




Article

Geochemistry of detrital zinc-rich chromite in conglomerates from eastern India

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Abstract

On the western part of the eastern Indian shield, Archaean basement (Bonai granite) is overlain by radioactive conglomerate. The conglomerate contains well-rounded, fractured Mg-poor (<0.18 wt.% MgO), Al-rich (up to 21.14 wt.% Al₂O₃) Zn-rich chromite (>35.0 wt.% Cr₂O₃) containing up to 15.5 wt.% ZnO. This is the first reported occurrence of detrital zinc-rich chromite having such unusually high ZnO from India, and to our knowledge, the third reported occurrence in radioactive quartz-pebble conglomerate after Witwatersrand, South Africa and Tarkwa, Ghana. Zinc-rich chromite grains are either clean or contain exsolution blebs/lamella of rutile. The conglomerates show evidence for post-depositional hydrothermal fluid influx and fluid-induced mineral alteration. The strong negative correlation between Zn and Fe²⁺, Al and Cr, and Al and Fe³⁺, and strong positive correlation between Zn and Al suggest secondary incorporation of Zn and Al by substitution of Fe²⁺ and Cr (and Fe³⁺), respectively, leading to partial transformation of (Fe)(Cr,Al,Fe³⁺)₂O₄ towards ZnAl₂O₄ composition. The chromite grains were possibly derived from komatiite. The timing of Zn enrichment, either at the provenance prior to sedimentation, or at the depositional site post-dating sedimentation, remains unresolved.

Keywords: zinc-rich chromite, detrital, conglomerate, geochemistry, alteration, origin, India

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Introduction

Zinc-rich spinel [(Zn,Fe,Mg)Al₂O₄] is a relatively common mineral, found in various geological settings and rock types, whereas zinc-rich chromite is rare. In this investigation, we use the term zinc-rich chromite to refer empirically to spinel that contains more than 35.0 wt.% Cr₂O₃ and more than 1.0 wt.% ZnO. Wylie *et al.* (1987) noted that Zn-rich chromite is unusual enough to be reported whenever found in nature. Zinc-rich chromite, containing up to 2.62 wt.% ZnO, was first reported in chromite ores by Donath (1931) from Ramberget, Norway and named subsequently as a new mineral ‘donathite’ by Seeliger and Mücke (1969). This name was later discredited by the International Mineralogical Association (Burns *et al.*, 1997). Electron probe microanalyses (EPMA) of samples with similar physical properties from the same region by Moore (1977) demonstrated that the chromite reported by Donath (1931) contains <0.3 wt.% ZnO and is therefore, not unusual. Some of the earliest reported and well-studied zinc-rich chromite containing more than a few wt.% ZnO originate from the ‘chromite dike’, i.e. serpentinised ultramafites, skarn and pelitic schist adjacent

to the ore zones in the Outokumpu mine, Finland (Thayer *et al.*, 1964; Knorring *et al.*, 1986; Wylie *et al.*, 1987; Weiser and Hirdes, 1997; Weiser *et al.*, 2008). The highest amount of ZnO (19.06 wt.%) in the zinc-rich chromite of Outokumpu was described from the pelitic schist by Wylie *et al.* (1987). Zinc-bearing and zinc-rich chromite have been described as inclusions in natural diamonds (Meyer and Boyd, 1972; Moorbath *et al.*, 1986; Griffin *et al.*, 1993; Taylor *et al.*, 2003; Donnelly *et al.*, 2007) and in meteorites (Bunch *et al.*, 1972; Chikami *et al.*, 1999). However, Zn-bearing and zinc-rich chromite have been described more commonly from metasomatised/serpentinised mafic–ultramafic rocks with/without associated magmatic/hydrothermal sulfide mineralisation (Groves *et al.*, 1977, 1983; Bevan and Mallinson, 1980; Donaldson and Bromley, 1981; Wylie *et al.*, 1987; Pan and Fleet, 1989, 1991; Zakrzewski, 1989; Béziat and Monchoux, 1991; Béziat *et al.*, 1993; Liipo *et al.*, 1995; Gahlan and Arai, 2007; Singh and Singh, 2013; Schulze *et al.*, 2014; Fanlo *et al.*, 2015; Matsumoto *et al.*, 2017). Béziat and Monchoux (1991) reported 27.16 wt.% ZnO in chromium-rich spinel containing 26.48 wt.% Cr₂O₃ in association with Au-bearing As–S mineralisation from metasomatised mafic rocks in the Salsigne district gold deposit, Montagne Noire, France. The Cr₂O₃ content in the grain containing the highest ZnO is much lower than typical chromite (<35.0 wt.% Cr₂O₃). Chromite grains containing >35.0 wt.% Cr₂O₃ in their investigation contains a maximum of 6.28 wt.%

