

ASSESSING AVAILABLE SOIL SULPHUR FROM PHOSPHOGYPSUM APPLICATIONS IN A NO-TILL CROPPING SYSTEM

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SUMMARY

The complex dynamics of the sulphur (S) cycle has prompted several questions concerning its bioavailability and evaluation by different extraction methods, and few studies under field conditions have been performed to elucidate available S on Brazilian soils under no-till (NT) cropping systems. A field experiment examined the effects of phosphogypsum (PG) applications on nutrient uptake and availability of S to maize and wheat crops under an NT system aiming to establish critic levels of S on an Oxisol (clay, kaolinitic, Rhodic Hapludox) in Parana state, Brazil. PG at the rates of 0, 3, 6 and 9 t ha⁻¹ was applied on the surface of the plots in 1998 upon conversion from pasture to an NT cropping system, and in 2004 subplots received either 0 or 6 t ha⁻¹ of PG on the surface. Effects of PG applications on nutrient uptake and available soil SO₄-S to the maize and wheat crops were evaluated in 2004–2005. Increasing the surface-applied PG rate in 1998 increased Mg uptake by maize, and N, K, Ca and Mg uptake by wheat plants. Reapplication of PG in 2004 increased the uptake of N and P by maize, and the uptake of N, K, Ca and Mg by wheat plants. The PG rates surface-applied in 1998 have resulted in a long-term residual effect on soil by increasing SO₄-S in surface and subsoil layers 6.5 and 7.5 years after application, affecting aboveground biomass, S uptake and S content in the leaves of the wheat crop. The reapplication of PG in 2004 increased maize S uptake, S content in leaves and grains and S exported by maize harvest, and increased wheat aboveground biomass, S uptake and S content in leaves. These effects were due to increasing available soil SO₄-S from different depths extracted by 0.5 M ammonium acetate (NH₄OAc) along with 0.25 M acetic acid (HOAc) and 0.01 M calcium phosphate (Ca(H₂PO₄)₂). Both extractants were very similar in their ability to assess available soil S in these crops, and the 0–0.20-m layer is shown to be adequate for evaluating the concentration of available SO₄-S in maize and wheat cultivated under NT.

INTRODUCTION

Sulphur (S) is an essential element for living organisms, being a component of many proteins, enzymes and cofactors (Solomon *et al.*, 2011). In the last few years, S deficiency and increased need for S fertilization are being reported in many parts of the world (Kost *et al.*, 2008; Scherer, 2009). Intensive land use, decreases in S emissions by power

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plants (Lehmann *et al.*, 2008), increasing crop productivity, continuous removal of nutrients by harvests and use of fertilizers with concentrated NPK formulations have made some plants strictly dependent on S fertilization (Kertesz and Mirleau, 2004). Although many studies with S were published in the 1970s and 1980s, there are still many questions concerning its complex dynamics.

Organic S represents about 90% of the total S pool in Brazilian soils (Neptune *et al.*, 1975) and although not readily plant-available, it can be an important S source in periods of deficiency (De Bona and Monteiro, 2010; Schmidt *et al.*, 2012). No-till (NT) cropping systems, widely adopted in the Southern region of Brazil (Derpsh and Friedrich, 2009), are known to enhance soil organic C content and to improve chemical, physical and biological attributes in tropical and subtropical soils (Lal, 2006). Thus, these systems may also promote the accumulation and turnover of organic S through increased crop residues and litter decomposition.

Phosphogypsum (PG) is a soil conditioner employed in NT systems with the goal of ameliorating root environment by alleviating aluminum (Al) toxicity and through calcium (Ca) supply (Farina *et al.*, 2000; Sumner, 1986; Toma *et al.*, 1999). It is also an important source of S for crops (Caires *et al.*, 2011a, b), since it provides S in forms readily available for plants, and, when applied in large amounts, it can cause long-term residual effects, mainly in clay soils (Caires *et al.*, 2006; 2011a; Toma *et al.*, 1999), due to retention of S on colloidal surfaces.

Determining availability of S in soil based on extraction methods is very challenging since several forms of S are found in soils (Goh and Pamidi, 2003). The available S fractions include readily soluble inorganic sulphate ($\text{SO}_4\text{-S}$), adsorbed $\text{SO}_4\text{-S}$ and a portion of organic S (C-bonded or ester-S), which can be mineralized during plant growth (Scherer, 2009). The most common methods used to determine available S are based on evaluating soluble + adsorbed S, since the inorganic fraction is the main S pool available in soils with high $\text{SO}_4\text{-S}$ adsorption capacity (Tabatabai, 2005), and the methods that evaluate only soluble S are not adequate for soils with high $\text{SO}_4\text{-S}$ adsorption capacity (Ribeiro *et al.*, 2001).

Numerous solutions are used in the extraction of available S, such as mixture of salts and weak acids, or extraction by ion exchange resin (Anderson *et al.*, 1992; Prochnow *et al.*, 1997; Ribeiro *et al.*, 1991). Currently, the most common solutions for extracting available S in Brazilian soils are 0.01 M calcium phosphate ($\text{Ca}(\text{H}_2\text{PO}_4)_2$), initially proposed by Fox *et al.* (1964) and the standard extractant adopted in the state of São Paulo; and 0.5 M ammonium acetate (NH_4OAc) along with 0.25 M acetic acid (HOAc) (Bardsley and Lancaster, 1965). The phosphate ion has more affinity for adsorption than the sulphate ion so that $\text{Ca}(\text{H}_2\text{PO}_4)_2$ solution will extract more of the $\text{SO}_4\text{-S}$ adsorbed to iron and aluminum oxides and hydroxides than the $\text{SO}_4\text{-S}$ adsorbed to soil organic fractions. In contrast, the NH_4OAc + HOAc solution will extract soluble and adsorbed S, and a portion of S linked to soil organic matter. This solution has been used because the adsorption of acetate is similar to $\text{SO}_4\text{-S}$ adsorption in soils (Tabatabai, 1987).

In Brazilian soils, there are few field studies focusing on the availability of S to plants, especially under NT systems. The objectives of this study were to (i) evaluate the

nutrient uptake and S availability in an NT soil to the maize and wheat crops provided by PG applications, and (ii) establish critical levels of S in soil for the crops using $\text{Ca}(\text{H}_2\text{PO}_4)_2$ and $\text{NH}_4\text{OAc} + \text{HOAc}$ extractant solutions. Since NT systems should increase organic S, which is also an S source, in deficient periods, we hypothesized that $\text{NH}_4\text{OAc} + \text{HOAc}$ would be a better extractant than $\text{Ca}(\text{H}_2\text{PO}_4)_2$ for assessing available soil S, especially in surface layers, where the organic matter content is higher than in subsurface layers.

