypsum.

Finally, the cooling time from 55°C to 30°C) varied between 16.5 min (Óbidos) and 21 min (Soure). The application of peloids usually lasts for 30 min for a temperature ranging from 45°C to 40°C. According to Legido et al. (2007), the cooling time of peloids which are suitable for therapy should be between 20 and 25 min. Thus, the Óbidos gypsum would not be suitable for this application.

Table 8 shows a qualitative summary of all of the assessed technological properties for practical applications in geophagy and pelotherapy.

Conclusions

All of the samples studied are relatively pure, with the Óbidos gypsum having the greatest purity from both mineralogical and chemical perspectives, as well as it showing the greatest content of fine-grained particles.

The Óbidos gypsum fulfils the requirements for almost all properties relevant to dermocosmetics, being the only sample that may be considered to be adequate for dermocosmetic applications (Olejnik, 1999; Baltar et al., 2005; Aguzzi et al., 2007; López-Galindo et al., 2007; Viseras et al., 2007; Rowe et al., 2009; Thomas & Puleo, 2009; Karakava et al., 2010), particularly for dental applications (Olejnik, 1999; El Kanouni et al., 2005). In contrast, the Soure and Loulé gypsums should be submitted to finer milling and reassessed.

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References

Aguzzi C., Cerezo P., Viseras C. & Caramella C. (2007) Use of clays as drug delivery systems: Possibilities and limitations. Applied Clay Science, 36, 22-36.

- Baltar C., Bastos F. & Luz A. (2005) Gipsita. Pp. 449-470 in: Rochas e Minerais Industriais (A.B. da Luz & F.F. Lins, editors). Centro de Tecnologia Mineral Ministério da Ciência e Tecnologia, Rio de Janeiro, Brazil.
- Brindley G.W. & Brown G. (1980) Crystal Structures of Clay Minerals and their X-ray Identification. Mineralogical Society, London, UK.
- El Kanouni M.A., Aazzab B., Tricha L., Samdi A., Moussa R., Hamel J. & Gomina M. (2005) Elaboration and characterization of dental plasters obtained from a Moroccan gypsum. Journal of Physics IV, 123, 245-249.
- Galhano C., Rocha F. & Gomes C. (1999) Geostatistical analysis of the influence of textural, mineralogical and geochemical parameters on the geotechnical behavior of the 'Clays Aveiro' formation (Portugal). Clay Minerals, 34, 109-116.
- Gomes C. & Silva J. (2007) Minerals and clay minerals in medical geology. Applied Clay Science, 36, 4-21.
- Harben P.W. (2002) Industrial minerals market profiles. Pp. 154-159 in: The Industrial Minerals Handbook: A Guide to Markets, Specifications and Prices (L. Taylor, editor). Pensord, Blackwood, UK.
- Karakaya M.C., Karakaya N., Sarıoğlan S. & Koral M. (2010) Some properties of thermal muds of some spas in Turkey. Applied Clay Science, 48, 531-537.
- Karakaya M.C., Karakaya N. & Aydin S. (2017) The physical and physicochemical properties of some Turkish thermal muds and pure clay minerals and their uses in therapy. Turkish Journal of Earth Sciences, 26, 395-409.
- Legido J.L., Medina C., Mourelle M.L., Carretero M.I. & Pozo M. (2007) Comparative study of the cooling rates of bentonite, sepiolite and common clays for their use inpelotherapy. Applied Clay Science, 36, 148-160.
- Lopez-Galindo A., Viseras C. & Cerezo P. (2007) Compositional, technical and safety specifications of clays to be used as pharmaceutical and cosmetic products. Applied Clay Science, 36, 51-63.
- Moura A., Velho J.L. & Alves W. (2015) Estudo preliminar do depósito de paligorsquite de Figueiró do Campo (Soure, Portugal). Cadernos Laboratorio Xeolóxico de Laxe, 38, 93-106.
- Olejnik R. (1999) Modern applications of gypsum products. ZKG International, 52, 649-653.
- Oliveira A., Rocha F., Rodrigues A., Jouanneau J., Dias A., Weber O. & Gomes C. (2002) Clay minerals from the sedimentary cover from the Northwest Iberian shelf. Progress in Oceanography, 52, 233-247.
- Olson D.W. (2002) Gypsum. Pp. 1-10 in: US Geological Survey Minerals Yearbook. US Geological Survey, Reston, VA, USA.
- Pena-Ferreira M.R., Santos D., Silva J.B., Amaral M.H., Sousa-Lobo J.M., Gomes J.H. & Gomes C. (2011) Desenvolvimento de formulações contendo argila esmectítica e areia carbonatada biogénica da ilha Porto Santo para aplicação em máscaras de limpeza e branqueadoras. Anales de Hidrología Médica, 4, 57-66.
- Quintela A., Terroso D., Almeida S.F.P., Reis P., Moura A., Correia A., Ferreira S.E., Forjaz V. & Rocha F. (2010) Geochemical and microbiological characterization of some Azorean volcanic muds after maturation. Research Journal of Chemistry and Environment, 14, 66-74.
- Quintela A., Costa C., Terroso D. & Rocha F. (2014) Abrasiveness Index of dispersions of Portuguese clays using the Einlehner method: influence of clay parameters. Clay Minerals, 49, 27-34.
- Quintela A., Costa C., Terroso D., Sá H., Nunes J.C. & Rocha F. (2015) Characterization and evaluation of hydrothermally influenced clayey sediments from Caldeiras da Ribeira Grande fumarolic field (Azores Archipelago, Portugal) used for aesthetic and pelotherapy purposes. Environmental Earth Sciences, 73, 2833-2842.