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ZnO. Among other metasomatised rocks, zinc-rich chromite is described from calcsilicate skarns e.g.: Outokumpu (Knorring *et al.*, 1986); Hemlo area, Ontario Canada (Pan and Fleet, 1989); and the Bandihalli Supracrustal belt, Dharwar craton, India (Taguchi *et al.*, 2012). These skarns are interpreted to be metasomatised ultramafic rocks. Zinc-rich chromite in other igneous rocks or in their altered equivalents is rare. Wagner and Velde (1985) described zinc-rich chromite (with 8.74 wt.% ZnO) in per-alkaline minette from the Aosta valley, Italy. Apart from mafic-ultramafic settings and igneous rocks, detrital zinc-rich chromite has been described from sedimentary environments, such as in the radioactive quartz-pebble conglomerate of the Witwatersrand, South Africa (Eales and Reynolds, 1983; referred to in Weiser and Hirdes, 1997), and Tarkwa, Ghana (Weiser and Hirdes, 1997); in quartzite in the Ferreira-Ficalho Thrust Fault, South Portugal (Figueiras and Waerenborgh, 1997); in Ni-laterite, east Vermion, Edessa and Olympos, northern Greece (Economou-Eliopoulos, 2003); from the Guaniamo river diamondiferous placer, Venezuela (Johan and Ohnenstetter, 2010); from contact-metamorphosed sedimentary rocks, Nagasawa, SW Japan (Matsumoto *et al.*, 2017); and from fuchsitic metasedimentary rocks, Jack Hills, Western Australia (Staddon *et al.*, 2021). The zinc-rich chromite in Tarkwa, Ghana contains 19.42 wt.% ZnO, whereas that from Venezuela contains up to 30.82 wt.% and is described as ‘zincchromite’ (ZnCr_2O_4 , a phase first described by Nesterov and Rumyantseva, 1987).

From the above discussion it is apparent that reports on zinc-rich chromite in sedimentary environments are rare and none have been reported previously from India. In this study we describe detrital zinc-rich chromite, containing up to ~15 wt.% ZnO from radioactive quartz-pebble conglomerate overlying Archaean granitic basement from eastern India. We also discuss the possible source and origin of this unusually Zn-rich chromite in the conglomerate.

Geological background

The Singhbhum Craton (Fig. 1) in eastern India is a granite-greenstone terrain. The Palaeoarchean tonalite–trondhjemite–granodiorite–granite consists of the Older Metamorphic Tonalite Gneisses and the Singhbhum granitoid complex (3.45–3.29 Ga; Moorbath *et al.*, 1986a; Goswami *et al.*, 1995; Mishra *et al.*, 1999; Acharyya *et al.*, 2010; Tait *et al.*, 2011; Upadhyay *et al.*, 2019; Nelson *et al.*, 2014; Upadhyay *et al.*, 2014; Dey *et al.*, 2017; Olierook *et al.*, 2019; Pandey *et al.*, 2019). The Palaeoarchean–Mesoarchean greenstone belts contain the Iron Ore Group comprising low-grade volcano-sedimentary rocks (Saha, 1994; Mukhopadhyay *et al.*, 2008). The Older Metamorphic Group consists of high-grade supracrustals of ortho-amphibolites, para-amphibolites, pelitic schists and quartzites. Relatively smaller granitoid bodies, namely the Mayurbhang granite and Bonai granite occur, respectively, on the east and west of the main Singhbhum granite province. Palaeoproterozoic volcano-sedimentary basins such as the Dhanjori, Simlipal and Jagannathpur rim the Archaean craton (Fig. 1a).

The quartz-pebble conglomerates described in this contribution are from the Birtola and Bagiyabahal areas located on the western margin of the Bonai granite. The study area generally comprises a thick volcano-sedimentary sequence that overlies the basement granite and underlies scree deposits (Fig. 1b). Two phases of granite magmatism at 3369 ± 57 and 3163 ± 126 Ma (Sengupta *et al.*, 1991) comprise the Bonai pluton. The volcano-sedimentary sequence consists of multiple metasedimentary and metavolcanic