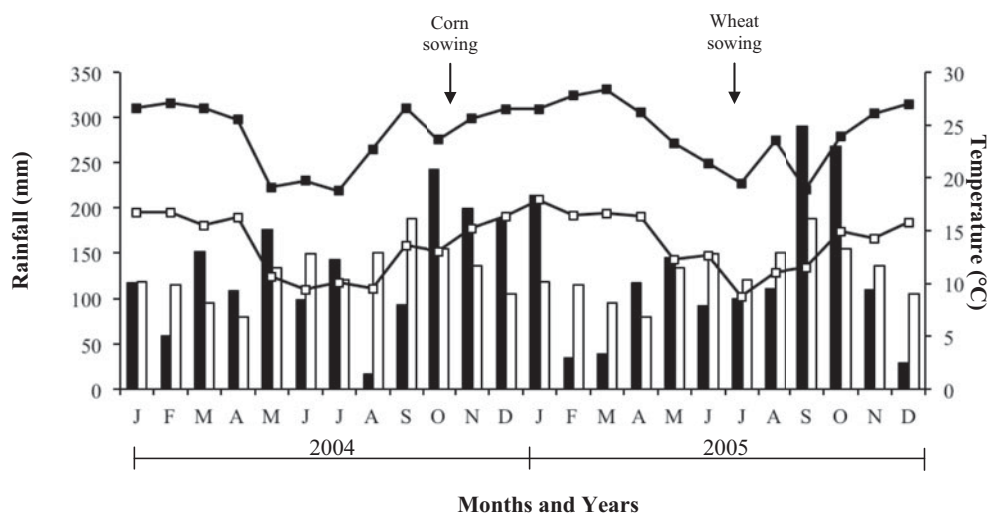
MATERIAL AND METHODS

The experiment was undertaken in Ponta Grossa, Parana state, Brazil (25°08'50" S, 50°06'20" W, 819 m a.s.l.) on an Oxisol (clay, kaolinitic, Rhodic Hapludox), previously used as pastureland. The trial area was dominated by the grass *Paspalum notatum* cultivar Pensacola, and it had no history of liming or fertilizer applications. In 1998, at the beginning of the experiment, soil chemical analyses of the topsoil (0–0.20 m) showed the following results: pH (1:2.5 soil/0.01 M CaCl_2 suspension) of 4.6; exchangeable Al, Ca, Mg and K contents of 3, 25, 20 and 3.6 $\text{mmol}_c \text{ dm}^{-3}$ respectively; total acidity at pH 7.0 (H + Al) of 78 $\text{mmol}_c \text{ dm}^{-3}$, P (Mehlich-1) of 0.3 mg dm^{-3} ; total organic C (by the Walkley–Black method) of 31 g dm^{-3} and base saturation of 38%. Particle-size distribution analyses revealed 580, 130 and 290 g kg^{-1} of clay, silt and sand respectively. A more detailed description of the experiment site can be found in Caires *et al.* (2011a).

Dolomitic lime at a rate of 4.5 t ha^{-1} was applied on the soil surface in July 1998. The lime rate was calculated to raise the base saturation at the topsoil (0–0.20 m) to 70% (Caires *et al.*, 2005). The dolomitic lime used contained 224 g kg^{-1} Ca, 140 g kg^{-1} Mg and 89% effective calcium carbonate equivalent. An NT system was established for the grain crop production involving no disturbance to the soil other than the sowing operation.

A randomised complete block design was used, with three replications in a split-plot arrangement. Plot size was 8 × 7 m and subplot size was 4 × 7 m. The treatments consisted of PG application on the soil surface at 0 and 6 t ha^{-1} in September 2004 to subplots within plots with previously surface-applied PG at 0, 3, 6 and 9 t ha^{-1} in October 1998. Phosphogypsum, a by-product obtained from a Brazilian phosphate fertilizer industry, contained 235 g kg^{-1} Ca, 153 g kg^{-1} S, 3 g kg^{-1} P and 156 g kg^{-1} water. The sequence of crops up to 2004 was as follows: soybean (*Glycine max* (L.) Merr.) in 1998–1999, barley (*Hordeum vulgare* L.) in 1999, soybean in 1999–2000, wheat (*Triticum aestivum* L.) in 2000, soybean in 2000–2001, maize (*Zea mays* L.) in 2001–2002 and soybean in 2002–2003 and 2003–2004. At the time of PG reapplication in 2004, soils had low acidity and non-toxic levels of exchangeable Al^{3+} as a function of the surface liming performed before the establishment of the experiment in 1998.

In October 2004, maize (hybrid DA 2A120) was sown at a seeding rate of six seeds m^{-2} , and a row spacing of 0.80 m. Fertilizers were applied at the rates of 114 kg ha^{-1} N (28 kg ha^{-1} N at planting and 86 kg ha^{-1} N as urea in top-dressing), 43 kg ha^{-1} P and 69 kg ha^{-1} K. In July 2005, wheat (cultivar OR 1) was sown at a seeding rate



■ Rainfall □ Rainfall, long-term (45 years) average ■- Maximum temperature □- Minimum temperature

Figure 1. Monthly rainfall (black areas), minimum (open square) and maximum (closed square) temperatures during the development period of the crops (2004–2005) and the 45-year average monthly rainfall (white areas) in Ponta Grossa, Southern Brazil.

of 60 seeds m^{-1} , and a row spacing of 0.17 m. Fertilizers were applied at the rates of 124 kg ha^{-1} N (24 kg ha^{-1} N at planting and 100 kg ha^{-1} N as urea in top dressing), 21 kg ha^{-1} P and 66 kg ha^{-1} K. These fertilizer rates were applied according to the soil test recommendations for the Parana state (Oliveira, 2003); in addition, other plant treatments and weed control were performed according to each crop's needs.

Seasonal rainfall and air temperature data for the duration of the experiment conducted are shown in Figure 1. Throughout the development period of the crops, air temperatures were normal. In the winter of 2005, there was an excessive rainfall, which caused decrease in wheat grain yield (Caires *et al.*, 2011a).

Maize (2004–2005) and wheat (2005) leaf samples (30 leaves per subplot) were collected during the flowering period by taking leaf samples from below and opposite to the ear for maize, and the flag leaf for wheat. The samples were washed in deionized water, dried in a forced-air oven at 60 °C until a constant weight was achieved and subsequently ground. At physiological maturity, aboveground biomass samples of maize (five plants per subplot) and wheat (one linear meter of plants) were also collected to measure total dry matter and S concentration. At the end of the cycle of maturation, crops were manually harvested and grain yields were determined at 130 g kg^{-1} moisture. Grain S content was also measured. Sulphur uptake and S exports were calculated by multiplying S concentration in aboveground biomass and grains by dry matter and grain yield respectively. The analyses of S in leaf tissues, dry matter and grains were performed using nitric–perchloric acid digestion, and measured using turbidimetry (as $BaSO_4$).

Soil samples were taken in May 2005 and 2006. To obtain a composite sample, 12 soil cores were taken at 0 to 0.05, 0.05 to 0.10 and 0.10 to 0.20-m depths, and five soil cores were sampled at 0.20 to 0.40 and 0.40 to 0.60-m depths in each subplot using a soil probe. Before chemical analysis, soils were air-dried and ground to pass through a 2-mm sieve. Sulphate was extracted with 0.01 M $\text{Ca}(\text{H}_2\text{PO}_4)_2$ using a 1:2.5 (v/v) soil/solution ratio and later determined turbidimetrically as BaSO_4 . The soil $\text{SO}_4\text{-S}$ content extracted with 0.5 M NH_4OAc along with 0.25 M HOAc was presented by Caires *et al.* (2011a). The extractant 0.01 M $\text{Ca}(\text{H}_2\text{PO}_4)_2$, initially proposed by Fox *et al.* (1964), is indicated for very weathered soils (Tabatabai, 1987), since the amount of phosphate present in the solution can replace most of the sulphate adsorbed in these soils, thus extracting the soluble + adsorbed fractions. NH_4OAc along with HOAc (Bardsley and Lancaster, 1965) extracts S that is soluble, adsorbed and part of the easily oxidizable organic fraction. Acetate is similar to sulphate in adsorption characteristics, which makes this solution a good extractor to predict $\text{SO}_4\text{-S}$ availability (Ribeiro *et al.*, 2001).

Data were analysed through an analysis of variance and simple regression methods using a randomized complete block design in a split-plot arrangement. The effects of the surface-applied gypsum rates in 1998 were analysed through regression analyses, and the criterion for choosing the model was the magnitude of the determination coefficients significant at $p < 0.05$. The effects of gypsum application in 2004 were compared using the F-test. Pearson product moment correlations were employed between soil-extracted S and plant S. Critical levels of $\text{SO}_4\text{-S}$ for achieving 90% of relative grain yield were evaluated for both crops by adjusting an inverse first-order equation ($\hat{y} = y_0 + a/x$) between soil $\text{SO}_4\text{-S}$ levels (at 0–0.20 m) and relative crop yield, using SigmaPlot v11 (Systat Software Inc., Chicago, IL).

