# Mineralogy of the Vattikod lamproite dykes, Ramadugu lamproite field, Nalgonda District, Telangana: A possible expression of ancient subduction-related alkaline magmatism along Eastern Ghats Mobile Belt, India

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# ABSTRACT

The mineralogy of nine recently discovered dykes (VL1:VL8 and VL10) in the vicinity of Vattikod village, Nalgonda district in Telangana State is described. The mineral assemblage present and their compositions are comparable to those of bona fide lamproites in terms of the presence of phlogopite (Ti-rich, Al-poor phlogopite and tetraferriphlogopite); amphiboles (potassic-arfvedsonite, potassic-richterite, potassicferro-richterite, potassic-katophorite, Ti-rich potassic-katophorite, Ti-rich potassic-magnesio-katophorite); Al-poor clinopyroxenes; feldspars (K-feldspar, Ba-K-feldspar and Na-feldspar) and spinels (chromitemagnetite and qandilite-ulvöspinel-franklinite). These dykes have undergone diverse and significant degrees of deuteric alteration as shown by the formation of secondary phases such as: titanite, allanite, hydro-zircon, calcite, chlorite, quartz and cryptocrystalline SiO<sub>2</sub>. On the basis of their respective mineralogy: the VL4 and VL5 dykes are classified as pseudoleucite-phlogopite lamproite; VL2 and VL3 dykes as pseudoleucite-amphibole-lamproite; and VL6, VL7 and VL8 as pseudoleucite-phlogopiteamphibole-lamproite. VL10 is extensively altered but contains fresh euhedral apatite microphenocrysts together with pseudomorphs after leucite and is classified as a pseudoleucite-apatite-(phlogopite?) lamproite. The mineralogy of the Vattikod lamproite dykes is compared with that of the Ramadugu, Somavarigudem and Yacharam lamproite dykes which also occur in the Ramadugu lamproite field. The lamproites from the Eastern Dharwar Craton are considered as being possible expressions of ancient subduction-related alkaline magmatism along the Eastern Ghats mobile belt.

**KEYWORDS:** lamproite, pseudoleucite, phlogopite, K-Na-amphiboles, Vattikod, Eastern Dharwar Craton, subduction.

## Introduction

LAMPROITES are mantle-derived, volatile-rich alkaline igneous rocks. These rocks are unusual in terms of their mineralogy and economically very important with respect to their diamond potential (Mitchell and Bergman, 1991; Mitchell, 1995).

\*E-mail:gurmeet28374@yahoo.co.in https://doi.org/10.1180/minmag.2017.081.045 Lamproite magmas have been considered as originating in two broad tectonic environments: (1) subduction settings and commonly termed Mediterranean lamproites (Mitchell and Bergman, 1991; Conticelli, 1998; Murphy *et al.*, 2002; Prelević *et al.* 2008; Tommasini *et al.*, 2011; Fritschle *et al.*, 2013; Perez-Valera *et al.*, 2013) or (2) within-plate cratonic regions (Leucite Hills, West Kimberley; Mitchell and Bergman, 1991; Mitchell, 1995). Lamproite magmas might originate from sources varying from the sub-continental

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lithospheric mantle to asthenospheric and deeper mantle material (Tainton and McKenzie, 1994; Mitchell, 1995; Murphy *et al.*, 2002; Nowell *et al.*, 2004; Davies *et al.*, 2006; Mirnejad and Bell, 2006; Chakrabarti *et al.*, 2007; Rapp *et al.*, 2008; Mitchell and Tappe, 2010; Tappe *et al.*, 2007, 2013).

