

X-ray powder diffraction data for compound $\text{Er}_3\text{Co}_4\text{Al}_{12}$ Zhenwei Wen, Chen Liu, Lingmin Zeng, and Jialin Yan^{a)}

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(Received 11 January 2013; accepted 20 April 2013)

A new compound $\text{Er}_3\text{Co}_4\text{Al}_{12}$ was prepared by arc melting under argon atmosphere. The powder X-ray diffraction data of $\text{Er}_3\text{Co}_4\text{Al}_{12}$ were successfully indexed, giving a hexagonal structure with $a = 8.6185(2) \text{ \AA}$, $c = 9.2347(3) \text{ \AA}$, and unit-cell volume $V = 594.04 \text{ \AA}^3$. Compound $\text{Er}_3\text{Co}_4\text{Al}_{12}$ has the $\text{Gd}_3\text{Ru}_4\text{Al}_{12}$ type-structure, $Z = 2$ and space group $P6_3/mmc$. © 2013 International Centre for Diffraction Data. [doi:10.1017/S0885715613000407]

Key words: erbium cobalt aluminum, powder X-ray diffraction

I. INTRODUCTION

The RE–Co–Al (RE = rare earth) system is the basic system for the rare earth-based bulk metallic glasses, which have been the subject of recent intensive studies owing to their excellent glass forming ability, good mechanical and magnetic properties (Luo and Wang, 2009). A new compound $\text{RE}_3\text{Co}_4\text{Al}_{12}$ (RE = Dy, Er) was found during our investigation of the phase equilibria in the Er–Co–Al and Dy–Co–Al ternary systems. Analysis of the X-ray diffraction data of the compound showed that $\text{RE}_3\text{Co}_4\text{Al}_{12}$ has the hexagonal $\text{Gd}_3\text{Ru}_4\text{Al}_{12}$ type-structure, $Z = 2$, space group $P6_3/mmc$, isostructural with the compound $\text{U}_3\text{Co}_4\text{Al}_{12}$ (Tougaard *et al.*, 2004). In this paper, the X-ray powder diffraction data up to $110^\circ 2\theta$ of the new compound $\text{Er}_3\text{Co}_4\text{Al}_{12}$ are reported.

II. EXPERIMENTAL

The sample was prepared by arc melting of Er (99.95%), Co (99.9%), and Al (99.99%) on a water-cooled copper hearth with a non-consumable tungsten electrode under argon atmosphere. The sample was turned and remelted three times to ensure sample homogeneity. The ingot obtained was annealed in an evacuated quartz tube at 600°C for 600 h and then cooled down slowly to 500°C , kept for 480 h before quenching in liquid nitrogen. Specimen for XRD measurements was ground in agate mortar and pestle to a particle size of less than $10 \mu\text{m}$, loaded in Al well and gently pressed. Two sets of the powder XRD patterns, with or without Si added as an internal standard, were collected at room temperature on a Rigaku D/Max 2500 V diffractometer with $\text{CuK}\alpha_1$ ($\lambda = 1.54060 \text{ \AA}$) and a graphite monochromator. The scan ranged from 10 to $110^\circ 2\theta$ with a step size of 0.02° and a counting time of 2 s per step. The XRD pattern recorded from the specimen added with high purity Si (99.999%) internal standard was used to obtain the observed peak positions $2\theta_{\text{obs}}$ for indexing, while the XRD pattern without Si internal standard was used for the observed peak height I_{obs} . The Jade software package (Materials Data, Inc., Livermore, California) was used to analyze the data. The $2\theta_{\text{obs}}$ values of the peaks were determined

by the Savitzky–Golay second derivative method after background subtracting, $K\alpha_2$ stripping and Si internal standard calibration using PDF 027-1402 (ICDD, 2011) with $\lambda = 1.540598 \text{ \AA}$. Pattern indexing was performed using the program TREOR (Werner *et al.*, 1985) and the values of unit-cell parameters were obtained.

III. RESULTS

The powder XRD pattern indicated that the trace amount of the impurity phase identified as Er_2O_3 (PDF 43-1007) (ICDD, 2011) was detected as well as $\text{Er}_3\text{Co}_4\text{Al}_{12}$. The strongest reflection of the Er_2O_3 phase did not overlap with those of $\text{Er}_3\text{Co}_4\text{Al}_{12}$ and its peak height is less than 3%. The experimental XRD pattern of compound $\text{Er}_3\text{Co}_4\text{Al}_{12}$ is shown in Figure 1. All lines except for those corresponding to Er_2O_3 were successfully indexed using TREOR and it showed that the compound $\text{Er}_3\text{Co}_4\text{Al}_{12}$ is hexagonal with unit-cell parameters $a = 8.6185(2) \text{ \AA}$, $c = 9.2347(3) \text{ \AA}$, and unit-cell volume $V = 594.04 \text{ \AA}^3$. The figure of merit for indexing F_N (Smith and Snyder, 1979) is $F_{30} = 123.2$ (0.0058, 42). The X-ray powder diffraction data for $\text{Er}_3\text{Co}_4\text{Al}_{12}$ are listed in Table I. The peak height values of the calculated relative