RESULTS

Soil-extractable Sulphur

The two extractants were positively correlated in the evaluation of available soil $\text{SO}_4\text{-S}$ (Figure 2) for both topsoil (0–0.20 m, $r = 0.98$) and all soil profiles (0–0.60 m, $r = 0.96$). The close relationships between $\text{SO}_4\text{-S}$ extracted by 0.01 M $\text{Ca}(\text{H}_2\text{PO}_4)_2$ and 0.5 M NH_4OAc along with 0.25 M HOAc indicated that both extractants had similar efficacy for determining available $\text{SO}_4\text{-S}$ in an Oxisoil with high clay and organic C contents. Since the slope of the equation was greater than 1, the extractant 0.5 M NH_4OAc along with 0.25 M HOAc extracted a little more $\text{SO}_4\text{-S}$ compared with 0.01 M $\text{Ca}(\text{H}_2\text{PO}_4)_2$, mainly on the soil surface layers, which have higher organic matter content. Since at the deepest soil layers organic matter content is lower and most of the sulphate is adsorbed into Fe- and Al-oxides and hydroxides, differences for both extractants in extracting sulphate in all soil profiles (0–0.60 m) were smaller.

The soil $\text{SO}_4\text{-S}$ contents extracted by 0.01 M $\text{Ca}(\text{H}_2\text{PO}_4)_2$ have shown residual effects of PG, 6.5 (2005) and 7.5 (2006) years after its application (Figure 3). Comparing this result with the soil $\text{SO}_4\text{-S}$ extracted by 0.5 M NH_4OAc along with 0.25 M HOAc (Caires *et al.*, 2011a), we observed that the effects of PG application on extractable

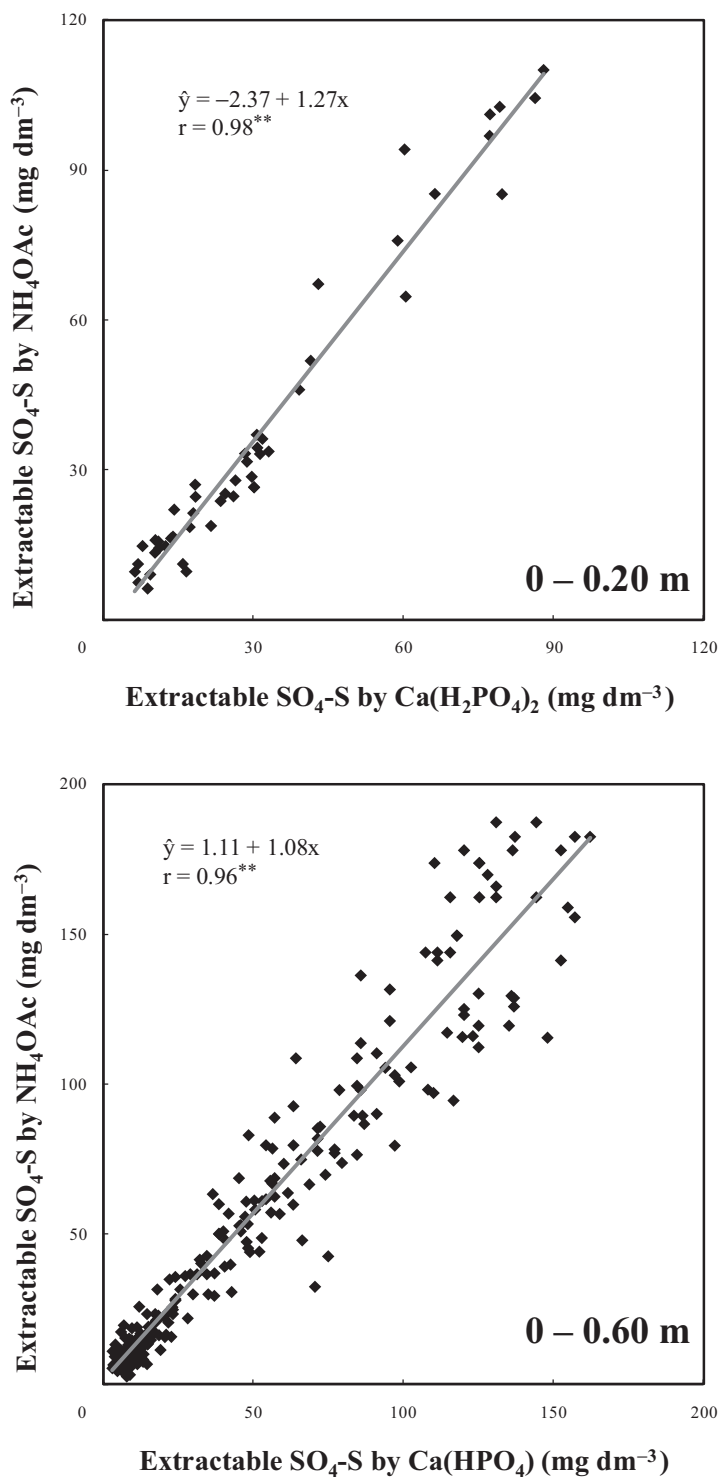


Figure 2. Correlation between chemical extractants for soil $\text{SO}_4\text{-S}$ (0.01 M $\text{Ca}(\text{H}_2\text{PO}_4)_2$ and 0.5 M NH_4OAc in 0.25 M HOAc). $^{**}p < 0.01$.

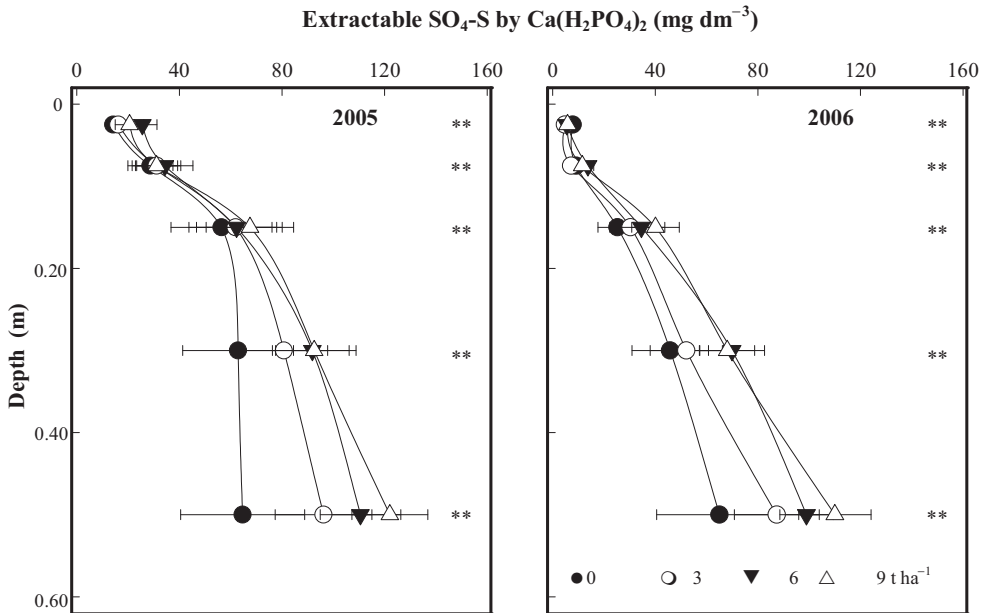


Figure 3. Extractable $\text{SO}_4\text{-S}$ by 0.01 M $\text{Ca}(\text{H}_2\text{PO}_4)_2$ at different depths, following phosphogypsum (PG) application on the soil surface in 1998. ** $p < 0.01$ by linear regression.

$\text{SO}_4\text{-S}$ content were relatively similar with both extractants, and the most pronounced effects occurred on subsoil layers. The soil $\text{SO}_4\text{-S}$ contents at 0.20–0.40 and 0.40–0.60-m depths after PG application at 9 t ha⁻¹ were respectively 90 and 120 mg dm⁻³ by extraction with 0.01 M $\text{Ca}(\text{H}_2\text{PO}_4)_2$ (Figure 3) and 130 and 145 mg dm⁻³ by extraction with 0.5 M NH_4OAc + 0.25 M HOAc (Caires *et al.*, 2011a) in 2005, and 70 and 110 mg dm⁻³ by extraction with 0.01 M $\text{Ca}(\text{H}_2\text{PO}_4)_2$ (Figure 3) and 70 and 105 mg dm⁻³ by extraction with 0.5 M NH_4OAc + 0.25 M HOAc (Caires *et al.*, 2011a) in 2006.