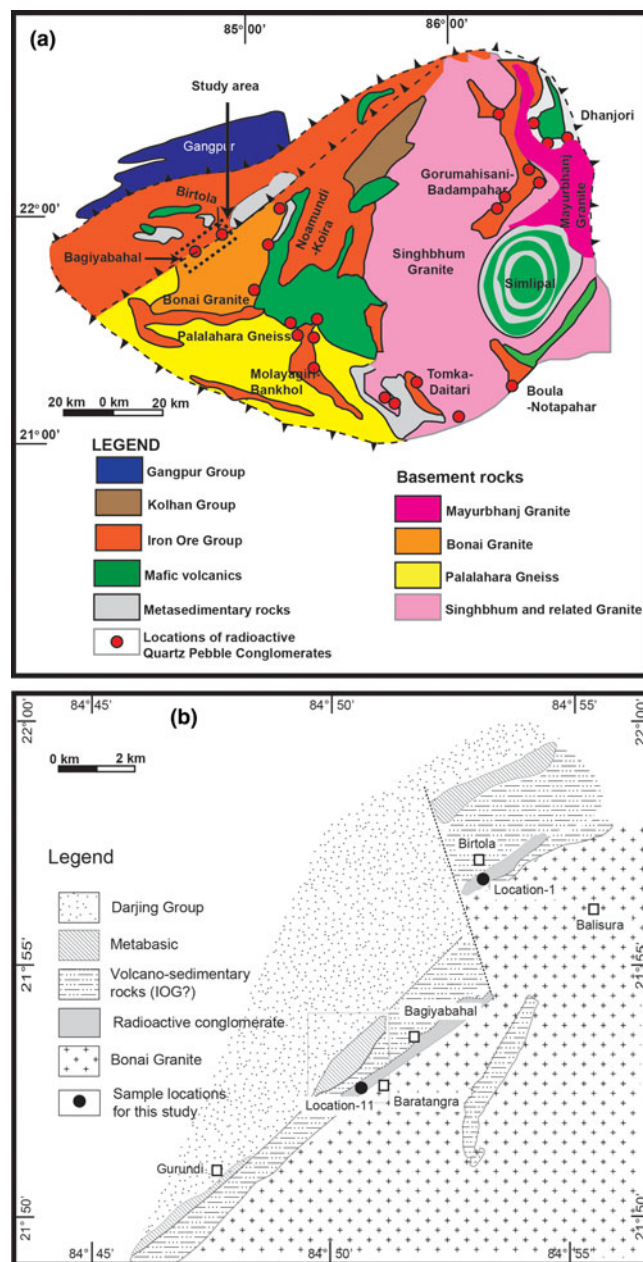


Fig. 1. (a) Geological map of the eastern Indian craton (after Saha, 1994). The locations of radioactive conglomerate are provided by the Atomic Minerals Directorate for Exploration and Research. The study area around Birtola and Bagiyabahal is located on the west of the Bonai granite (marked by a dashed rectangular box). (b) Geological map around the study area (modified after Kumar *et al.*, 2009) with the sample locations.

units. The metasedimentary rocks are represented by intercalated metaconglomerate and quartzite horizons. Sedimentary structure and facies analyses suggest that the conglomerates represent channel thalweg deposits overlying fluvial bounding surfaces. Some of the intercalated ultramafic volcanic rocks are komatiitic containing serpentinised forsteritic olivine. The komatiite contains disseminated chromite, pentlandite, Co-pentlandite and Ni-pyrrhotite. The chromite grains (>35.0 wt.% Cr_2O_3) have low values of Mg# (0.01 to 0.17; average: 0.04; $n = 50$) [$\text{Mg}/(\text{Mg} + \text{Fe}^{2+})$], moderate to high values of Cr# (0.51 to 0.88; average: 0.74; $n = 50$) [$(\text{Cr}/(\text{Cr} + \text{Al}))$], low Y_{Fe} (0.11 to 0.49; average: 0.22; $n = 50$) [$\text{Fe}^{3+}/(\text{Fe}^{3+} + \text{Al} + \text{Cr})$],

contain variable TiO₂ (0.23 to 5.99 wt.%; average: 2.41) and MnO (0.53 to 2.81 wt.%; average 1.642) and up to 1.25 wt.% ZnO (unpublished data) and are compositionally similar to chromite in komatiite metamorphosed at greenschist facies (Barnes and Roeder, 2001). In places, sulfide veinlets consisting of chalcopyrite–sphalerite–pyrrhotite–quartz ± biotite ± chlorite occur in the ultramafic rocks (Fig. 3a,b). Sphalerite grains occasionally contain blebs of chalcopyrite, a texture commonly known as chalcopyrite disease (Fig. 3b).

Analytical methods

Back-scattered electron (BSE) images of the samples and semi-quantitative analyses of the minerals were obtained using a JEOL JSM 6490 Scanning Electron Microscope (SEM) hosted at the Department of Geology and Geophysics, Indian Institute of Technology (IIT), Kharagpur. The major element compositions of chromite were determined at the same institute using a Cameca SX 100 electron microprobe equipped with four wavelength dispersive spectrometers with an acceleration voltage of 15 kV and beam current of 20 nA with 1 µm beam size. The dwell times were set at 10 s for the peak and 5 s on the background. The following reference materials and emission lines were used: CrKα (Cr₂O₃), AlKα (Al₂O₃), FeKα (hematite), ZnKα (sphalerite), TiKα (rutile), MgKα (MgO), MnKα (rhodinite) and SiKα (orthoclase).

Sample description, texture and geochemistry of zinc-rich chromite.

Samples were collected from conglomerates from two locations, Birtola (Location 1: 21°56′51.80″N 84°52′51.50″E) and from the Bagiyabahal area (Location 11: 21°52′59.00″N 84°50′54.80″E) (Fig. 1b). The mineralogy of the conglomerate and other rocks were investigated using standard polished thin sections with a petrographic optical microscope, scanning electron microscope and electron microprobe. The conglomerates are polymictic and matrix-supported (Fig. 2). The pebbles, up to ~1.5 cm in diameter, are represented predominantly by coarse-grained clasts derived from a quartz vein with minor clasts of fine-grained chert and quartzite. The clasts are stretched and fractured in places implying post-depositional deformation of the conglomerates. The matrix is composed of quartz, chlorite, sericite and the following heavy minerals: uraninite, zircon, chromite, ilmenite, rutile, apatite, monazite, xenotime, pyrite and zircon (Kumar *et al.*, 2009). Uraninite, zircon and chromite are well-rounded and detrital, similar to other Late Archaean to Palaeoproterozoic quartz-pebble conglomerate-hosted uranium mineralisation. Previous studies report up to 0.039 wt.% U₃O₈ (Kumar *et al.*, 2009) and 100 ppb to 2.25 ppm Au in the radioactive conglomerates (Jana *et al.*, 2016). Both well-rounded and sub-rounded to angular zircons are present; the latter exhibit evidence for fluid-induced alteration to quartz and xenotime. The rims of Fe-oxides and Fe-hydroxides can be enriched in Cu in places. Disseminated ilmenite grains are commonly brecciated and are enriched in Zn along the grain boundaries (Fig. 3c), containing up to 1.5 wt.% ZnO (SEM-energy-dispersive spectroscopy (EDS) data). Both sub-rounded, possibly detrital, and euhedral authigenic pyrite are also present. Sparse, angular sphalerite is occurs in association with authigenic pyrite (Fig. 3d). These textural features suggest that the conglomerate was subjected to fluid-induced modification.

