

# CO<sub>2</sub>-C losses and carbon quality of selected Maritime Antarctic soils

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**Abstract:** Polar Regions are the most important soil carbon reservoirs on Earth. Monitoring soil carbon storage in a changing global climate context may indicate possible effects of climate change on terrestrial environments. In this regard, we need to understand the dynamics of soil organic matter in relation to its chemical characteristics. We evaluated the influence of chemical characteristics of humic substances on the process of soil organic matter mineralization in selected Maritime Antarctic soils. A laboratory assay was carried out with soils from five locations from King George Island. We determined the contents of total organic carbon, oxidizable carbon fractions of soil organic matter, and humic substances. Two *in situ* field experiments were carried out during two summers, in order to evaluate the CO<sub>2</sub>-C emissions in relation to soil temperature variations. The overall low amounts of soil organic matter in Maritime Antarctic soils have a low humification degree and reduced microbial activity. CO<sub>2</sub>-C emissions showed significant exponential relationship with temperature, suggesting a sharp increase in CO<sub>2</sub>-C emissions with a warming scenario, and Q10 values (the percentage increase in emission for a 10°C increase in soil temperature) were higher than values reported from elsewhere. The sensitivity of the CO<sub>2</sub>-C emission in relation to temperature was significantly correlated with the humification degree of soil organic matter and microbial activity for Antarctic soils.

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**Key words:** CO<sub>2</sub> emission, C sequestration, C stocks, permafrost, polar soils

## Introduction

Soils constitute the major carbon reservoir of the terrestrial system (Silva & Mendonça 2007), hence the study of carbon dynamics at different soil types, even in natural environments and in areas under anthropogenic influence, has been an important issue due to its relation to the greenhouse gas effect (Batjes 1996). In contrast with other cold regions (Arctic and alpine environment), the Antarctic continent has been comparatively less studied, although it represents an opportunity for exploring phenomena. Terrestrial Antarctic ecosystems are restricted in the ice-free zone, distributed basically along the coast or isolated mountains chain, representing less than 0.5% of the total area of that continent (Campbell & Claridge 1987).

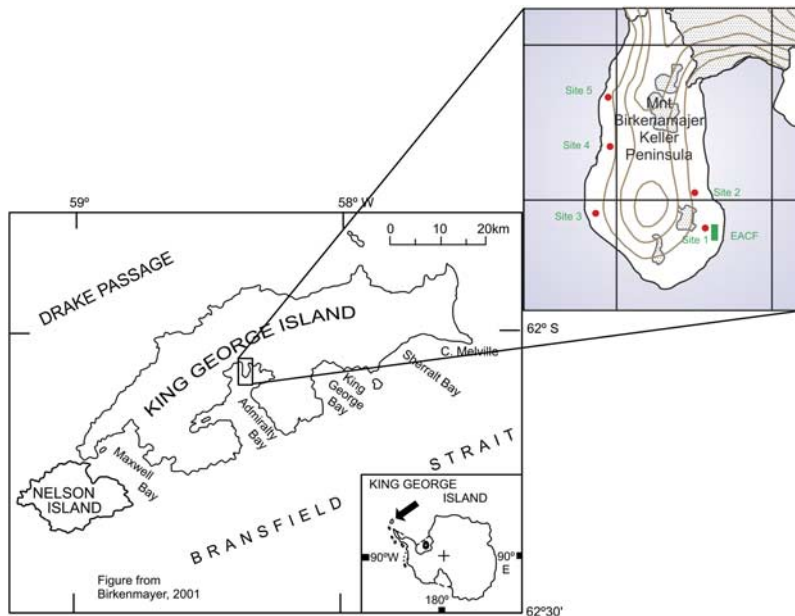
The severe climate and low water availability are important factors driving the Antarctic soils formation (Campbell & Claridge 1987), and soil losses by periglacial erosion influences the amount of organic matter in Antarctic soils, which are highly variable (Michel *et al.* 2006, Carvalho *et al.* 2010).

Soil C stock is a result of the combination between the primary production by autotrophic organisms and the organic

matter decay promoted microbial activity (Silva & Mendonça 2007). For this reason, the monitoring of soil C stocks with emphasis on soil temperature may indicate future changes in stocks due to changes in the terrestrial environment. The Maritime Antarctic has experienced significant temperature increases in the last 50 years (Vaughan *et al.* 2001, Quayle *et al.* 2002, Steig *et al.* 2009). Higher temperatures may increase the soil organic matter (SOM) decomposition due to the great exposure of formerly frozen soil (La Scala *et al.* 2010, Mendonça *et al.* 2011), resulting in decreasing soil C stocks (Carvalho *et al.* 2010, La Scala *et al.* 2010).