The identification of a suite of rocks as lamproite or kimberlite relies on detailed mineralogical studies to establish the presence or absence of typomorphic minerals (Mitchell and Bergman, 1991, Mitchell, 1995). Identification of lamproites on the basis of their bulk-rock geochemistry is possible only for fresh rocks. Note that unless the rocks are glassy the whole-rock composition is actually determined by the mineralogy and not vice versa. The analysis of altered rocks is not a useful endeavour and can lead to completely inappropriate genetic conclusions. In this work, using a mineralogical-genetic system (Mitchell, 1995; Mitchell and Bergman, 1991) we classify the dyke rocks in the Vattikod area as bona fide lamproites and show these to be members of a suite of lamproites emplaced along the eastern margins of the Eastern Dharwar Craton (Ahmed and Kumar, 2012; Kumar et al., 2013a; Chalapathi Rao et al., 2014). On the basis of our classification we make further inferences as to the role of subduction in their genesis but no claims that these rocks are analogous to young Mediterranean-type lamproites. Rather, they are considered as rocks crystallized from magmas originating from ancient metasomatized lithospheric mantle which contains a subducted component (Chalapathi Rao et al., 2004; Mitchell, 2006; Chakrabarti et al., 2007; Das Sharma and Ramesh, 2013; Gurmeet Kaur and Mitchell, 2013; Gurmeet Kaur and Mitchell, 2016: Gurmeet Kaur et al., 2016).

# **Dharwar Craton lamproites**

The Dharwar Craton contains numerous kimberlites and lamproites (Neelkantam, 2001; Fareeduddin and Mitchell, 2012; Chalapathi Rao and Srivastava, 2016; Shaikh *et al.*, 2016). These rocks are disposed almost parallel to the interface of the juxtaposed Eastern Ghats Mobile Belt and the Eastern Dharwar Craton (Fig. 1; Neelkantam, 2001; Fareeduddin and Mitchell, 2012; Gurmeet Kaur and Mitchell, 2016). The lamproite fields in the Eastern Dharwar Craton are: (1) The P2-West, P12, P5, P13, TK1 and TK4 intrusions of Wajrakarur field; (2) the Chelima, Zangamarajupalle, Garledinne, Banganapalle lamproites of the Cuddapah Basin; (3) the Krishna lamproite field; and (4) the Ramadugu lamproite field (Fareeduddin and Mitchell, 2012; Gurmeet Kaur *et al.*, 2013; Gurmeet Kaur and Mitchell, 2016; Chalapathi Rao and Srivastava, 2016 and references therein; Shaikh *et al.*, 2016). Many of the above rocks were considered previously to be 'kimberlites' in previous investigations but are now reclassified as lamproites (see Fareedudddin and Mitchell, 2013; Gurmeet Kaur *et al.*, 2013; Gurmeet Kaur and Mitchell, 2013; Gurmeet Kaur *et al.*, 2016).

New hypotheses on the nature of the crustmantle lithosphere components of Eastern Dharwar Craton have been proposed on the basis of seismic tomographical studies. The crustal thickness of 33– 39 km with a Moho depth of ~40 km and an average heat flow of  $36\pm 8$  mW/m<sup>2</sup> has been evaluated for the Eastern Dharwar Craton (Gupta *et al.*, 2003; Roy and Mareschal, 2011; Kumar *et al.*, 2013*b*). Of particular importance with respect to lamproite genesis, Das Sharma and Ramesh (2013) suggest that a thick lithospheric root underlies southeast India, with the Archaean Eastern Dharwar Craton and the Proterozoic Eastern Ghats Mobile Belt being underlain by a relict subducted slab within the upper mantle.

# Ramadugu lamproite field

The Ramadugu lamproite field was discovered by Sridhar and Rau (2005) in the Nalgonda district of Telangana state (formerly Andhra Pradesh) during a diamond exploration programme initiated by the Geological Survey of India along the Krishna River. The Ramadugu lamproite field lies northwest of the Cuddapah Basin and close to the Krishna lamproites in the east (Fig. 1). The Ramadugu lamproite field consists of dykes Somavarigudem, occurring at Ramadugu, Yacharam and Vattikod (Fig. 2). The Ramadugu lamproites are emplaced in granodiorites and granites of the Peninsular Gneissic Complex of the Eastern Dharwar Craton (Fig. 2). The lamproite dykes have a general NW-SE strike with variable lengths from a few metres to  $\sim$ 700 m, with a maximum width of 3.5 m (Sridhar and Rau, 2005; Ahmed and Kumar, 2012; Kumar et al., 2013a; Chalapathi Rao et al., 2014). The lamproites occurring in the Krishna, Nallamalai and Ramadugu lamproite fields are diamondiferous (Chalapathi Rao et al., 2014).