Chromite, 5–6 grains within the scale of a thin section (including the pebbles and matrix), are present as disseminated grains in the matrix of the conglomerates. Most chromite grains are well-rounded with a diameter of up to ~200 µm (Fig. 3). Some grains are extensively fractured. Two different types of grains are present (Fig. 3). One type appears clean and has nearly homogeneous grey shades in BSE images (Fig. 3e and h). The other type has variation in grey shades in BSE images with darker sub-rounded disseminated spots or oriented darker lamellar inclusions (Fig. 3f and g). Some grains have patchy darker rims compared to the cores (Fig. 3h).

Compositions of zinc-rich chromite are given in Table 1. In general, chromite is poor in Mg with a very low Mg# (0 to 0.02; average 0.01; *n* = 26), the highest MgO content is 0.19 wt.%. The Cr₂O₃, Al₂O₃ and FeO content are strongly variable and range between 35.86 and 51.08 wt.%, 5.97 and 20.14 wt.% and 21.26 and 35.06 wt.% respectively. Overall, they have moderate values of Cr# (0.54 to 0.79; average: 0.68; *n* = 26) and low Y_{Fe} (0.03 to 0.16; average: 0.10). The MnO content of most chromite grains ranges between 0.18 to 0.56 wt.%; two spots had 2.25 and 2.55 wt.% MnO. Vanadium and Ti were detected in all grains and the V₂O₃ and TiO₂ contents range from 0.12 to 0.29 wt.% and 0.33 to 3.53 wt.% respectively. The actual compositions of the circular and lamellar spots (darker in BSE) in chromite could not be obtained due to their small size. However, higher TiO₂ contents were noted from these dark spots (in SEM-EDS and EPMA). One such dark lamellum analysed gave 70.03 wt.%, 6.03 wt.%, 13.97 wt.%, 4.91 wt.% and 3.34 wt.% of TiO₂, FeO, Cr₂O₃, Al₂O₃ and ZnO respectively. On the basis of this observation and data, sub-rounded and lamellar dark spots are considered to be rutile. All the chromite grains analysed are very rich in Zn, the ZnO concentrations range between 7.11 and 15.51 wt.%. In the partially-zoned (in BSE) zinc-rich chromite grain (Fig. 3h) the rim is enriched in Zn and Al and depleted in Fe and Cr; the core contains 7.10, 15.56, 35.43, 39.50 wt.% and the rim contains 15.51, 19.24, 25.25 and 38.01 wt.% of ZnO, Al₂O₃, FeO and Cr₂O₃ respectively. The core is slightly more enriched in Ti than the altered rim (0.47 wt.% and 0.26 wt.% TiO₂ at the core and rim respectively).

The formula of chromite was calculated on the basis of 32 oxygens. The Fe³⁺ and Fe²⁺ contents were calculated based on stoichiometry. In general, Zn-rich grains are also rich in Al, but depleted in Cr and Fe²⁺. This is demonstrated by a strong positive correlation between Al and Zn, strong negative correlation between Al and Cr, and between Zn and Fe⁺² (Fig. 4a–c), and moderate negative correlation between Al and Fe³⁺ (not shown). Zinc also has a moderate negative correlation with Mn, however it is not as strong as the correlation between Zn and Fe²⁺ (Fig. 4d). There is no discernible correlation between Mg and Zn (not shown). Titanium has a discernible negative correlation with Al and Zn (Fig. 4e,f). The compositions of zinc-rich chromite are plotted in ZnO vs. MnO and MgO vs. ZnO + MnO diagrams for comparison with Zn-rich chromites found in different host rocks/minerals (Arai and Ishimaru, 2011) (Fig. 4g,h). It is evident from this diagram that although the Zn-rich chromites analysed in this work are most similar to those found in altered/metamorphosed peridotites, although they are depleted in MgO and MnO compared to such chromite.