The organic matter mineralization process affects primarily the less persistent SOM forms, resulting in a gradual accumulation of organic materials with higher recalcitrance in substances such as chitin, uric acid and humic substances (Myrcha *et al.* 1983). This process is part of the C cycle and promotes, simultaneously, enhanced CO<sub>2</sub> in the atmosphere and reduction of soil C stocks. The microbial activity in Antarctic soils, like elsewhere, is dependent on multiple factors, such as nutrient availability, pH, moisture, salinity, besides temperature.

The prevailing lower soil temperatures in Antarctica reduce SOM mineralization. Cyanobacteria, lichens, algae,



**Fig. 1.** Map of King George Island, highlighting Keller Peninsula, where soils were collected (adapted from Birkenmajer 2002).

bryophytes and a few higher plants present on the soil surface are capable of fixing atmospheric C in Antarctic soils (Simas *et al.* 2006). However, the majority of global warming models predict an increase in temperature especially in periglacial regions, which will greatly affect CO<sub>2</sub>-C production and SOM content (La Scala *et al.* 2010). In addition, increasing soil temperature results in higher soil CO<sub>2</sub>-C emissions according to different vegetation covers (Mendonça *et al.* 2011). There is a consensus that the characterization of C dynamics in Antarctic soils is fundamental to establish the relation between soil properties and climatic changes (Beyer *et al.* 2004).

To examine the consequences of increasing soil temperature in Antarctica the understanding of mineralization rate of SOM, including an enhanced knowledge of C quality and soil microbial activity, is necessary. The aim of this study was to analyse the influence of these characteristics on the

mineralization process of SOM in selected, representative Maritime Antarctic soils.

## Material and methods

### Characterization of the studied sites

Admiralty Bay is located at King George Island (62°03'–62°05'S, 58°23'–58°24'W) which is the largest island in the South Shetland Islands (Simas *et al.* 2007). During the 2008–09 summer, soils were collected in three replicates at 0–10 cm and 10–20 cm in five different soils locations across Keller Peninsula, as shown in Fig. 1.

Based on the physical-chemical, mineralogical and morphological soil characteristics (Table I), soil located in sites 1 and 2 were classified as basaltic/andesitic soils while those in sites 3, 4 and 5 were classified as acid sulphate soils (Simas *et al.* 2006). Prior to analytical procedures, all

**Table I.** Co-ordinates, physical and mineralogical characteristics of the soils collected in the Keller Peninsula.

Sample	Co-ordinates	Stone (%)	Soil density (g cm <sup>-3</sup> )	Mineralogy	Texture
P1	21 E 0427137 UTM 3115552	53.9	1.60		
P2	21 E 0427091 UTM 3116260	46.1	1.36	HIS-sme > pyr, pl > all	Sandy silt
P3	21 E 0425658 UTM 3115916	63.9	1.43		
P4	21 E 0425922 UTM 3117161	73.1	1.24	ka, ch, I-S > ja > feh	Sandy clay
P5	21 E 0426010 UTM 3117588	53.4	1.31		

HIS-sme = interstratified smectite-hydroxy-interlayered-smectite, pyr = pyroxene, pl = plagioclase, all = allophane, ka = kaolinite, ch = chlorite, I-S = interstratified illite-smectite, ja = jarosite, feh = ferrihydrite.

**Table II.** Characterization of the soil organic matter of the soil samples.

Samples	TOC g kg <sup>-1</sup>	TN g kg <sup>-1</sup>	C stock kg m <sup>-2</sup>	C/N	C <sub>labile</sub> %	C <sub>recalc</sub> %	qMIC %	C <sub>fulv</sub> g kg <sup>-1</sup>	C <sub>hum</sub> g kg <sup>-1</sup>
P1 0–10 cm	4.39	0.12	0.334	36.36	36.37	63.77	1.91	0.14	0.00
P1 10–20 cm	3.68	0.08	0.236	47.65	46.66	53.39	1.42	0.09	0.00
P2 0–10 cm	7.57	0.32	0.507	23.45	48.19	51.78	1.52	0.90	0.00
P2 10–20 cm	5.96	0.16	0.243	38.06	38.39	61.64	1.07	0.22	0.00
P3 0–10 cm	12.04	0.83	0.961	14.54	69.05	30.87	0.27	2.10	1.47
P3 10–20 cm	5.12	0.41	0.407	12.59	49.35	50.70	1.07	0.46	0.25
P4 0–10 cm	9.85	0.50	0.787	19.62	68.55	31.41	0.56	1.58	0.74
P5 0–10 cm	13.51	0.76	0.873	17.75	50.2	49.79	0.25	1.91	2.80
P5 10–20 cm	9.46	0.66	0.676	14.23	64.41	35.64	0.75	1.83	0.36