FIG. 1. Distribution of kimberlites and lamproites in the Bundelkhand, Singhbhum, Bastar and Dharwar cratons of the Indian subcontinent. Diamonds ( $\blacklozenge$ ), circles ( $\blacklozenge$ ) and crosses (x) in the figure refer to kimberlites, lamproites and deformed alkaline rocks and carbonatites (DARC) locations in the southern Indian sub-continent, respectively. Bu – Bunder lamproites, M – Majhgawan lamproite field, B – Basna kimberlite field, Na – Nawapara lamproite field, Mp – Mainpur kimberlite field, Tk – Tokapal kimberlite field, Ra – Ramadugu lamproite field, N – Narayanpet kimberlite field, R – Raichur kimberlite field, T – Tungabhadra kimberlite field, W – Wajrakarur kimberlite field, NI – Nallamalai lamproite field, K – Krishna lamproite field, and D – Damodar valley lamproites (Gurmeet Kaur and Mitchell, 2016).

# Vattikod dykes

The Vattikod lamproite dykes, within the Ramadugu lamproite field, were discovered by Ahmed and Kumar (2012). The ten dykes (VL1: VL10) are spread over an area of ~6 square km to the west of Vattikod village (N16°55'13.2″ E79° 05'55″) and ~22 km north-west of Ramadugu village (Fig. 2; modified after Kumar *et al.*, 2013*a*). The dyke swarm follows a WNW-ESE to NW-SE trend traversing the Peninsular Gneissic Complex (Fig. 2). The lengths of the dykes are difficult to ascertain as most are covered by soil (Supplementary Figure S1, see below).

Nine dyke samples, VL1 to VL8 and VL10, were collected during March 2014 from the vicinity of Vattikod village. Brief field records of the occurrence are given in Table 1. (For detailed field records refer to the report of Ahmed and Kumar (2012) and Kumar *et al.* (2013*a*).

A very brief general account of the petrology of a few Vattikod dykes was given by Kumar *et al.* (2013*a*). In the present study each dyke was characterized on the basis of its major, minor and accessory mineralogy. We also attempt to determine the sequence of evolution of these nine dykes on the basis of their typomorphical mineralogy. The mineralogy of the Vattikod dykes is compared



FIG. 2. Ramadugu lamproite field consisting of the Ramadugu, Yacharam, Somavarigudem and Vattikod lamproite dykes, Nalgonda district, Telangana, India. The location of Vattikod lamproites and other Ramadugu lamproite dykes are marked on the map.

with that of the Ramadugu, Somavarigudem and Yacharam lamproite dykes, which also form part of the Ramadugu lamproite field and lie to the southeast of the Vattikod dykes. processed using Oxford *Aztec* software. Standards used are those given by Liferovich and Mitchell (2005).

# **Analytical techniques**

Representative samples of Vattikod lamproites were investigated by back-scattered electron (BSE) imagery and quantitative energy-dispersive X-ray spectrometry using a Hitachi SU-70 scanning electron microscope at Lakehead University, Ontario, Canada. All raw X-ray data were acquired using a beam current of 300 pA, an accelerating voltage of 20 kV and 30–60 s counting times and

# Petrography and mineralogy of Vattikod dykes

The Vattikod dykes are fine-grained rocks with phenocrysts and microphenocrysts of pseudoleucite, phlogopite, clinopyroxene, apatite and pseudomorphed olivine (Figs 3–7). Phlogopite and apatite are the only preserved phenocryst and microphenocryst primary phases (Figs 4*c*, 6*c*), as all other phenocryst phases such as leucite, clinopyroxene and olivine have been pseudo-morphed by K-feldspar, calcite, apatite, chlorite,

TABLE 1. Details of Vattikod lamproite dykes, Ramadugu lamproite field, Telangana, India (modified after Kumar *et al.*, 2013*b*).