Discussion

The chromite is designated as zinc-rich chromite on the basis of >1.0 wt.% ZnO and >35.0 wt.% Cr₂O₃ (cf. Arai and Ishimaru,

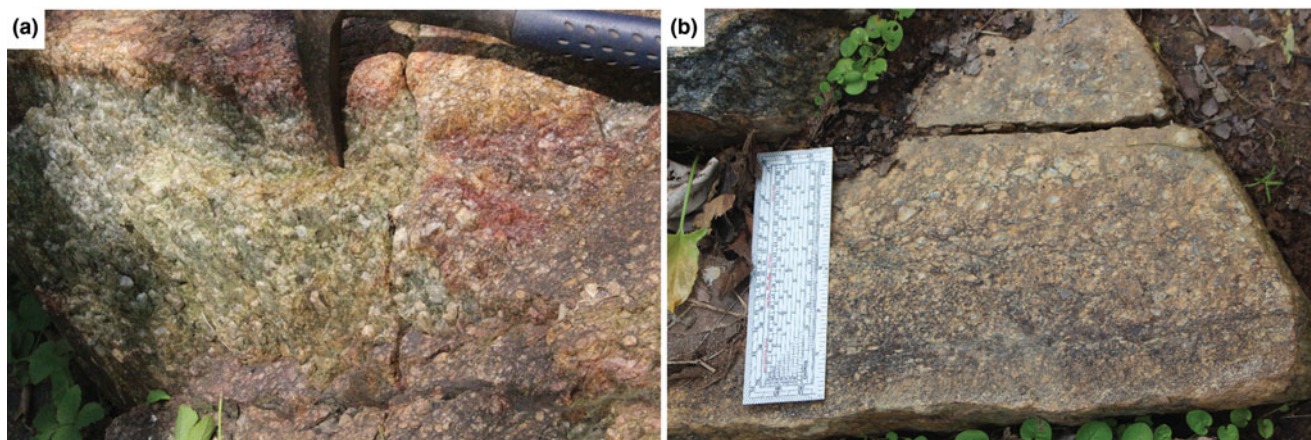


Fig. 2. Field photograph of radioactive conglomerates. Samples in (a) and (b) are from Location 1 and Location 11, respectively.

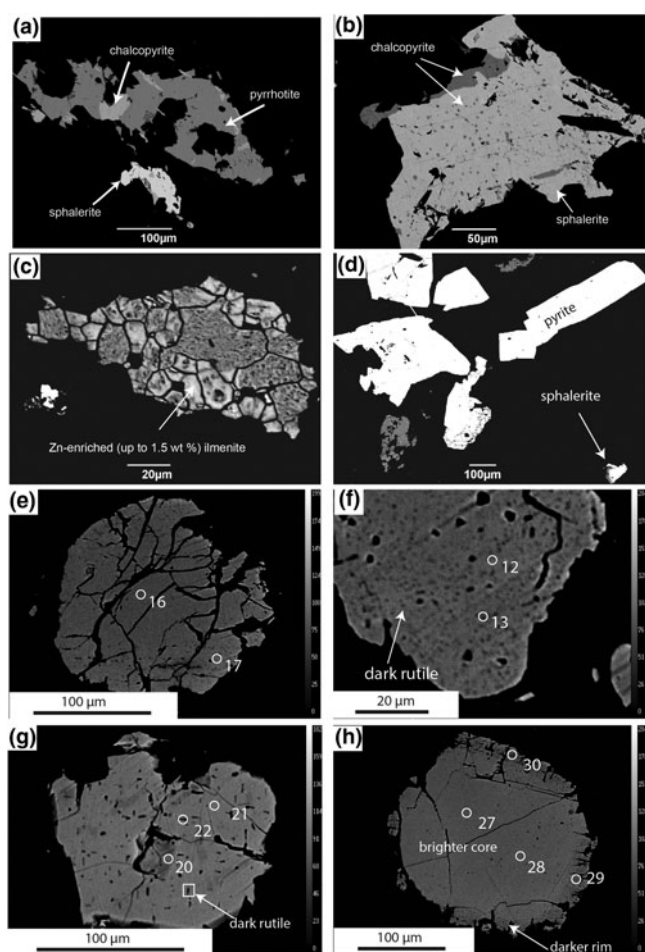


Fig. 3. Back-scattered electron images of different minerals from the sulfide vein in ultramafic rock (a and b) and conglomerate (c to h). (a, b) Sulfide minerals from the sphalerite-bearing sulfide veins seen in the mafic-ultramafic rocks; note sphalerite contains chalcopyrite (darker spots) blebs. (c) Ilmenite with localised Zn enrichment. (d) Sphalerite in association with well-developed angular grains of pyrite. (e–h) Well-rounded variably fractured zinc-rich chromite in conglomerate. Some grains are clean and nearly homogeneous (e and h) whereas some are heterogeneous (f and g) and contain spherical (f) or lamellar (g) rutile inclusions (exsolution?). The grain in (h) has a darker rim that is enriched in Zn and Al and depleted in Fe and Cr. The numerals on figures refer to spot numbers in Table 1. Chromite grains in (e) and (f) are from Location 1 whereas those in (g) and (h) are from Location 11. See text for discussion.