TOC = total organic carbon, TN = total nitrogen, C<sub>labile</sub> = C in the labile form, C<sub>recalc</sub> = C in the recalcitrant form, qMIC = percentage of the total C in the microbial biomass, C<sub>fulv</sub> = C in the fulvic acid fraction, C<sub>hum</sub> = C in the humic acid fraction.

samples were passed through a 2 mm sieve to quantify the percentage of coarse fragments, larger than 2 mm.

Total soil organic carbon (TOC) content in all studied locations was determined according to Yeomans & Bremner (1988). Oxidizable organic C labile and recalcitrant fractions (C<sub>labile</sub> and C<sub>recalc</sub>) were determined according to Chan *et al.* (2001), while total nitrogen was determined by the Kjeldahl method (Bremner & Mulvaney 1982). Microbial biomass (C<sub>mic</sub>) was determined by the irradiation extraction method (Islam & Weil 1998, Ferreira *et al.* 1999). The C<sub>mic</sub>/TOC was used as an indicator of soil microbial biomass pool (Marchiori & Mello 1999), and considered a proxy of SOM quality.

Soil humic and fulvic acids fractions were extracted, fractionated and purified according to the International Humic Substances Society (IHSS) (Swift 1996), and following, total C content was determined according the method described in Yeomans & Bremner (1988). Soil bulk density was determined in soil samples with the use of paraffin impregnation method (Embrapa 1997). C stock of surface horizons of soil was calculated using the following formula:

$$Cst = (TOC)(E.C.)(D)[1 - (\% \text{fragments} > 2 \text{ mm})/100],$$

where Cst = C stock (kg m<sup>-2</sup>), TOC = total soil organic carbon (g C kg<sup>-1</sup> soil), E.C. = thickness of the soil (m),

and D = soil density (kg soil m<sup>-3</sup>), expressed in g cm<sup>-3</sup> or 10<sup>-3</sup> kg m<sup>-3</sup>.

#### In situ soil CO<sub>2</sub>-C emission experiment

The field experiment was conducted in the vicinity of Comandante Ferraz Brazilian Antarctic Station, Keller Peninsula. Soils from locations 1–5 (0–10 cm), with and without vegetation (mixed *Deschampsia* and mosses), were kept at open environmental conditions. Measurement of CO<sub>2</sub>-C emission were conducted during two summer seasons, from 24 January to 10 February 2008 and from December 2008 to March 2009, totalling 21 days of readings during these periods, in accordance with weather conditions.

Samples were collected from five sites and were kept in open-air on a 60 x 60 cm wooden board, composed by 21 PVC collars (10 cm diameter each), containing 300 g of bare soil or 200 g of soil + 100 g of natural vegetation (mixed *Deschampsia* and mosses). An automated soil CO<sub>2</sub> flux system (LI-8100, Li-Cor Environmental) was coupled to the collars to measure CO<sub>2</sub> emitted at each sample. At measurement mode, three replicates were applied for each collar, performing 63 measurements per day under contrasting soil temperature conditions. The LI-8100

**Table III.** Descriptive statistics of CO<sub>2</sub>-C emission from soil with or without vegetation (S+V or S, respectively).

Samples	<i>n</i>	Mean (g m <sup>-2</sup> h <sup>-1</sup> )	Mean (g (g soil C) <sup>-1</sup> h <sup>-1</sup> )	Standard deviation	C.V.	Min.	Max.
P1 S	28	0.096	0.07289	0.058	0.607	0.000	0.250
P1 S+V	28	0.281	-	0.125	0.446	0.100	0.610
P2 S	35	0.138	0.06077	0.111	0.802	0.030	0.590
P2 S+V	35	0.428	-	0.193	0.450	0.157	1.000
P3 S	35	0.218	0.06035	0.117	0.534	0.080	0.667
P3 S+V	35	0.585	-	0.217	0.371	0.203	1.140
P4 S	35	0.114	0.03858	0.106	0.933	0.000	0.600
P4 S+V	35	0.372	-	0.249	0.668	0.060	1.150
P5 S	28	0.155	0.01894	0.055	0.353	0.060	0.253
P5 S+V	35	0.709	-	0.523	0.738	0.207	2.860
Mean S		0.144	0.03553	0.093	-	0.034	0.472
Mean S+V		0.475		0.295	-	0.145	1.352

C.V. = coefficient of variation.