Results have shown great movement of $\text{SO}_4\text{-S}$ in the soil eight months after PG reapplication (2005), consequently increasing available soil $\text{SO}_4\text{-S}$ levels (Figure 4) in all sampled layers. Higher evidence of $\text{SO}_4\text{-S}$ movement was observed 1.5 years after PG reapplication (2006). Higher $\text{SO}_4\text{-S}$ concentrations, as measured by both extractants, were found in subsoil layers (Figure 4). The soil $\text{SO}_4\text{-S}$ contents at 0.20–0.40 and 0.40–0.60-m depths after PG reapplication at 6 t ha⁻¹ were respectively 120 and 140 mg dm⁻³ by extraction with 0.01 M $\text{Ca}(\text{H}_2\text{PO}_4)_2$ (Figure 4) and 165 and 160 mg dm⁻³ by extraction with 0.5 M NH_4OAc + 0.25 M HOAc (Caires *et al.*, 2011a) in 2005, and 85 and 125 mg dm⁻³ by extraction with 0.01 M $\text{Ca}(\text{H}_2\text{PO}_4)_2$ (Figure 4) and 90 and 115 mg dm⁻³ by extraction with 0.5 M NH_4OAc + 0.25 M HOAc (Caires *et al.*, 2011a) in 2006. The higher affinity for anion phosphate, lower soil acidity and higher cation exchange capacity on the soil surface layers might explain the lower $\text{SO}_4\text{-S}$ contents on the soil surface. Of the total $\text{SO}_4\text{-S}$ provided by the reapplication of 6 t ha⁻¹ of PG (480 mg dm⁻³), 37% was recovered (180 mg dm⁻³) by extraction with

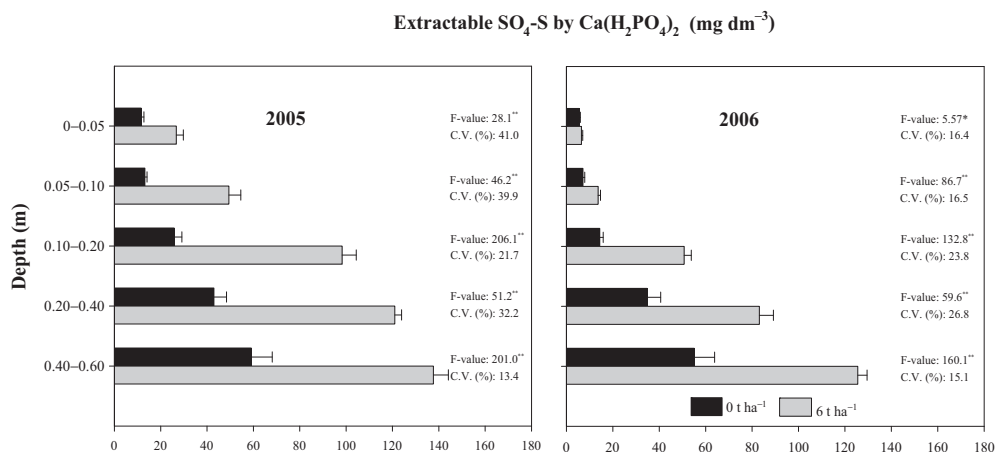


Figure 4. Extractable SO₄-S by 0.01 M Ca(H₂PO₄)₂ in different soil depths following phosphogypsum (PG) reapplication on the soil surface in 2004. * $p < 0.05$ and ** $p < 0.01$.

Ca(H₂PO₄)₂ and 56% (270 mg dm⁻³) with NH₄OAc + HOAc (Caires *et al.*, 2011a) at 0 to 0.60-m depth. Only a small fraction of the total applied SO₄-S was taken up by the crops or lost by erosion through runoff, indicating substantial movement to deeper layers beyond the depth of 0.60 m.

Effects of Sulphur on maize and wheat plants

The attributes related to maize and wheat nutrition and aboveground biomass were not influenced by the interaction between the surface-applied PG rates in 1998 and 2004. Maize total aboveground biomass, evaluated at physiological maturity, was affected by PG treatments with linear response. Wheat aboveground biomass (\hat{y} , in kg ha⁻¹) increased with PG rates (x , in t ha⁻¹) ($\hat{y} = 3975.14 + 600.18x - 65.12x^2$, $R^2 = 0.99$). The maximum wheat dry mass production was obtained with the estimated rate of 4.6 t ha⁻¹ of PG. Besides increasing biomass production, wheat plants also showed increased accumulation of S in aboveground biomass and leaves, even after 7.5 years of PG application (Table 1). Other plant nutrients accumulated on the aboveground biomass, such as Mg in the maize plants and N, K, Ca and Mg in the wheat plants, were also influenced by PG application (Table 2).

Reapplication of PG in 2004 increased total aboveground biomass, S uptake, leaf S content and S exports by the maize crop (Table 3). Wheat harvests in 2005 were also increased by PG reapplication with positive responses on aboveground biomass production, S uptake and leaf S content (Table 3). Reapplication of PG increased the uptake of N and P on maize plants, and that of N, K, Ca and Mg on wheat plants (Table 4).

Sulphur uptake by plants and S concentration of maize and wheat leaves were more sensitive indicators of increase of S in the soil than was the S exported by grains (Table 5). Correlating the available soil S-SO₄ content with plant parameters, we observed that both extractants were very similar (Table 5).

Table 1. Dry matter and S contents in maize and wheat plants following phosphogypsum (PG) application on the soil surface in 1998.

PG (t ha ⁻¹)	Dry matter (kg ha ⁻¹)	S uptake (kg ha ⁻¹)	Leaf S (g kg ⁻¹)	Grain S (g kg ⁻¹)	S export (kg ha ⁻¹)
Maize 2004–2005					
0	20,895 (±384)	15.6(±1.25)	2.3 (±0.22)	0.9 (±0.04)	9.1 (±0.42)
3	21,012 (±656)	13.7 (±1.58)	2.2 (±0.21)	0.8 (±0.07)	8.0 (±0.80)
6	21,120 (±870)	15.1 (±1.62)	2.3 (±0.21)	0.9 (±0.04)	8.4 (±0.63)
9	22,212 (±829)	15.5 (±1.93)	2.6 (±0.25)	0.9 (±0.05)	9.1 (±0.63)
Effect	L*	NS	NS	NS	NS
R ²	0.74	–	–	–	–
C.V. (%)	4.4	21.4	16.3	11.6	21.4
Wheat 2005					
0	4000 (±328)	9.4 (±1.06)	3.1 (±0.23)	1.1 (±0.05)	2.1 (±0.19)
3	5114 (±735)	12.9 (±2.38)	3.2 (±0.24)	1.1 (±0.04)	2.1 (±0.17)
6	5307 (±585)	13.8 (±1.76)	3.5 (±0.21)	1.0 (±0.02)	1.9 (±0.16)
9	4077 (±387)	9.1 (±0.86)	3.4 (±0.15)	1.1 (±0.03)	2.3 (±0.23)
Effect	Q*	Q**	L**	NS	NS
R ²	0.99	0.97	0.75	–	–
C.V. (%)	22.2	24.8	5.7	7.7	19.9

NS: non-significant; * $p < 0.05$ and ** $p < 0.01$; L: significant by linear regression; Q: significant by quadratic regression. Standard errors ($n = 6$) are given in parentheses.

S uptake and S exports were calculated by multiplying S concentration in aboveground biomass and grains by dry matter and grain yield respectively.

Table 2. Uptake of macronutrients in maize and wheat plants following phosphogypsum (PG) application on the soil surface in 1998.