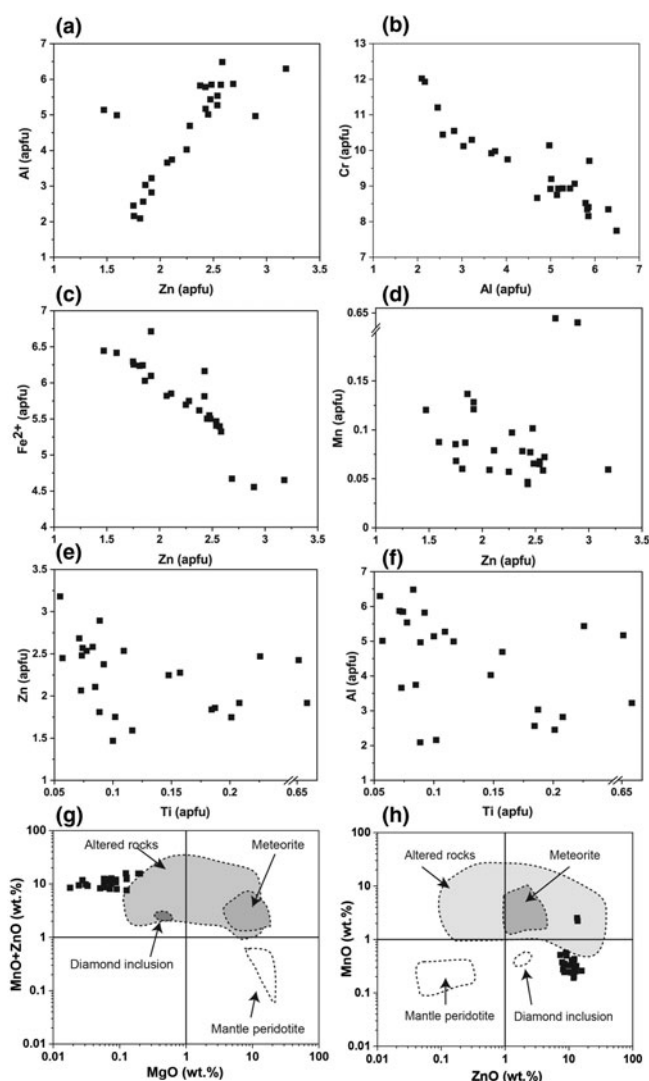


Fig. 4. Binary atoms per formula unit (apfu) and oxide distribution diagrams. Note the strong positive correlation between Zn and Al, negative correlation between Al and Cr, and between Zn and Fe²⁺. Zn also has a moderate negative correlation with Mn. Titanium has a negative correlation (excepting those with high Ti values; due to mixed analyses) with Zn (e) and Al (f). The chromites investigated are most similar to zinc-rich chromite in altered/metamorphosed peridotite although they contain much less MnO and MgO compared to such chromite. The fields drawn are based on data presented in Arai and Ishimaru (2011) for spinel containing >1.0 wt. % ZnO.

Table 1. Representative compositions (wt.%) of zinc-rich chromite.