**Table IV.** Parameters of the model between CO<sub>2</sub>-C emissions and soil temperature for the studied soil and Q10 factor.

Location	ln(FCO <sub>2</sub> ) = A + B*Tsoil				Q10
	A	B (°C <sup>-1</sup> )	R	P	
P1 S	-3.154 ± 0.221	0.189 <sub>a</sub> ± 0.055	0.566	2.1 × 10 <sup>-2</sup>	6.593 <sub>a</sub> ± 4.558
P1 S+V	-1.878 ± 0.140	0.157 <sub>a</sub> ± 0.038	0.632	3.0 × 10 <sup>-4</sup>	4.816 <sub>a</sub> ± 2.027
P2 S	-2.818 ± 0.190	0.166 <sub>a</sub> ± 0.044	0.551	6.0 × 10 <sup>-4</sup>	5.238 <sub>a</sub> ± 2.636
P2 S+V	-1.595 ± 0.107	0.167 <sub>a</sub> ± 0.025	0.765	8.9 × 10 <sup>-8</sup>	5.312 <sub>a</sub> ± 1.362
P3 S	-2.050 ± 0.113	0.101 <sub>b</sub> ± 0.023	0.609	1.0 × 10 <sup>-4</sup>	2.732 <sub>a</sub> ± 0.648
P3 S+V	-1.097 ± 0.105	0.124 <sub>a,b</sub> ± 0.023	0.679	7.5 × 10 <sup>-6</sup>	3.463 <sub>a</sub> ± 0.844
P4 S	-2.914 ± 0.221	0.112 <sub>b</sub> ± 0.041	0.432	1.1 × 10 <sup>-2</sup>	3.059 <sub>a</sub> ± 1.432
P4 S+V	-1.557 ± 0.260	0.085 <sub>b,c</sub> ± 0.058	0.248	1.5 × 10 <sup>-1</sup>	2.347 <sub>a</sub> ± 1.751
P5 S	-2.240 ± 0.164	0.069 <sub>c</sub> ± 0.034	0.372	5.1 × 10 <sup>-2</sup>	1.990 <sub>a</sub> ± 0.728
P5 S+V	-0.903 ± 0.194	0.086 <sub>b</sub> ± 0.039	0.360	3.4 × 10 <sup>-2</sup>	2.368 <sub>a</sub> ± 1.032
	Mean	0.126		Mean	3.792
	s.e.	0.042		s.e.	1.579

\*Means followed by same letter do not differ significantly by student's *t*-test at the 5% level of significance.

A and B = linear and angular coefficients, obtained from the linear regression analysis, respectively; R = linear correlation coefficient; P = significance level; and s.e. = standard error.

system is based on the infrared absorption spectroscopy that analyse the time changes of CO<sub>2</sub> concentration inside the chamber once it is placed onto the soil PVC collars. As the internal chamber is a closed system (internal volume = 854 cm<sup>3</sup>), with a fixed contact area to the soil (exposed area = 83 cm<sup>2</sup>), changes in CO<sub>2</sub> concentration inside the chamber once it was placed on the collars, was used to calculate emissions at measurement mode, during 1.5 min reading. As soon as emissions were measured, *in situ* soil temperature was also recorded in all treatments with a field thermometer placed at 10 cm depth.

The relation between CO<sub>2</sub> and soil temperature was described by the equation  $FCO_2 = F_0 \times \exp(b \times T_{soil})$ , with the natural log (Ln) of the CO<sub>2</sub> emission we have  $\ln(FCO_2) = \ln(F_0 \times \exp(b \times T_{soil}))$ , the result is  $\ln(FCO_2) = \ln(F_0) + b \times T_{soil}$ . A linear relation between  $\ln(FCO_2)$  and the *Tsoil* is expected in the environments where soil temperature is a limiting factor.

Based on the B coefficients it is possible to derive the Q10 factor, which represents the percentage increase in emission for a 10°C increase in soil temperature. This is derived as  $Q10 = e^{10 \times B}$ .