PG (t ha ⁻¹)	N (kg ha ⁻¹)	P (kg ha ⁻¹)	K (kg ha ⁻¹)	Ca (kg ha ⁻¹)	Mg (kg ha ⁻¹)
Maize 2004–2005					
0	217.8 (±5.32)	44.7 (±2.20)	240.7 (±18.24)	20.8 (±2.35)	59.0 (±3.73)
3	207.2 (±11.55)	42.5 (±2.68)	221.0 (±18.22)	20.3 (±3.19)	60.9 (±4.39)
6	206.5 (±9.42)	49.5 (±3.12)	249.6 (±25.01)	23.2 (±3.21)	63.2 (±5.05)
9	222.6 (±13.9)	51.8 (±6.56)	285.5 (±26.46)	27.4 (±1.92)	78.1 (±5.74)
Effect	NS	NS	NS	NS	L
R ²	–	–	–	–	0.78
C.V. (%)	8.2	18.0	20.1	30.4	18.2
Wheat 2005					
0	106.6 (±8.77)	18.4 (±4.35)	110.7 (±16.98)	7.0 (±0.84)	20.7 (±1.88)
3	124.8 (±19.29)	18.2 (±3.42)	163.4 (±27.60)	8.7 (±1.84)	24.0 (±3.24)
6	132.7 (±15.04)	25.4 (±5.44)	190.0 (±39.04)	10.2 (±1.81)	26.2 (±3.33)
9	101.6 (±10.38)	15.7 (±3.51)	125.2 (±26.15)	6.7 (±0.75)	18.4 (±1.65)
Effect	Q*	NS	Q*	Q*	Q*
R ²	0.94	–	0.95	0.85	0.89
C.V. (%)	23.3	38.6	33.3	30.1	24.4

NS: non-significant; * $p < 0.05$ and ** $p < 0.01$; L: significant by linear regression; Q: significant by quadratic regression. Standard errors ($n = 6$) are given in parentheses.

Nutrient uptake was calculated by multiplying the concentration in aboveground biomass by dry matter.

Table 3. Dry matter and S content in maize and wheat plants after phosphogypsum (PG) reapplication on the soil surface in 2004.

PG (t ha ⁻¹)	Dry matter (kg ha ⁻¹)	S uptake (kg ha ⁻¹)	Leaf S (g kg ⁻¹)	Grain S (g kg ⁻¹)	S export (kg ha ⁻¹)
Maize 2004–2005					
0	20,373 (±290)	12.6 (±0.56)	2.0 (±0.10)	0.8 (±0.04)	7.6 (±0.31)
6	22,246 (±517)	17.4 (±1.05)	2.7 (±0.12)	0.9 (±0.02)	9.7 (±0.33)
F-value	8.2*	13.6**	32.3**	10.7*	23.6**
C.V. (%)	7.5	21.5	13.7	11.3	11.8
Wheat 2005					
0	4180 (±344)	8.8 (±0.69)	3.1 (±0.14)	1.1 (±0.02)	1.9 (±0.10)
6	5069 (±410)	13.8 (±1.25)	3.5 (±0.13)	1.1 (±0.03)	2.3 (±0.14)
F-value	8.0*	26.1**	48.2**	0.76 (NS)	4.9 (NS)
C.V. (%)	16.7	21.4	4.0	9.6	21.8

NS: non-significant; * $p < 0.05$ and ** $p < 0.01$.

Standard errors (n = 12) are given in parentheses.

S uptake and S exports were calculated by multiplying S concentration in aboveground biomass and grains by dry matter and grain yield respectively.

Table 4. Uptake of macronutrients in maize and wheat plants after phosphogypsum (PG) reapplication on the soil surface in 2004.

PG (t ha ⁻¹)	N (kg ha ⁻¹)	P (kg ha ⁻¹)	K (kg ha ⁻¹)	Ca (kg ha ⁻¹)	Mg (kg ha ⁻¹)
Maize 2004–2005					
0	200.6 (±4.65)	42.7 (±1.37)	229.6 (±13.94)	21.0 (±2.06)	62.1 (±3.07)
6	226.2 (±7.57)	51.5 (±3.45)	268.7 (±16.97)	24.9 (±1.81)	68.5 (±4.45)
F-value	10.7*	5.6*	2.8	1.1	1.7
C.V. (%)	9.1	19.3	23.0	34.0	18.5
Wheat 2005					
0	102.8 (±7.88)	17.8 (±2.76)	119.7 (±17.21)	5.9 (±0.41)	19.8 (±1.49)
6	130.1 (±10.56)	21.0 (±3.28)	175.0 (±21.97)	10.4 (±1.03)	24.9 (±2.13)
F-value	6.8*	1.5	7.0*	30.1**	13.2**
C.V. (%)	22.0	32.8	34.8	24.8	15.4

* $p < 0.05$ and ** $p < 0.01$.

Standard errors (n = 12) are given in parentheses.

Nutrient uptake was calculated by multiplying the concentration in aboveground biomass by dry matter.

The critical level of SO₄-S for achieving 90% of relative grain yield was evaluated for both crops, but results were significant only for the maize crop. The obtained results were very similar for both extractants (Figure 5), estimated as 16.2 mg dm⁻³ by 0.01 M Ca(H₂PO₄)₂ and 15.5 mg dm⁻³ by 0.5 M NH₄OAc along with 0.25 M HOAc.

DISCUSSION

Soil-extractable Sulphur

The solutions of NH₄OAc + HOAc and Ca(H₂PO₄)₂ employed in our study extract soluble and adsorbed SO₄-S (Bardsley and Lancaster, 1965; Fox *et al.*, 1964).

Table 5. Correlation coefficients (Pearson) between soil S extracted by $\text{Ca}(\text{H}_2\text{PO}_4)_2$ and NH_4OAc and foliar S, S uptake by maize and wheat plants and S exported by crop harvest ($n = 24$).

Depth (m)	Foliar S		S uptake		S export	
	NH_4OAc	$\text{Ca}(\text{H}_2\text{PO}_4)_2$	NH_4OAc	$\text{Ca}(\text{H}_2\text{PO}_4)_2$	NH_4OAc	$\text{Ca}(\text{H}_2\text{PO}_4)_2$
Maize 2004–2005						
0–0.05	0.58**	0.59**	0.25 (NS)	0.44*	0.25 (NS)	0.35*
0.05–0.10	0.57**	0.50**	0.64**	0.59**	0.44*	0.38*
0.10–0.20	0.66**	0.57**	0.63**	0.56**	0.58**	0.52**
0.20–0.40	0.69**	0.61**	0.62**	0.57**	0.64**	0.58**
0.40–0.60	0.63**	0.71**	0.48**	0.55**	0.50**	0.62**
0–0.20	0.66**	0.58**	0.62**	0.58**	0.54**	0.49**
Wheat 2005						
0–0.05	0.59**	0.27 (NS)	0.44*	0.05 (NS)	–0.12 (NS)	0.11 (NS)
0.05–0.10	0.50**	0.46**	0.68**	0.62**	0.27 (NS)	0.34*
0.10–0.20	0.52**	0.49**	0.72**	0.62**	0.47*	0.53**
0.20–0.40	0.41*	0.45*	0.62**	0.56**	0.23 (NS)	0.26 (NS)
0.40–0.60	0.49**	0.48**	0.54**	0.52**	0.42*	0.44*
0–0.20	0.55**	0.50**	0.74**	0.63**	0.44*	0.51**

NS: non-significant; * $p < 0.05$ and ** $p < 0.01$.

The phosphate ion has more affinity for adsorption than the sulphate ion so that $\text{Ca}(\text{H}_2\text{PO}_4)_2$ solution extracts more of the $\text{SO}_4\text{-S}$ adsorbed to Fe- and Al-oxides and hydroxides than the $\text{SO}_4\text{-S}$ adsorbed to soil organic fractions. In contrast, the $\text{NH}_4\text{OAc} + \text{HOAc}$ solution extracts soluble and adsorbed S and a portion of S linked to soil organic matter. In our study, both extractants had similar efficacy for determining available $\text{SO}_4\text{-S}$ in an Oxisoil with high clay and organic C contents (Figure 2). Because of high organic matter content, especially in topsoil (0–0.20 m), the extractant 0.5 M NH_4OAc along with 0.25 M HOAc extracted a little more $\text{SO}_4\text{-S}$ compared with 0.01 M $\text{Ca}(\text{H}_2\text{PO}_4)_2$. Ribeiro *et al.* (2001) observed that either NH_4OAc or $\text{Ca}(\text{H}_2\text{PO}_4)_2$ was efficient as soil $\text{SO}_4\text{-S}$ extractant, showing high correlation with S uptake by sorghum plants. However, in soils with a low $\text{SO}_4\text{-S}$ adsorption capacity, the more efficient extractant was 0.01 M CaCl_2 , showing that S availability in these soils was mainly driven by the mineralization of organic forms (Ribeiro *et al.*, 2001).