Loc. No. Spot No.	1 1/1.	1 2/1.	1 3/1.	1 7/1.	1 8/1.	1 9/1.	1 10/1.	1 11/1.	1 12/1.	1 13/1.	1 14/1.	1 15/1.	1 16/1.
MgO	0.07	0.08	0.12	0.12	0.03	0.06	0.06	0.08	0.08	0.06	0.03	0.06	0.08
TiO ₂	0.35	0.44	0.40	0.52	3.14	0.37	0.73	1.79	0.27	1.06	0.86	0.96	0.68
Cr ₂ O ₃	36.93	38.21	35.86	40.08	39.84	40.80	38.28	38.42	40.95	40.02	44.17	46.23	42.72
FeO	30.89	31.08	29.93	29.54	28.14	28.07	33.43	28.60	29.83	28.29	35.06	34.24	31.41
MnO	0.25	0.34	0.31	0.27	0.19	0.29	0.40	0.20	0.32	0.43	0.56	0.53	0.23
Al ₂ O ₃	17.76	17.89	20.14	15.87	15.48	16.72	13.91	17.49	14.97	16.34	8.88	8.30	11.84
SiO ₂	0.02	0.10	0.21	0.04	0.04	0.03	0.06	0.05	0.05	0.06	0.02	0.01	0.11
V ₂ O ₃	0.15	0.17	0.13	0.25	0.27	0.26	0.24	0.22	0.26	0.27	0.27	0.23	0.20
ZnO	12.46	11.65	12.80	12.19	11.59	12.23	10.78	11.71	11.68	11.86	8.69	9.00	10.55
Total	99.23	100.04	100.16	98.88	98.87	99.02	98.11	98.65	98.53	98.55	98.67	99.60	97.95
Apfu													
Mg	0.03	0.03	0.05	0.05	0.01	0.02	0.03	0.03	0.03	0.03	0.01	0.02	0.04
Ti	0.07	0.09	0.08	0.11	0.67	0.08	0.16	0.38	0.06	0.23	0.19	0.21	0.15
Cr	8.16	8.35	7.75	8.93	8.93	9.07	8.67	8.52	9.20	8.94	10.12	10.55	9.75
Fe ²⁺	5.40	5.62	5.33	5.47	6.16	5.41	5.75	5.82	5.50	5.55	6.03	6.10	5.70
Fe ³⁺	1.82	1.56	1.51	1.50	0.51	1.19	2.26	0.90	1.58	1.13	2.47	2.17	1.89
Mn	0.06	0.08	0.07	0.07	0.04	0.07	0.10	0.05	0.08	0.10	0.14	0.13	0.06
Al	5.85	5.83	6.49	5.27	5.17	5.54	4.69	5.79	5.01	5.44	3.03	2.82	4.03
Si	0.03	0.03	0.06	0.01	0.01	0.01	0.02	0.01	0.01	0.02	0.01	0.01	0.03
V	0.03	0.04	0.03	0.06	0.06	0.06	0.06	0.05	0.06	0.06	0.06	0.05	0.05
Zn	2.57	2.38	2.58	2.54	2.43	2.54	2.28	2.42	2.45	2.47	1.86	1.92	2.25
Loc. No. Spot No.	1 17/1.	1 18/1.	1 19/1.	11 20/1.	11 21/1.	11 23/1.	11 24/1.	11 25/1.	11 26/1.	11 27/1.	11 28/1.	11 29/1.	11 30/1.
MgO	0.06	0.02	0.05	0.18	0.19	0.03	0.07	0.02	0.07	0.09	0.12	0.06	0.12
TiO ₂	0.82	3.53	0.90	0.40	0.34	0.33	0.39	0.40	0.45	0.55	0.47	0.35	0.26
Cr ₂ O ₃	44.23	44.72	47.77	43.87	44.26	43.11	43.26	51.08	49.95	40.02	39.50	37.74	38.01
FeO	35.13	31.04	32.84	21.26	21.27	33.24	32.19	31.34	31.25	34.69	35.43	29.27	25.26
MnO	0.34	0.49	0.34	2.25	2.55	0.24	0.32	0.24	0.27	0.37	0.51	0.28	0.25
Al ₂ O ₃	7.29	9.39	7.01	14.41	17.96	10.67	10.90	5.97	6.07	15.03	15.57	17.62	19.24
SiO ₂	0.05	0.09	0.12	0.09	0.01	0.03	0.12	0.22	0.10	0.07	0.05	0.33	0.08
V ₂ O ₃	0.26	0.21	0.24	0.14	0.13	0.18	0.21	0.12	0.08	0.20	0.20	0.26	0.22
ZnO	8.34	8.92	7.98	13.40	13.11	9.61	9.79	8.24	7.86	7.65	7.11	11.93	15.51
Total	96.69	98.55	97.34	96.11	99.99	97.59	97.35	97.72	96.28	98.95	99.33	97.95	99.14
Apfu													
Mg	0.02	0.01	0.02	0.08	0.08	0.01	0.03	0.01	0.03	0.04	0.05	0.02	0.05
Ti	0.18	0.77	0.20	0.09	0.07	0.07	0.08	0.09	0.10	0.12	0.10	0.07	0.05
Cr	10.45	10.30	11.21	10.15	9.71	9.92	9.98	12.02	11.93	8.92	8.75	8.41	8.35
Fe ²⁺	6.24	6.71	6.30	4.56	4.67	5.82	5.85	6.24	6.26	6.42	6.45	5.50	4.65
Fe ³⁺	2.53	0.85	1.85	0.64	0.26	2.27	2.00	1.57	1.64	1.76	1.86	1.40	1.21
Mn	0.09	0.12	0.09	0.56	0.60	0.06	0.08	0.06	0.07	0.09	0.12	0.07	0.06
Al	2.57	3.22	2.45	4.97	5.87	3.66	3.75	2.09	2.16	4.99	5.14	5.85	6.30
Si	0.01	0.03	0.04	0.03	0.00	0.01	0.04	0.07	0.03	0.02	0.01	0.09	0.02
V	0.06	0.05	0.06	0.03	0.03	0.04	0.05	0.03	0.02	0.05	0.04	0.06	0.05
Zn	1.84	1.92	1.75	2.89	2.69	2.07	2.11	1.81	1.75	1.59	1.47	2.48	3.18

2011). To our knowledge, this is the first reported occurrence of zinc-rich chromite having such high concentrations of ZnO from India and the third reported global occurrence of detrital zinc-rich chromite found in radioactive quartz-pebble conglomerates, known for elevated U and Au concentrations.

The strong negative correlation between Zn and Fe²⁺, Al and Cr, and Al and Fe³⁺, and strong positive correlation between Zn and Al, suggest that Zn substitutes predominantly for Fe²⁺ whereas Al substitutes for Cr and Fe³⁺ in the chromite structure. Albeit rare, on the basis of Zn and Al enrichment at the rim compared to the core of some grains, it is suggested that this substitution took place during an overprinting process resulting in partial transformation of (Fe)(Cr,Al,Fe³⁺)₂O₄ to ZnAl₂O₄, the latter still containing significant amounts of Fe and Cr due to incomplete substitution. A very low content of Mg and no discernible correlation between Mg and Zn suggest that the chromite was inherently low in Mg. It is suggested that the pristine chromite was Mg-poor and Fe-rich and contained some Fe³⁺. A negative

correlation between Mn and Zn indicates that Mn was inherent to pre-existing chromite prior to Zn enrichment. Depletion of Ti in the altered rim and discernible negative correlation of Ti with Al and Zn is suggestive of the primary nature of Ti in chromite and existence of Ti-rich chromite prior to Zn enrichment. Haggerty (1972) proposed that Ti enters into Cr-rich spinel by a coupled substitution, Ti⁴⁺+Fe²⁺ → 2Cr³⁺. In a study of exsolved rutile in chromite, Mitra and Samanta (1996) suggested that oxidation of chromite at high temperature and pressure (750°C and 5.7 kb) during the uplift of the host rock, results in exsolution of Ti (due to oxidation of Fe²⁺ to Fe³⁺ and consequent charge imbalance) from chromite and formation of rutile exsolution. Considering the high temperature and pressure of formation of rutile lamella in chromite, it is suggested that the exsolution lamella/blebs of rutile in some of the zinc-rich chromite investigated were produced at the provenance prior to their deposition in the conglomerates. Consequently, the presence of both chromite lacking inclusions and chromite with rutile exsolution

in the same conglomerate can be best explained by derivation of chromite from multiple sources.