Dyke Nos.	Dimensions		Trends	Latitude and Longitude details
VL1	1.7 m wide		N72°W	N16°55'07.5", E79°05'05.2"
VL2, VL3	639 m long, 2 m wide		N60°W	N16°54'59.3"; E79°05'18.4"
VL4	4.3 m long, 0.15 m wide		N60°W	N16°55'02.1"; E79°05'19.5"
VL5	7 m long, 0.15–0.30 m wide		$N47^{\circ}W - N70^{\circ}W$	,
VL6	50 m long, 0.65 m wide	(		
VL7	20 m long, 0.28 m wide	- 2	$N50^{\circ}W - N60^{\circ}W$	N16°55'21.6"; E79°04'21.7"
VL8	3 m long, 0.08–0.10 m wide	l		,
VL10	8.4 m long, 0.60 m wide		N80°E	N16°54'58.7"; E79°05'28.3"



FIG. 3. Plane-polarized light images: (*a*) the texture of the VL2 showing greenish-brown prismatic amphiboles, ovoid pseudoleucites, and groundmass titanite aggregates in fine grained matrix; (*b*) the texture of the VL5 lamproite dyke illustrating the presence of phenocrysts of phlogopite-tetraferriphlogopite and pseudoleucite set in a fine-grained groundmass material with predominant phlogopite; (*c*) flow texture around pseudoleucite in VL 5; and (*d*) the texture of VL6 with amphiboles, phlogopites and pseudoleucites set in a fine-grained groundmass.

quartz and cryptocrystalline SiO<sub>2</sub> (Figs 3b,c, 4a, 6b). The phenocryst and microphenocryst phases are set in a fine-grained matrix composed of phlogopite-tetraferriphlogopite, amphibole, clinopyroxene, K-feldspar (pseudoleucite), spinel, apatite, monazite, calcite, baryte, titanite, rutile and allanite. The mesostasis in which the above minerals are set is composed of chlorite, quartz and cryptocrystalline SiO<sub>2</sub>. Other minor phases are dolomite, magnetite, pyrite, Ba-K-feldspar, Nafeldspar, hydro-zircon, strontianite and Co-Nibearing copper sulfides. Clasts composed of finergrained material have been observed in VL2 and VL4. These clasts consist of material very similar to the groundmass phases in VL2 and VL4. The VL2 clasts contain more titanite, rutile, hydro-zircon, cryptocrystalline SiO2 and magnetite in comparison to VL2 groundmass which has more pseuodoleucite, amphibole, apatite and chlorite (Fig. 4b). The VL4 clasts have more rutile and cryptocrystalline SiO<sub>2</sub> and less calcite and K-feldspar, in comparison to a groundmass of VL4.

Phlogopite occurs as phenocrysts, microphenocrysts and as a groundmass phase. Phlogopite phenocrysts and microphenocrysts are prominent in dyke VL5 and are rarely preserved in the other dykes due to diverse degrees of alteration. Phlogopites are zoned, and exhibit the typical yellow-orange pleochroism of lamproite phlogopite together with thin rims of dark red tetraferriphlogopite (Fig. 3a). Phlogopites in VL5 are devoid of inclusions, and unaltered (Fig. 4c) in comparison to phlogopites in other dykes which have corroded margins, are poikilitic, and show alteration to chlorite (Fig. 4d). The poikilitic phlogopites commonly have inclusions of apatite, spinel,



FIG. 4. BSE images: (a) calcite-K-feldspar pseudomorph, titanite and cryptocrystalline  $SiO_2$  in highly altered VL1 dyke; (b) finer grained clast enriched in rutile, hydro-zircon, titanite and cryptocrystalline  $SiO_2$  in VL2 dyke; (c) zoned microphenocryst of phlogopites set in groundmass material enriched in phlogopites in VL5; and (d) groundmass phlogopites altering to chlorite in VL8.