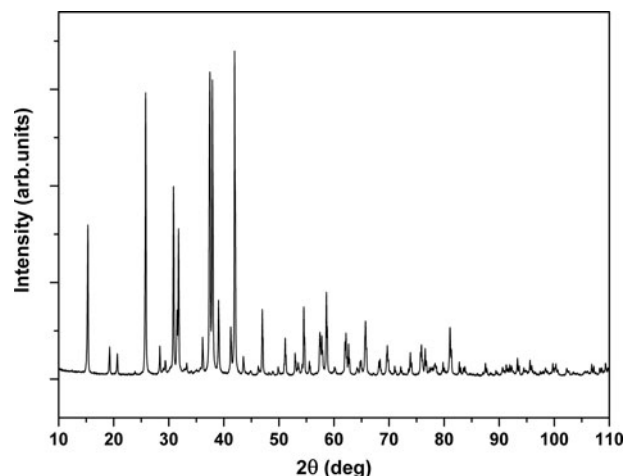


Figure 1. Powder diffraction pattern of $\text{Er}_3\text{Co}_4\text{Al}_{12}$ ($\text{CuK}\alpha_1$ radiation, $\lambda = 1.54060 \text{ \AA}$, 40 kV, 150 mA, step size 0.02° , and counting time 1 s per step).

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TABLE I. X-ray powder diffraction data for $\text{Er}_3\text{Co}_4\text{Al}_{12}$ [$P6_3/mmc$, $a = 8.6185(2)$ Å, $c = 9.2347(3)$ Å, $V = 594.04$ Å³, $Z = 2$, $\text{CuK } \alpha_1$, and $\lambda = 1.54060$ Å]. Only those peaks with I_{obs} of 1 or greater are presented.

No.	$2\theta_{\text{obs}}$ (°)	d_{obs} (Å)	I_{obs}	h	k	l	$2\theta_{\text{cal}}$ (°)	d_{cal} (Å)	I_{cal}	$\Delta 2\theta$ (°)
1	15.236	5.811	36	1	0	1	15.251	5.8049	52	-0.015
2	19.196	4.620	7	0	0	2	19.207	4.6174	13	-0.011
3	20.580	4.312	5	1	1	0	20.594	4.3093	3	-0.014
4	23.825	3.732	1	2	0	0	23.824	3.7319	1	0.001
5	25.718	3.461	76	2	0	1	25.727	3.4601	80	-0.009
6	28.310	3.150	8	1	1	2	28.306	3.1504	7	0.004
7	30.778	2.903	54	2	0	2	30.781	2.9024	59	-0.003
8	31.402	2.846	17	1	0	3	31.410	2.8457	22	-0.008
9	31.696	2.821	41	2	1	0	31.692	2.8211	36	0.004
10	33.181	2.698	2	2	1	1	33.179	2.6980	1	0.002
11	36.062	2.489	10	3	0	0	36.072	2.4879	10	-0.010
12	37.341	2.406	79	2	1	2	37.324	2.4073	60	0.017
13	37.858	2.375	88	2	0	3	37.856	2.3747	100	0.002
14	38.995	2.308	22	0	0	4	38.981	2.3087	39	0.014
15	41.181	2.190	14	3	0	2	41.182	2.1902	18	-0.001
16	41.898	2.154	100	2	2	0	41.895	2.1546	80	0.003
17	43.480	2.0797	5	2	1	3	43.478	2.0798	6	0.002
18	46.202	1.9633	3	2	0	4	46.200	1.9634	3	0.002
19	46.920	1.9349	21	3	0	3	46.918	1.9350	28	0.002
20	49.821	1.8288	3	4	0	1	49.816	1.8290	1	0.005
21	50.899	1.7926	4	1	0	5	50.890	1.7929	2	0.009
22	51.080	1.7867	11	2	1	4	51.080	1.7867	14	0.000
23	52.880	1.7300	7	4	0	2	52.879	1.7300	6	0.001
24	53.285	1.7178	2	3	1	3	53.285	1.7178	1	0.000
25	53.475	1.7121	3	3	2	0	53.469	1.7123	4	0.006
26	54.145	1.6925	3	3	0	4	54.152	1.6923	2	-0.007
27	54.458	1.6835	22	3	2	1	54.455	1.6836	19	0.003
28	55.462	1.6554	4	2	0	5	55.466	1.6553	7	-0.004
29	57.397	1.6041	13	4	1	1	57.402	1.6040	7	-0.005
30	57.737	1.5955	12	4	0	3	57.729	1.5957	9	0.008
31	58.558	1.5751	26	2	2	4	58.552	1.5752	23	0.006
32	60.042	1.5396	2	0	0	6	60.064	1.5391	3	-0.022
33	61.960	1.4965	11	3	2	3	61.965	1.4964	16	-0.005
34	62.120	1.4930	9	5	0	0	62.131	1.4928	8	-0.011
35	62.582	1.4831	10	3	0	5	62.588	1.4830	12	-0.006
36	64.179	1.4500	2	1	1	6	64.206	1.4494	2	-0.027
37	64.700	1.4396	4	4	1	3	64.696	1.4396	3	0.004
38	64.842	1.4368	3	3	3	0	64.859	1.4364	4	-0.017
39	65.562	1.4227	13	2	0	6	65.554	1.4229	10	0.008
40	65.680	1.4204	16	5	0	2	65.683	1.4204	17	-0.003
41	68.135	1.3751	4	3	2	4	68.123	1.3753	7	0.012
42	68.322	1.3718	3	3	3	2	68.335	1.3716	5	-0.013
43	69.538	1.3508	7	2	1	6	69.517	1.3511	9	0.021
44	69.638	1.3491	9	4	2	2	69.643	1.3490	6	-0.005
45	70.981	1.3268	3	5	1	1	70.991	1.3266	4	-0.010
46	72.099	1.3090	3	3	0	6	72.102	1.3089	5	-0.003
47	73.515	1.2872	2	5	1	2	73.503	1.2874	1	0.012
48	73.823	1.2826	7	4	2	3	73.841	1.2823	5	-0.018
49	75.680	1.2557	7	3	2	5	75.678	1.2557	6	0.002
50	75.821	1.2537	9	5	0	4	75.830	1.2536	8	-0.009
51	76.519	1.2440	8	6	0	0	76.519	1.2440	4	0.000
52	77.302	1.2333	2	6	0	1	77.338	1.2328	2	-0.036
53	77.760	1.2272	2	4	3	0	77.771	1.2270	2	-0.011
54	78.199	1.2214	3	4	1	5	78.184	1.2216	2	0.015
55	79.761	1.2014	4	6	0	2	79.778	1.2011	5	-0.017
56	80.922	1.1870	6	4	0	6	80.897	1.1873	3	0.025
57	81.040	1.1856	16	4	3	2	81.016	1.1859	5	0.024
58	82.723	1.1657	4	3	0	7	82.738	1.1655	5	-0.015
59	83.481	1.1570	2	5	2	2	83.480	1.1570	2	0.001
60	83.796	1.1535	2	6	0	3	83.807	1.1534	2	-0.011
61	87.480	1.1141	4	5	2	3	87.480	1.1141	4	0.000
62	89.365	1.0955	2	6	0	4	89.400	1.0951	1	-0.035
63	90.603	1.0837	2	4	3	4	90.621	1.0835	2	-0.018
64	91.282	1.0774	3	4	4	0	91.290	1.0773	2	-0.008
65	91.939	1.0714	3	5	0	6	91.921	1.0716	5	0.018