Sulphate adsorption capacity has been positively related to Fe- and Al-oxide contents (Alves and Lavorentti, 2004; Chao *et al.*, 1964), and negatively related to pH level (Zhang and Yu, 1997) as well as to organic matter content (Couto *et al.*, 1979). Residual effects of PG application on soil $\text{SO}_4\text{-S}$ availability were found in some long-term studies performed in soils with a pH value of around 4.0 in subsurface layers (Farina *et al.*, 2000; Toma *et al.*, 1999). In our study, soil pH in CaCl_2 was 5.0 and ΔpH had negative values (varying from –0.9 to –0.6) in all soil profiles (0–0.60 m). Because of low acidity and high soil organic C contents determined by the Walkley–Black method, whose levels ranged from 27 g dm^{-3} (0–0.05 m) to 17 g dm^{-3} (0.40–0.60 m), the dominance of negative electric charges should have favoured the $\text{SO}_4\text{-S}$ movement from PG application (Figure 3) and re-application (Figure 4) to even deeper soil layers. However, even on an Oxisol with a high density of electrical charges

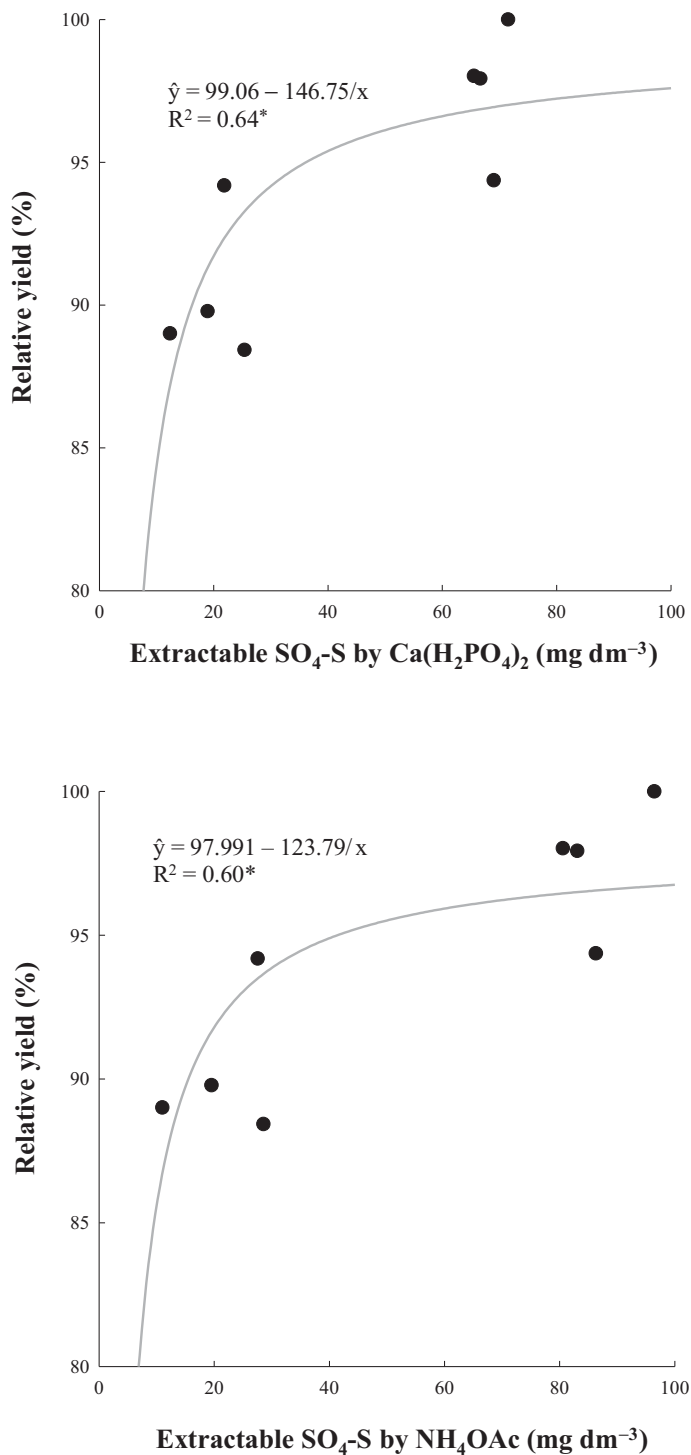


Figure 5. Relations between maize relative yield in 2004–2005 and soil SO₄-S content extracted by 0.01 M Ca(H₂PO₄)₂ and 0.5 M NH₄OAc in 0.25 M HOAc. * $p < 0.05$.

due to high organic C contents throughout the soil profile, a considerable amount of PG sulphate remained in the subsoil for several years, which is relevant information for diagnosing availability of S in soils.

Effects of Sulphur on maize and wheat plants

By the time of cropping, maize and wheat crops showed no visual signs of S deficiency. Even so, application of PG in 1998 increased leaf S content and total S uptake by the wheat crop in 2005 (Table 1). In addition, reapplication of PG in 2004 also increased leaf S content and total S uptake in both maize (2004–2005) and wheat (2005) crops (Table 3). Although grasses in general require less S compared with legumes or brassicas (Dick *et al.*, 2008), beneficial effects of the application of various S sources on S nutrition of grasses have been reported in the literature. Bullock and Goodroad (1989) observed an increase in S content in maize leaves and grain due to S application as ammonium sulphate or elemental S. An increase in maize grain S content was also observed by Ferreira *et al.* (2001) and Chen *et al.* (2008) with the application of ammonium sulphate and S from flue-gas desulphurisation respectively. In a greenhouse study using isotopic labeling S, Silva *et al.* (2003) found that corn and soybean showed different behaviours in S absorption and redistribution. In spite of having larger root absorption, corn retained great part of that S in the root, while soybean absorbed considerably less, but presented greater translocation efficiency. This result may also help explain the fact that maize more often responds to S application than soybean. Another reason for the greater response of grasses to S additions could be a function of plant breeding, which provided more productive, and also more N demanding, varieties.

The interaction between N \times S is reported in the literature, since both nutrients are the components of amino acids and essential for protein synthesis (Epstein and Bloom, 2006). Therefore, plant S requirement is closely related to N nutrition (Tables 2 and 4) (Chen *et al.*, 2008). Nitrogen and S interaction was reported on aboveground dry mass production of *Brachiaria brizantha*, cv. Marandu, in a greenhouse trial (De Bona and Monteiro, 2010), on *Brachiaria decumbens* in field trial (Schmidt *et al.*, 2012) and on maize (Chen *et al.*, 2008) and wheat (Salvagiotti *et al.*, 2009) grain yield under field conditions. These studies have shown that S nutrition has promoted N uptake and increased N use efficiency by plants, which can reduce production costs and degradation of water quality associated with oversupply of N (Chen *et al.*, 2008; Salvagiotti *et al.*, 2009).

When studying soil extractants, the best method is the one that provides better correlation to plant uptake. Cui *et al.* (2006) evaluated different extractants in China pastures, and observed high correlation coefficients between soil S and S absorbed by the crop, especially with 0.01 M Ca(H₂PO₄)₂ as extractant. However, the authors pointed out that in their study the correlation coefficients between soil S and leaf S were unsatisfactory for field experimentation, which did not occur in our study.