clinopyroxene and chlorite. Most of the phenocrysts and microphenocrysts of phlogopites in other Vattikod dykes are pseudomorphed by chlorite, titanite and allanite (Fig. 6a). Titanite and allanite occur mainly along the cleavage planes and margins of the altered phlogopites. Groundmass phlogopite occurs in VL4, VL5, VL6, VL7, VL8 and is negligible-to-subordinate in VL2 and VL3 dykes. In VL1 phlogopite has not been identified, whereas the former presence of phlogopites in VL10 is suggested by the presence of chlorite pseudomorphs. Dykes VL4 and VL5 preserve the freshest groundmass phlogopites. The phlogopites in VL6, VL7 and VL8 are partially- to- almost completely- altered to chlorite whereas phlogopites in VL2 and VL3 are almost completely altered to chlorite. Groundmass micas are tetraferriphlogopites, which are identified by their reddish reverse pleochroism (Figs 3d). The VL5 groundmass

phlogopites exhibit a flow texture (Figs. 3*c*, 6*b*), which is unique relative to phlogopites in all other Vattikod dykes.

Amphiboles are a common groundmass phase in the VL2, VL3, VL6, VL7 and VL8 dykes and occur as zonation-free slender prisms and as wedgeshaped, euhedral crystals (Figs 5a-d). The crystals range in size from 200-10 µm. The amphiboles are both zoned and zonation-free. The zoned amphibole can have three different zones (Fig. 5d; Table 3). The amphiboles are associated with apatite, hydro-zircon (Fig. 5a), allanite (Fig. 5b) and baryte (Fig. 5c), and can be enclosed within rutile grains (Fig. 5d). Clinopyroxene occurs as microphenocrysts and as a groundmass phase (300-30 µm) in dykes VL2, VL3 and VL7 and are subordinate in comparison to amphiboles. Most of the clinopyroxenes are altered to chlorite (Figs 6a). In places, pseudomorphs of magnesian chlorite



FIG. 5. BSE images: (a) euhedral prismatic and wedge shaped amphiboles, both zoned and unzoned in VL2 in association with groundmass calcite, hydro-zircon, and apatite; (b) euhedral-to-subhedral amphiboles in association with allanite in VL6; (c) euhedral-to-subhedral amphiboles in association with baryte in VL6; and (d) amphiboles completely enclosed inside rutile in VL8.

after clinopyroxene have developed titanite and calcite rims.

The most common mineral in the Vattikod dykes is K-feldspar which occurs as phenocrysts and microphenocrysts, pseudomorphs after leucite, and as fine-grained groundmass material. That the pseudoleucite represents former primary leucite is a conclusion drawn from the typical habit of the pseudomorphs (Figs 3b,c). In dykes VL6, VL7 and VL8, most of the leucite pseudomorphs contain hydro-zircon aggregates occupying the core together with K-feldspar and calcite (Fig. 6b). K-feldspar, calcite, apatite, chlorite, Na-feldspar and K-Ba feldspar (hyalophane) are also components of the pseudomorphs. The groundmass K-feldspars also form small ovoids (<50 µm) mostly formed after leucites (Fig. 3b,d). K-feldspar is also found as an interstitial material which is considered to be a late-stage crystallization phase. Fresh leucite has not been observed in any of the Vattikod dykes.

Spinel occurs as euhedral-to-subhedral crystals (<50 µm) as a groundmass phase (Fig. 7*a*). It also occurs within, or at, the margins of some of the phenocrystal phases such as pseudoleucites and pseudomorphed phlogopites. The groundmass spinels are both zoned and zonation-free. Apatite occurs as a phenocryst- to- microphenocryst phase in dykes VL2, VL3 and VL10 (Fig. 6c). Groundmass apatite occurs principally as euhedral-tosubhedral grains (50-5 µm), and as anhedral aggregates (Fig. 6d). Apatite crystals are also poikilitically-enclosed by groundmass phlogopites together with titanite aggregates. Apatites are also associated closely with other groundmass phases such as titanite, rutile, calcite, monazite and hydrozircon (Fig. 6d). Monazite-(Ce) occurs as a late-