### Statistical analysis

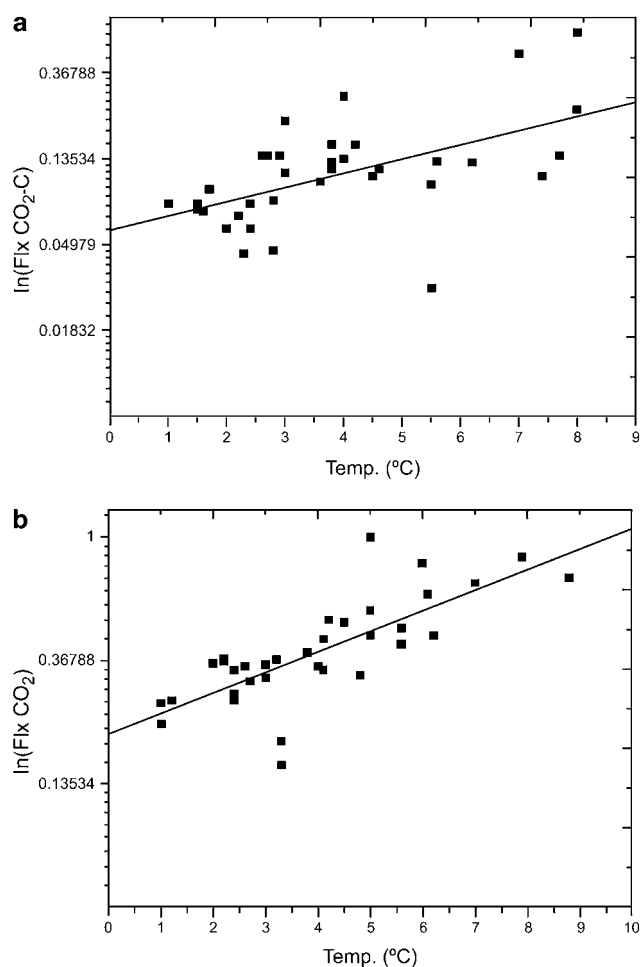
Data was submitted to the students *t*-test (significance level of 5%) in order to compare the mean values by using the SAEG software (Funarbe 2007). Graphs and linear regressions were performed using the Origin 8.0 software (Origin Lab Corp).

## Results

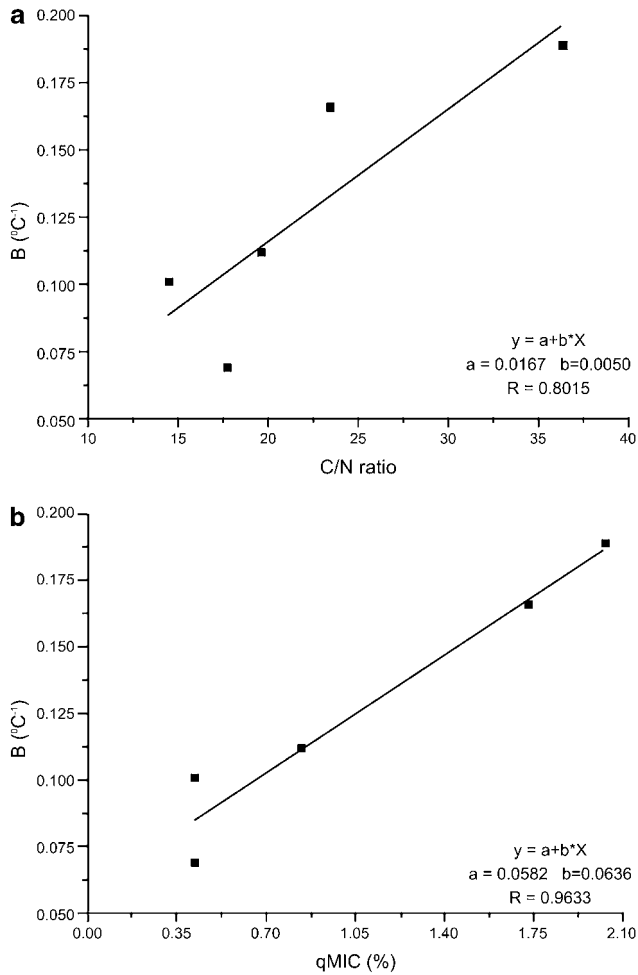
### Soil organic matter characterization

Table II presents the results of SOM characterization in all locations studied for both depths, except for location 4, where the bedrock was found at 10 cm depth and no soil could be collected. The lowest C stocks were found in

samples 1 and 2 (basaltic/andesitic soils), both very rocky and shallow. The microbial coefficient (qMIC) ratio indicates higher values for soils 1 and 2, contrasting to



**Fig. 2.** Logarithm of CO<sub>2</sub>-C emission as related to soil temperature for site 2. **a.** Bare soil, and **b.** soil with vegetation.