The above discussion suggests that the original chromite (prior to Zn enrichment) was Mg-poor, Fe-rich and contained substantial amounts of Ti, Mn and Fe³⁺. One possible provenance of such chromite might be komatiite. Komatiites, containing accessory chromite, are known from Archaean greenstone sequences located on the western part of Singhbhum craton (Chaudhuri *et al.*, 2017; Mondal *et al.*, 2019; Bachhar *et al.*, 2021), wherein chromite in serpentinised komatiite is reported to be Mg-poor (1.24 wt.% MgO) and Fe-rich (32.56 wt.% FeO) (Bachhar *et al.*, 2021). However, to our knowledge, there is no published literature on the geochemistry of chromite from the study area located on the western part of the Singhbhum craton. During this investigation we have found accessory chromite in komatiite overlying the conglomerate investigated. Interestingly, this chromite is Mg-poor, Fe-rich and contains variable proportions of Fe³⁺, Ti and Mn (Section “Geological Background”) similar to the inferred composition of the pristine chromite prior to Zn enrichment. Modification of compositionally similar chromite via the substitution mechanism that we have proposed above [Zn → (Fe²⁺ + Mn); Al → (Cr + Fe³⁺)] can lead to the formation of zinc-rich chromite in the conglomerate. However, as this komatiite occurs above the conglomerate it cannot be the source of the chromite in the conglomerate. We hypothesise that similar, but older komatiites from the greenstone sequences (Iron Ore Group) in the western margin of the Singhbhum craton are the potential source. Note that a komatiitic source for detrital chromite has been proposed for the Witwatersrand conglomerate (Barnes and Roeder, 2001).

The origin of zinc-rich chromite is contentious. It has been suggested that crystallisation of Zn-bearing or zinc-rich chromite from, or by, interaction with sulfide melts related to sulfide mineralisation at the magmatic stage (Czamanske *et al.*, 1976; Groves *et al.*, 1977, 1983; Donaldson and Bromley, 1981; Liipo *et al.*, 1995; Weiser and Hirdes, 1997). However, others suggest that zinc-rich chromite forms by hydrothermal/metamorphic fluid-induced alteration of pre-existing chromite (Knorring *et al.*, 1986; Wylie *et al.*, 1987; Pan and Fleet, 1989, 1991; Béziat and Monchoux, 1991; Béziat *et al.*, 1993; Jurković and Jakšić, 1994; Challis *et al.*, 1995; Gahlan and Arai, 2007; Johan and Ohnenstetter, 2010; Taguchi *et al.*, 2012; Singh and Singh, 2013; Schulze *et al.*, 2014; Fanlo *et al.*, 2015). For detrital zinc-rich chromite, Zn enrichment has been proposed to have taken place at the provenance site prior to deposition in the sedimentary rock by Weiser and Hirdes (1997) and Johan and Ohnenstetter (2010) for conglomerate at Tarkwa, Ghana and the Guaniamo river and diamondiferous placer, Venezuela. In contrast, some other studies (Economou-Eliopoulos, 2003; Matsumoto *et al.*, 2017; Staddon *et al.*, 2021) suggested Zn enrichment of chromite after deposition in the sedimentary rock such as those in Ni-laterite, northern Greece, metasedimentary rocks, Nagasawa, SW Japan and fuchsite-bearing metasedimentary rocks in Jack Hills, Western Australia. Proposed fluids for the post-depositional modification of chromite are diverse. Economou-Eliopoulos (2003) envisaged that Zn enrichment was caused by diagenetic fluid. Modification by metamorphic fluid and hydrothermal fluid triggered by contact metamorphism have been proposed, respectively, for Jack Hill and Nagasawa (Matsumoto *et al.*, 2017; Staddon *et al.*, 2021). Whether the Zn enrichment of the chromite investigated here took place at the provenance site or at the depositional site remains unresolved. So far, there are no reported occurrences of zinc-rich chromite (similar in

composition to those in the studied conglomerate), in any igneous rock from the study area or anywhere from the eastern Indian shield. This restricts us from proposing or discarding possible pre-sedimentation Zn enrichment at the provenance site. However, fluid-induced modification of conglomerate, the presence of authigenic sphalerite and localised Zn enrichment of fractured ilmenite in the conglomerate, and the presence of sphalerite–chalcopyrite–pyrrhotite-bearing sulfide veins in rocks stratigraphically overlying conglomerates are some of the evidence for interaction of fluid with a conglomerate matrix causing fluid-assisted Zn introduction/re-distribution in the conglomerate and associated rocks postdating the deposition of chromite in the conglomerate. Therefore, post-sedimentation modification of chromite to zinc-rich chromite by metamorphic or hydrothermal fluids at the depositional site cannot be ruled out either. Further study is warranted to confirm the provenance, mechanism and timing of Zn enrichment of chromite.

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