Continued

TABLE I. Continued

No.	$2\theta_{\text{obs}}$ (°)	d_{obs} (Å)	I_{obs}	h	k	l	$2\theta_{\text{cal}}$ (°)	d_{cal} (Å)	I_{cal}	$\Delta 2\theta$ (°)
66	92.264	1.0685	2	2	1	8	92.276	1.0684	2	-0.012
67	93.302	1.0593	5	7	0	1	93.309	1.0592	2	-0.007
68	94.481	1.0492	2	4	4	2	94.483	1.0491	2	-0.002
69	95.617	1.0397	5	4	2	6	95.591	1.0399	5	0.026
70	96.195	1.0350	1	6	2	0	96.184	1.0350	2	0.011
71	97.405	1.0253	1	4	1	7	97.424	1.0251	1	-0.019
72	97.999	1.0207	1	6	1	4	97.969	1.0209	1	0.030
73	98.440	1.0173	2	2	2	8	98.408	1.0175	3	0.032
74	99.742	1.0075	4	7	0	3	99.732	1.0075	2	0.010
75	100.301	1.0033	4	5	2	5	100.293	1.0034	4	0.008
76	102.243	0.9895	2	2	0	9	102.262	0.9894	4	-0.019
77	103.48	0.9810	1	6	2	3	103.471	0.9811	1	0.009
78	105.663	0.9667	1	7	1	2	105.658	0.9667	1	0.005
79	106.815	0.9594	4	4	3	6	106.807	0.9595	2	0.008
80	107.177	0.9572	1	3	2	8	107.178	0.9572	2	-0.001
81	108.262	0.9506	2	5	4	1	108.261	0.9506	2	0.001
82	108.587	0.9486	2	3	0	9	108.596	0.9486	2	-0.009
83	109.319	0.9443	4	6	2	4	109.291	0.9445	6	0.028

intensities I_{cal} were obtained using the LAZY-PULVERIX program (Yvon K *et al.*, 1977) based on the structure of the compound $\text{U}_3\text{Co}_4\text{Al}_{12}$ (Tougait *et al.*, 2004). All reflections are consistent with the $P6_3/mmc$ space group.

SUPPLEMENTARY MATERIALS AND METHODS

The Supplementary material referred to in this article can be found online at journals.cambridge.org/pdj.

ACKNOWLEDGEMENTS

This work was supported by the Natural Science Foundation of Guangxi (No. 0991053) and SRF for ROCS, SEM.

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