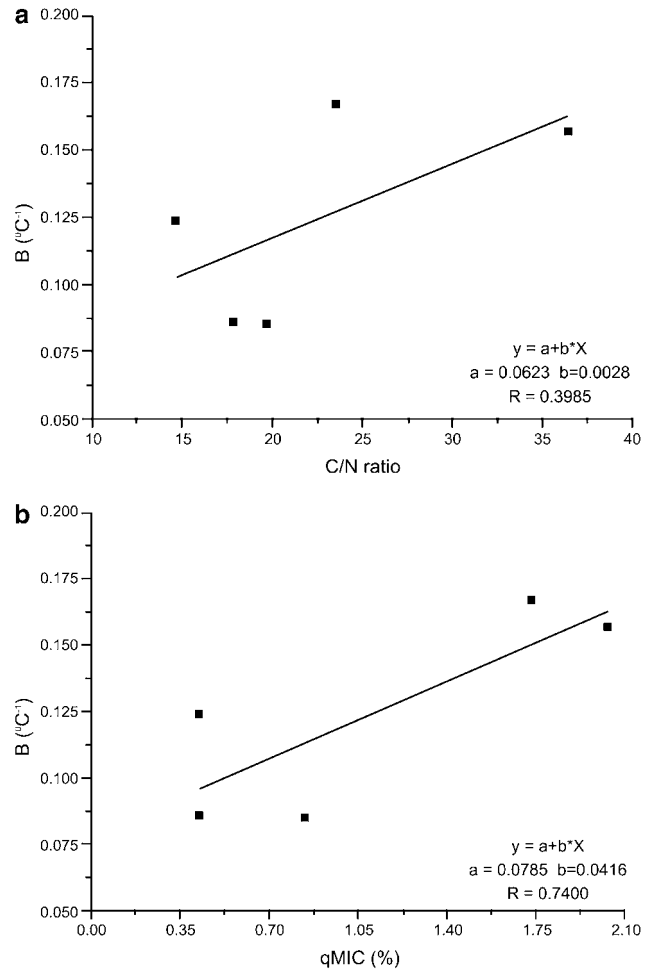


**Fig. 3.** Relationship between the B factor of CO<sub>2</sub>-C emissions for tests with bare soil (S) versus **a.** the values of C/N ratio, and **b.** microbial coefficient (qMIC) for Antarctic soils.

different proportion of C<sub>labile</sub> and C<sub>recalc</sub> fractions. Differences associated with biological and chemical methods can represent different C fractions.

#### Soil CO<sub>2</sub>-C emission

Table III presents the mean CO<sub>2</sub>-C values from the two *in situ* summer experiments. The mean CO<sub>2</sub>-C emissions varied from 0.034–0.472 g of CO<sub>2</sub>-C m<sup>-2</sup> h<sup>-1</sup> for soil samples (S) and from 0.145–1.352 g of CO<sub>2</sub>-C m<sup>-2</sup> h<sup>-1</sup> for soil plus vegetation samples (S+V). Changes in CO<sub>2</sub>-C can be related to soil temperature, as shown in Table IV. Significant ( $P < 0.05$ ) exponential relationship between CO<sub>2</sub>-C and soil temperature was observed for all locations, in samples either with or without vegetation. The relation  $F_{CO_2-C} = F_0 e^{B \cdot T_{soil}}$ , was linearly fit by applying  $\ln(F_{CO_2-C}) = \ln(F_0) + B(T_{soil})$  (Fig. 2). Results indicate that CO<sub>2</sub>-C emission sensitivity to soil temperature (B coefficient in Table IV) is similar in all soils studied, irrespective of location and presence of vegetation. This is corroborated by the observed B values  $\pm$  standard errors.



**Fig. 4.** Relationship between the B factor of CO<sub>2</sub>-C emissions for tests with bare soil with vegetation (S+V) versus **a.** the values of C/N ratio, and **b.** microbial coefficient (qMIC) for Antarctic soils.

The significant and positive linear correlation of the B factor with C/N ratio and qMIC is illustrated in Figs 3 & 4. These results show how sensitive soil CO<sub>2</sub>-C emissions are to increasing soil temperature.