Many solutions can be used as SO₄-S extractants in Brazilian soils to obtain the available contents, but the most frequently used are Ca(H₂PO₄)₂ and NH₄OAc along with HOAc (Caires *et al.*, 2011a; Prochnow *et al.*, 1997; Ribeiro *et al.*, 2001). Since

the choice of an extractant is a function of its physical and chemical characteristics (Anderson *et al.*, 1992; Blair *et al.*, 1991) and its correlation to the plant, in our study the extractant $\text{Ca}(\text{H}_2\text{PO}_4)_2$ could be replaced by $\text{NH}_4\text{OAc} + \text{HOAc}$, which is cheaper and subjected to less interference in quality depending on solubility. Correlations were positive in almost all soil layers studied. Moreover, the 0–0.20-m depth has been shown to be adequate for diagnosing available $\text{SO}_4\text{-S}$ content in the maize and wheat crops cultivated under NT. Prochnow and Boaretto (1995), studying available S in sicilian lemon, observed that S absorption occurs mainly in the 0–0.20-m depth, indicating that this depth must be considered to evaluate the available soil $\text{SO}_4\text{-S}$ content. Besides, sampling at just 0.20-m depth has practical advantages for farmers, and therefore increases the possibility of adopting methods of S diagnosis, which may increase overall S use efficiency.

A possible involvement in acid–subsoil amelioration and/or cationic balance on crop response to PG apart from S supply is difficult to exclude entirely in our study. In addition, PG contained P (3 g kg^{-1}) in its composition. Increasing the surface-applied PG rate in 1998 linearly increased the uptake of Mg by maize and raised the uptake of N, K, Ca and Mg by wheat (Table 2). Besides, PG reapplication in 2004 increased the uptake of N and P by maize, and the uptake of N, K, Ca and Mg by wheat (Table 4). The displacement of other cations from the exchange complex is the main effect of PG applications to soils. Therefore, Mg and K leaching has often been observed in the studies involving PG applications (Caires *et al.*, 2011b; Farina *et al.*, 2000; Oliveira and Pavan, 1996). The leaching of exchangeable Mg after PG application can be beneficial for plant nutrition, increasing the uptake of Ca and K (Tables 2 and 4), especially when the soil has high Mg levels and a low Ca/Mg ratio in the most superficial layers (Caires *et al.*, 2004, 2011b), as observed in our study. Increase in Mg uptake by maize and wheat crops as a result of PG application in 1998 (Table 2) and by the wheat crop after PG reapplication in 2004 (Table 4) was certainly due to gains in the aboveground biomass of plants promoted by PG applications (Tables 1 and 3). Although PG is also effective at improving P plant nutrition under NT systems (Caires *et al.*, 2003; 2011b), the P uptake only increased in the maize crop after PG reapplication in 2004 (Table 4).

Despite complex changes in soil chemical properties and plant nutrition as a result of PG application, because wheat and corn grain yields were positively related only to the Ca and S leaf contents, it was considered more probable that yield benefits imparted by PG resulted from increased Ca and $\text{SO}_4\text{-S}$ availability to cereal crops (Caires *et al.*, 2011a). In an attempt to isolate the effect of soil $\text{SO}_4\text{-S}$ availability from PG to cereal crops, the results showed that the critical levels of $\text{SO}_4\text{-S}$ in the topsoil (0–0.20 m) for the NT maize crop were 16.2 mg dm^{-3} by 0.01 M $\text{Ca}(\text{H}_2\text{PO}_4)_2$, and 15.5 mg dm^{-3} by 0.5 M NH_4OAc along with 0.25 M HOAc (Figure 5). Since root length of the maize and wheat crops under NT is concentrated at 65–80% in the topsoil (0–0.20 m) (Caires *et al.*, 2002a, b, 2008), this soil layer was demonstrated to be sufficient to determine available $\text{SO}_4\text{-S}$ content to maize and wheat crops in this cropping system. Since closed relationships were obtained, these critical levels of $\text{SO}_4\text{-S}$ may be useful for NT farming because many doubts remained concerning the establishment of critical levels of $\text{SO}_4\text{-S}$ in most soils of the world. For tropical soils

under conventional tillage systems, an $\text{SO}_4\text{-S}$ concentration of 10 mg dm^{-3} extracted with NH_4OAc or $\text{NH}_4\text{OAc} + \text{HOAc}$ has been proposed as a critical level (Kang and Osiname, 1976; Vitti, 1989). An increase in required $\text{SO}_4\text{-S}$ levels is expected nowadays, especially for the maize crop, which has experienced a twofold increase in yield over the last two decades (IBGE, 2012; Wallington *et al.*, 2012) due to improved combined agronomic practices such as breeding and higher inputs of fertilizers, mainly N, thus exporting more nutrients and requiring higher S replenishment than before.

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REFERENCES

- Alves, M. E. and Lavorentti, A. (2004). Sulfate adsorption and its relationships with properties of representative soils of the São Paulo State, Brazil. *Geoderma* 118:89–99.
- Anderson, G., Lefroy, R. N., Chinoin, N. and Blair, G. (1992). Soil sulfur testing. *Sulphur in Agriculture* 16:6–14.
- Bardsley, C. E. and Lancaster, J. D. (1965). Sulphur. In *Methods of Soil Analysis: Chemical and Microbiological Properties*, 1103–1116. (Ed. C. A. Black). Madison, WI: ASA.
- Blair, G. J., Chinoin, N., Lefroy, R. D. B., Anderson, G. C. and Crocker, G. J. (1991). A soil sulfur test for pastures and crops. *Australian Journal of Soil Research* 29:619–626.
- Bullock, D. G. and Goodroad, L. L. (1989). Effect of sulphur rate, application method, and source on yield and mineral content of corn. *Communications in Soil Science and Plant Analysis* 20:1209–1217.
- Caires, E. F., Alleoni, L. R. F., Cambri, M. A. and Barth, G. (2005). Surface application of lime for crop grain production under a no-till system. *Agronomy Journal* 97:791–798.
- Caires, E. F., Barth, G., Garbuio, F. J. and Kusman, M. T. (2002a). Influence of surface lime application in a no-tillage system on soil acidity, corn root growth and nutrition. *Revista Brasileira de Ciência do Solo* 26:1011–1022. (In Portuguese with abstract in English)
- Caires, E. F., Blum, J., Barth, G., Garbuio, F. J. and Kusman, M. T. (2003). Changes in chemical soil characteristics and soybean response to lime and gypsum applications in a no-tillage system. *Revista Brasileira de Ciência do Solo* 27:275–286. (In Portuguese with abstract in English)
- Caires, E. F., Churka, S., Garbuio, F. J., Ferrari, R. A. and Morgano, M. A. (2006). Soybean yield and quality as a function of lime and gypsum applications. *Scientia Agricola* 63:370–379.
- Caires, E. F., Feldhaus, I. C., Barth, G. and Garbuio, F. J. (2002b). Lime and gypsum application on the wheat crop. *Scientia Agricola* 59:357–364.
- Caires, E. F., Garbuio, F. J., Churka, S., Barth, G. and Corrêa, J. C. L. (2008). Effects of soil acidity amelioration by surface liming on no-till corn, soybean, and wheat root growth and yield. *European Journal of Agronomy* 28:57–64.
- Caires, E. F., Garbuio, F. J., Churka, S. and Joris, H. A. W. (2011a). Use of gypsum for crop grain production under a subtropical no-till cropping system. *Agronomy Journal* 103:1804–1814.
- Caires, E. F., Kusman, M. T., Barth, G., Garbuio, F. J. and Padilha, J. M. (2004). Changes in soil chemical properties and corn response to lime and gypsum applications. *Revista Brasileira de Ciência do Solo* 28:125–136. (In Portuguese with abstract in English)
- Caires, E. F., Maschietto, E. H. G., Garbuio, F. J., Churka, S. and Joris, H. A. W. (2011b). Surface application of gypsum in low acidic oxisol under no-till cropping system. *Scientia Agricola* 68:209–216.
- Chao, T. T., Harward, M. E. and Fang, S. C. (1964). Iron or aluminum coatings in relation to sulfate adsorption characteristics of soil. *Soil Science Society of America Proceedings* 28:623–625.
- Chen, L., Kost, D. and Dick, W. A. (2008). Flue gas desulfurization products as sulfur sources for corn. *Soil Science Society of America Journal* 72:1464–1470.