- Rebelo M., Viseras C., Galindo A.L., Rocha F. & Silva E.F. (2011a) Characterization of Portuguese geological materials to be used in medical hydrology. *Applied Clay Science*, **51**, 258–266.
- Rebelo M., Viseras C., Galindo A.L., Rocha F. & Silva E.F. (2011b) Rheological and thermal characterization of peloids made of selected Portuguese Geological materials. *Applied Clay Science*, 52, 219–227.
- Rowe R., Sheskey P.J. & Quinn M.E. (editors) (2009) Handbook of Pharmaceutical Excipients, 6th edition. Pharmaceutical Press, London, UK.
- Schaefer C.O. (2013) Valorização de fonte alternativa de sulfato de cálcio para a produção de argamassas autonivelantes. Ed. Universidade Federal de Santa Catarina, Brazil. Available at: https://repositorio.ufsc.br/xmlui/ handle/123456789/123140.
- Schleier R., Galitesi C.R. & Ferreira E. (2014) Silício e cálcio uma abordagem antroposófica (Silicon and calcium – an anthroposophic approach). Arte Médica Ampliada, 34, 102–113.

- Scott F. (2011) GR4CM: Gypsum Recycling for Cement Manufacture. Waste Minimisation Fund Feasibility Study. Available at: http://www.tnc.co.nz/ files/gr4cm_ms2_report_031111.pdf.
- Thomas M.V. & Puleo D.A. (2009) Calcium sulfate: properties and clinical applications. Journal of Biomedical Materials Research. Part B, Applied Biomaterials, 88B, 597–610.
- Velho J. & Campos C. (2006) Geologia do depósito de gesso de S. José do Pinheiro (Soure). Pp. 1071–1074 in: Proc. VII Congresso Nacional de Geologia, Évora. Available at: https://www.socgeol.org/documents/geologia-do-deposito-degesso-de-s-jose-do-pinheiro-soure.
- Viseras C., Aguzzi C., Cerezo P. & Lopez-Galindo A. (2007). Uses of clay minerals in semisolid health care and therapeutic products. *Applied Clay Science*, 36, 37–50.