**Table V.** Q10 values for soils from different regions, estimated by several authors.

Reference	Q10	Material/location
Fang <i>et al.</i> 1998	2.5	soils, Florida
Hanson <i>et al.</i> 2003	2.5	forest soils, USA
Boone <i>et al.</i> 1998	2.5–3.9	forest soils
Epron <i>et al.</i> 1999	2.3–3.9	forest soils
Davidson <i>et al.</i> 2006	2.8–5.0	forest soils, USA
Larionova <i>et al.</i> 2007	0.9–3.4	soils Moscow, Russia
Reichstein <i>et al.</i> 2005	2.4–3.2	soils
Lützwow & Kögel-Knabner 2009	4.0–6.0	soils from different regions
Gershenson <i>et al.</i> 2009	1.6–2.4	soils, California
McCulley <i>et al.</i> 2007	1.4–3.3	soils, Texas
Raich & Schlesinger 1992	1.3–3.3	soils

The Q10 values are reported in Table IV. The mean Q10 value in our study ( $3.792 \pm 1.579$ ) is higher than that observed elsewhere by several authors (Table V).

## Discussion

### *Soil organic matter characterization*

The mean C content for all soil samples ( $7.95 \pm 3.47 \text{ g kg}^{-1}$ ) is within the range of previous studies (Bölter 1995, Bölter *et al.* 1997, Simas *et al.* 2008, Francelino *et al.* 2011), but lower than the ones found in Carvalho *et al.* (2010) for mineral soils of King George Island, and other terrestrial studies elsewhere in Antarctica (Campbell & Claridge 1987, Simas *et al.* 2008). However, soil C stocks presented values similar to those reported by Michel *et al.* (2006), with a range  $0.1\text{--}0.9 \text{ kg m}^{-2}$  (in the upper 10 cm layer) for Admiralty Bay soils.

The high C/N ratio of the soil is usually related to low decomposition rate of soil organic matter (Silva & Mendonça 2007). Higher C/N ratio was observed in samples 1 and 2 for both depths, indicating a low decomposition degree of SOM. On the other hand, samples 3, 4 and 5, corresponding to acid sulphate soils, presented intermediate C/N ratios, varying from 12.6–19.6. Considering that all samples are subjected to cold climate conditions, and with a poor vegetation cover (basically *Deschampsia antarctica* Desv. and mosses), the differences in C/N ratio may be attributed to varying soil moisture conditions. Samples 1 and 2 are loamy sand soils contrasting to sandy clay soils 3, 4 and 5. It is well-documented that higher clay content effectively protects soil C due to colloidal interactions (Silva & Mendonça 2007). Under experimental conditions it is expected only limited colloidal interactions between clay and SOM occur since there is little soil structural development. Hence, only resistant structural SOM with high C/N ratio, and recalcitrant fractions, remain.

Soils 1 and 2 have high pH, nutrient contents and cation exchange capacity (Simas *et al.* 2006), which may positively affect soil microbial activity. In these areas erosion is intense, and soils are less developed and have lower SOM content, compared with soils 3, 4 and 5. Our data are supported by the results of Hopkins *et al.* (2009), which indicated that with permafrost melting, the SOM from Maritime Antarctica can have a relatively fast turnover, which may be related to recent exposure of labile material, protected by the frozen state.

The low or negligible contents of humic and fulvic acid fractions in the soils indicate that inherited humin is the main route for humic substances formation. Low temperature and little clay content of Antarctic soils do not favour the physical protection of SOM, increasing C losses (Silva & Mendonça 2007).

### *Soil CO<sub>2</sub>-C emission*

The presence of vegetation, regardless of the soil studied, resulted in 230% emission increase, or the equivalent of  $0.331 \text{ g of CO}_2\text{-C m}^{-2} \text{ h}^{-1}$ . In any bare soil condition, the

ratio between total emission (or mean emission) and the soil C stock, would be related to the rate of the soil C decay (time<sup>-1</sup>). In this regard, our results indicate that the higher this ratio, the lower the soil C stock. This means that the soil C losses through CO<sub>2</sub> are the main mechanism which controls the soil C stock in Maritime Antarctic soils.

The difference between minimum and maximum mean values for bare soil emissions were as high as 1288%, while in S+V samples differences were 832%. The lowest emissions, with negligible values, were observed in soil 1 (basaltic/andesitic soils, sandy soil), without vegetation, which also had the lowest soil C stocks (Table II). On the other hand higher mean emissions during the two summer periods were obtained for soil 5 with vegetation ( $2.860 \text{ g C-CO}_2 \text{ m}^{-2} \text{ h}^{-1}$ , acid sulphate clay soil), which showed higher soil C stocks. The mean emissions suggest that soil C losses through CO<sub>2</sub> were greater in soil 5, under *D. antarctica* plus moss cover. On the other hand higher emissions from bare soils were observed in soil 3, and similar figures were obtained for soils 2 and 4. These results suggest that soil exposure following permafrost melting may enhance the decay of native SOM to CO<sub>2</sub>, as already suggested by previous studies (Michaelson *et al.* 2004, Carvalho *et al.* 2010, La Scala *et al.* 2010, Mendonça *et al.* 2011). Overall, the maximum, minimum and the mean values of C emission from soil plus vegetation were greater than the bare soil emissions, which can be attributed to plant root respiration (Tang & Baldocchi 2005), being influenced by different plant photosynthetic activity (Kuzyakov & Gavrichkova 2010).