FIG. 6. BSE images: (*a*) development of titanite along the cleavage planes of a prismatic ferromagnesian mineral (phlogopite/pyroxene?) now pseudomorphed by chlorite; (*b*) K-feldspar and calcite pseudomorph after leucite in VL5, with a clear flow texture in VL5 visible; (*c*) euhedral groundmass apatites in VL5, with rutile grains visible; and (*d*) anhedral patches of monazite in VL4 along with groundmass phlogopites.

stage anhedral groundmass phase (up to 50  $\mu$ m) in VL4 and VL5 dykes (Fig. 6*d*). Monazite occurs in association with phlogopite, apatite, K-feldspar, calcite and cryptocrystalline SiO<sub>2</sub>.

Titanite, allanite, calcite, baryte and hydro-zircon are ubiquitous groundmass phases in all of the Vattikod dykes (Figs 7*a*–*d*). Titanite occurs primarily as: (1) aggregates forming part of the groundmass; (2) inside, and along, the margins of the pseudomorphs after leucite, clinopyroxene and phlogopite. The titanite of parageneses (1) seems to be a latestage phase in the groundmass, whereas paragenetic type (2) is secondary phase formed as a result of alteration/reaction between some phases and deuteric fluids. Rutile in the dykes occurs as a late-stage mineral in variable sizes as subhedral-to-euhedral crystals (100  $\mu$ m to <5  $\mu$ m), and is associated commonly with titanite, apatite and hydro-zircon.

Allanite occurs as aggregates forming part of the groundmass and is commonly seen replacing chlorite pseudomorphs (Figs 7c,d). Rutile is not a common phase in lamproites but has been reported from Raniganj lamproites (Mitchell and Fareeduddin, 2009). Rutile is reported from Somavarigudem lamproite dykes of Ramadugu lamproite field (Chalapathi Rao et al., 2014). Rutile is a common phase in ultramafic lamprophyres and calcite kimberlites (Zurevinski and Mitchell, 2011; Tappe et al., 2006, 2014). Allanites have not been previously recognized in Vattikod and other Ramadugu lamproites (Kumar et al., 2013a; Chalapathi Rao et al., 2014). Titanites are not primary phases and have formed as a result of deuteric alteration. Titanites, both primary and secondary, have been reported from other Ramadugu lamproites by Chalapathi Rao et al., (2014).



FIG. 7. BSE images: (*a*) subhedral spinels inside leucite pseudomorphs; (*b*) intergrown magnetite and pyrite in VL6; (*c*) aggregate of allanite and baryte in the groundmass of VL6; and (*d*) allanite and titanite replacing an earlier phase along the margins, which is now chlorite in VL10.

Calcite occurs as a groundmass phase forming: (1) late-stage aggregates (Fig. 5a); and (2) pseudomorphs after some earlier-crystallized minerals such as leucite, pyroxene (Fig. 6b). The groundmass calcites in almost all dykes are probably latestage residual phases as they fill the interstitial spaces between the earlier-formed groundmass phases (Fig. 5a). Baryte occurs as anhedral patches in the groundmass (Fig. 5c), and occurs in association with calcite, amphibole and allanite and pseudomorphed leucite. Hydro-zircon is present in all the dykes and is also present in the cores of pseudomorphed leucite as aggregates of very fine grains. Strontianite and dolomite are very rarely present as a groundmass phase in the VL4 and VL5 dykes.

Chlorite and quartz are present as a mesostasis material in almost all the dykes. Chlorite replaces groundmass phlogopite, amphibole and pyroxene. Relatively coarse-grained aggregates of quartz are commonly seen in the pseudomorphed phases after leucite. Cryptocrystalline  $SiO_2$  is present in all dykes as a late-stage, anhedral groundmass phase.

Magnetite and pyrite (Fig. 7*b*) are observed in almost all the dykes in accessory amounts. Rarely present are very small anhedral crystals of Co-Nibearing chalcopyrite.