- Couto, W., Lathell, D. J. and Bouldin, J. R. (1979). Sulfate adsorption by two oxisols and alfisols of the tropics. *Soil Science* 127:108–116.
- Cui, X. Y., Wang, S. P., Wang, Y. F., Xing, X. R., Chen, Z. Z., Haneklaus, S., Fleckenstein, J. and Schnug, E. (2006). Suitability of extractants for predicting available sulfur in natural grassland in the Inner Mongolia Steppe of China. *Communications in Soil Science and Plant Analysis* 37:721–732.
- De Bona, F. D. and Monteiro, F. A. (2010). Nitrogen and sulfur fertilization and dynamics in a Brazilian Entisol under pasture. *Soil Science Society of America Journal* 74:1248–1258.
- Derpsch, R. and Friedrich, T. (2009). Global overview of conservation agriculture adoption. In *Proceedings of the 4th World Congress on Conservation Agriculture*, 429–438 (Ed. P. K. Joshi). New Delhi, India: Lead Papers.
- Dick, W. A., Kost, D. and Chen, L. (2008). Availability of sulfur to crops from soil and other sources. In *Sulfur: A Missing Link Between Soils, Crops and Nutrition*, 59–82 (Ed. J. Jez). Madison, WI: ASA.
- Epstein, E. and Bloom, A. J. (2005). *Mineral Nutrition of Plants: Principles and Perspectives*, 2nd edn. Sunderland, UK: Sinauer.
- Farina, M. P. W., Channon, P. and Thibaud, G. R. (2000). A comparison of strategies for amelioration subsoil acidity: II long-term soil effects. *Soil Science Society of America Journal* 64:652–658.
- Ferreira, A. C. B., Araújo, G. A., Pereira, P. R. G. and Cardoso, A. A. (2001). Corn crop characteristics under nitrogen, molybdenum and zinc fertilization. *Scientia Agricola* 58:131–138. (In Portuguese with abstract in English)
- Fox, R. L., Olson, R. A. and Rhoades, H. F. (1964). Evaluating the sulfur status of soil by plant and soil test. *Soil Science Society of America Proceedings* 28:243–246.
- Goh, K. M. and Pamidi, J. (2003). Plant uptake of sulphur as related to changes in the HI-reducible and total sulphur fractions in soil. *Plant & Soil* 250:1–13.
- IBGE (2012). *Levantamento Sistemático da Produção Agrícola*. Rio de Janeiro, Brazil: Fundação Instituto Brasileiro de Geografia e Estatística. Available at: <http://www.ibge.gov.br/home/estatistica/indicadores/agropecuaria/lspa/> (accessed 20 July 2012).
- Kang, B. and Osiname, O. (1976). Sulphur response of maize in Western Nigeria. *Agronomy Journal* 68:333–336.
- Kertesz, M. A. and Mirleau, P. (2004). The role of soil microbes in plant sulphur nutrition. *Journal of Experimental Botany* 55:1939–1945.
- Kost, D. A., Chen, L. and Dick, W. A. (2008). Predicting plant sulfur deficiency in soils: results from Ohio. *Biology and Fertility of Soils* 44:1091–1098.
- Lal, R. (2006). Enhancing crop yields in the developing countries through restoration of the soil organic carbon pool in agricultural lands. *Land Degradation and Development* 17:197–209.
- Lehmann, J., Solomon, D., Zhao, F. J. and McGrath, S. P. (2008). Atmospheric SO₂ emissions since the late 1800s change organic sulfur forms in humic substance extracts of soils. *Environmental Science and Technology* 42:3550–3555.
- Neptune, A. M. L., Tabatabai, M. A. and Hanway, J. J. (1975). Sulfur fractions and carbon-nitrogen-phosphorus relationships in some Brazilian and Iowa soils. *Soil Science Society of America Proceedings* 39:51–55.
- Oliveira, E. L. (2003). *Suggestion of Fertilization and Liming for Crops of Interest Economic at the Parana State*, IAPAR Circ. 128. Londrina, Brazil: Instituto Agronômico do Paraná (IAPAR), 30 pp. (In Portuguese)
- Oliveira, E. L. and Pavan, M. A. (1996). Control of soil acidity in no-tillage system for soybean production. *Soil & Tillage Research* 38:47–57.
- Prochnow, L. I. and Boaretto, A. E. (1995). Soil sampling depth for available sulphur evaluation in sicilian lemon (*Citrus limon* Burm. orchard). *Scientia Agricola* 52:101–106. (In Portuguese with abstract in English)
- Prochnow, L. I., Boaretto, A. E. and Vitti, G. C. (1997). Ion-exchange resin to evaluate sulphur availability in soils. *Revista Brasileira de Ciência do Solo* 21:335–339. (In Portuguese with abstract in English)
- Ribeiro, A. C., Accioly, L. J. O., Alvarez-Venegas, V. H., Braga, J. M. and Alves, V. M. C. (1991). Evaluation of soil available sulfur by the ion-exchange resin method. *Revista Brasileira de Ciência do Solo* 15:321–327. (In Portuguese with abstract in English)
- Ribeiro, E. S., Dias, L. E., Alvarez-Venegas, V. H., Mello, J. W. V. and Daniels, W. L. (2001). Dynamics of sulfur fractions in Brazilian soils submitted to consecutive harvests of sorghum. *Soil Science Society of America Journal* 65:787–794.
- Salvagiotti, F., Castellarin, J. M., Miralles, D. J. and Pedrol, H. (2009). Sulfur fertilization improves nitrogen use efficiency in wheat by increasing nitrogen uptake. *Field Crops Research* 113:170–177.
- Scherer, H. W. (2009). Sulfur in soils. *Journal of Plant Nutrition and Soil Science* 172:326–335.
- Schmidt, F., De Bona, F. D., Silveira, C. P. and Monteiro, F. A. (2012). Soil sulfur fractions dynamics and distribution in a tropical grass pasture amended with nitrogen and sulfur fertilizers. *Journal of Plant Nutrition and Soil Science* 175:60–67.

- Silva, D. J., Alvarez Venegas, V. H., Ruiz, H. A. and Sant'Anna, R. (2003). Translocation and redistribution of sulphur in corn and soybean plants. *Pesquisa Agropecuária Brasileira* 38:715–721. (In Portuguese with abstract in English)
- Solomon, D., Lehmann, J., Zarruk, K. K., Dathe, J., Kinyangy, J., Liang, B. and Machado, S. (2011). Speciation and long - and short-term molecular-level dynamics of soil organic sulfur studied by x-ray absorption near-edge structure spectroscopy. *Journal of Environmental Quality* 40:704–718.
- Sumner, M. E., Shahandeh, H., Bouton, J. and Hammel, J. (1986). Amelioration of an acid soil profile trough deep liming and surface application of gypsum. *Soil Science Society of America Journal* 50:1254–1258.
- Tabatabai, M. A. (1987). Physicochemical fate of sulfate in soils. *Journal of Journal of the Air Pollution Control Association* (Journal of the Air & Waste Management Association) 37:34–38.
- Tabatabai, M. A. (2005) Chemistry of sulfur in soils. In *Chemical Processes in Soils*, 193–226 (Eds. M. A. Tabatabai and D. L. Sparks). Madison, WI: SSSA.
- Toma, M., Sumner, M. E., Weeks, G. and Saigusa, M. (1999). Long-term effects of gypsum on crop yield and subsoil chemical properties. *Soil Science Society of America Journal* 39:891–895.
- Vitti, G. C. (1989). Sulfur in soil. In *Interpretation of Chemical Analysis of Soil and Plant for Fertilization*, 129–173 (Eds. L. T. Büll and C. A. Rosolem). Botucatu, Brazil: Studies and Research Foundation for Agriculture and Forestry. (In Portuguese)
- Wallington, T. J., Anderson, J. E., Mueller, S. A., Kolinski Morris, E. and Ginder, J. M. (2012). Corn ethanol production, food exports and indirect land use. *Environmental Science and Technology* 46:6379–6384.
- Zhang, G. Y. and Yu, T. R. (1997). Coordination adsorption of anions. In *Chemistry of Variable Charge Soils*, 175–218 (Ed. T. R. Yu). New York, NY: Oxford University Press.