The CO<sub>2</sub>-C emission is sensitive to soil temperature and the significance of B values shows three different sensitivity ranges, a higher range for soils 1 and 2, a mean for soils 3 and 4, and a lower range for soil 5, with  $0.069 \text{ C}^{-1}$ . Soils with higher B factor are those with lower SOM contents and C stocks (soils 1 and 2). These are also loamy sand soils, with lower clay content than the remaining soils 3, 4 and 5. The former soils have higher SOM humification degree, similar SOM content, C/N ratio and soil C stock, whereas soil 5 presented lower labile C content compared to soils 3 and 4. This is related to the lower B factor of soil 5 in relation to the others, since the labile C pool, readily oxidizable, is less at soil 5.

It is known that the sensitivity of soil C to soil temperature is affected by numerous factors that are directly or indirectly related to temperature (Yuste *et al.* 2007). These relations are better illustrated in Fig. 3 as the B factor, which expresses how sensitive soil CO<sub>2</sub>-C emission is to increasing soil temperature. We observed a significant and positive linear correlation with C/N ratio and qMIC. These data contrast with the results of Hopkins *et al.* (2009) that showed positive relation of high C/N ratio to C stocks of Antarctic soils. Nitrogen deficiency has been linked to limiting decomposition of plant-based materials

(Jingguo & Bakken 1997). However, highest bacterial biomass is located in surface levels, independent of actual high C/N ratios (Bölder *et al.* 1999). It is also well known that N has significant importance in SOM accumulation, since it is effective for organic matter humification routes (Silva & Mendonça 2007). These soil characteristics are therefore attributes that may be useful to infer soil C losses through CO<sub>2</sub> emission in ice-free zones, during the Antarctic summer. By comparing the data in Fig. 4, a decreasing trend in the coefficient of determination for the soil with vegetation suggests that other factors, besides the C/N ratio and qMIC, are affecting the sensitivity of emissions. This result is corroborated by data obtained by La Scala *et al.* (2010), showing that soil temperature exerts a controlling factor on temporal variations in soil CO<sub>2</sub>-C emissions, and points out that vegetation is an additional important effect.

The experimental results of vegetated soils indicate, after cluster analysis, that B values of soil 3 plus vegetation were intermediate between values for soils 1 and 2 and soil 4 and 5 (Table IV). This is probably due to the higher changes in CO<sub>2</sub>-C emissions promoted by photosynthetic activity and evolved root exudates (Mendonça *et al.* 2011), which are more subject to microbial degradation. The studied soils have SOM with greater susceptibility to degradation when exposed to environmental conditions than SOM from other regions (Carvalho *et al.* 2010). Results in the present work is in the range of Q10 reported for forest soils (Epron *et al.* 1999, Davidson *et al.* 2006), which have much greater SOM contents, hence having high sensitivities.

The above discussion requires a potential scenario in which the regional temperature would significantly increase, consequently reducing the soil C stock by increasing CO<sub>2</sub> emissions. However, increasing soil biomass due to greater photosynthesis and primary productivity may counteract this trend (Michel *et al.* 2006). Hence, the SOM balance would be a result of the soil C input/output at this new hypothetical scenario.

## Conclusions

Maritime Antarctic soils having lower humification degree showed higher CO<sub>2</sub>-C emissions. A significant relationship ( $P < 0.05$ ) between CO<sub>2</sub>-C emission and soil temperature was observed for both bare soil or vegetated soil.

The emission of CO<sub>2</sub>-C showed significant exponential relationship ( $P < 0.05$ ) with temperature, with increasing emission of CO<sub>2</sub>-C with increasing temperature.

The sensitivity of CO<sub>2</sub>-C emissions in relation to temperature showed significant correlation with the degree of humification and microbial activity. Thus, further losses of CO<sub>2</sub>-C with rising temperature are expected in soils with a lower degree of humification.

The average Q10 values for the soils did not differ, but overall values were higher than those observed elsewhere.

The high Q10 value is related to the fragility of the SOM with low degree of humification and chemical characteristics of humic substances, becoming more susceptible to microbial degradation.

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