## Mineral compositions

#### Phlogopite

Representative compositions of phlogopite are given in Table 2 and Supplementary Table 1. The cores of phenocrysts of phlogopites contain 10.8–10.5 wt.%  $Al_2O_3$  and 4.8–4.7 wt.%  $FeO_T$  with the rims being relatively depleted in  $Al_2O_3$  (6.5–6.3 wt.%) and enriched in  $FeO_T$  (15.0–13.5 wt.%). No significant difference exists between the TiO<sub>2</sub> contents of the core and rim of zoned phenocrysts and microphenocrysts (Table 2). The groundmass phlogopites and tetraferriphlogopites contain

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Š		Strı Si

	1	2	3	4	5	6	7	8	9	10	11	12	13
Wt.%	VL4	VL4	VL5 C	VL5 R	VL5 C	VL5 R	VL5	VL5	VL6	VL6	VL7	VL7	VL8
SiO <sub>2</sub>	40.99	40.68	41.65	41.66	41.53	41.20	41.96	40.07	39.55	39.74	39.68	39.08	42.29
TiO <sub>2</sub>	6.11	5.69	5.50	5.42	5.57	5.64	5.92	5.15	5.12	5.12	5.22	4.43	6.00
$Al_2 \tilde{O}_3$	6.43	7.31	10.77	6.33	10.53	6.45	6.01	8.03	8.57	7.13	7.40	8.40	5.25
FeO(t)	15.78	16.72	4.73	13.45	4.85	15.01	16.58	15.88	16.62	20.65	20.88	20.52	16.69
MgO	14.30	14.21	23.73	17.05	23.48	15.84	15.37	15.34	15.60	13.44	13.71	14.31	14.93
$K_2O$	9.26	9.13	10.19	9.72	10.31	9.81	9.73	9.04	9.51	8.91	9.15	9.05	9.12
BaO	1.35	1.62	1.01	1.06	0.81	1.08	0.96	1.16	1.62	1.51	n.d.	1.22	1.54
F	1.72	1.23	1.60	1.82	1.68	1.46	1.12	0.99	0.67	0.76	1.02	0.92	1.09
Total	94.22	95.36	97.58	94.69	97.08	95.03	96.53	94.67	96.59	96.50	96.04	97.01	95.82
Structura	al formula ca	lculated on	the basis of	16 cations									
Si	6.557	6.443	5.992	6.492	6.005	6.453	6.514	6.314	6.119	6.279	6.230	6.081	6.670
Ti	0.735	0.678	0.595	0.635	0.606	0.664	0.691	0.610	0.596	0.608	0.616	0.518	0.712
Al	1.212	1.364	1.826	1.163	1.794	1.191	1.100	1.491	1.563	1.328	1.369	1.540	0.976
Fe	2.111	2.215	0.569	1.753	0.586	1.966	2.153	2.093	2.150	2.729	2.742	2.670	2.201
Mg	3.410	3.355	5.090	3.961	5.061	3.699	3.557	3.603	3.598	3.166	3.209	3.319	3.510
Κ	1.890	1.845	1.870	1.932	1.902	1.960	1.927	1.817	1.877	1.796	1.833	1.796	1.835
Ba	0.085	0.101	0.057	0.065	0.046	0.066	0.058	0.072	0.098	0.093	_	0.074	0.095
F	0.870	0.616	0.728	0.897	0.768	0.723	0.550	0 493	0.328	0.380	0.507	0.453	0.544

BLE 2. Representative compositions (wt.%) of phlogopites.

n.d. - not detected; FeO(t) -total Fe expressed as FeO; C - Core and R - Rim.

TABLE 3.	Representative	compositions	(wt.%)	of amphiboles.
	1	1	< / >	1

Wt.%	1 VL2	2 VL2 C	3 VL2 R	4 VL3	5 VL6	6 VL6	7 VL7	8 VL7	9 VL8 C	10 VL8 R1	11 VL8 R2
SiO <sub>2</sub>	52.46	52.74	53.75	54.00	53.02	50.61	48.97	51.03	52.36	52.27	53.83
TiO <sub>2</sub>	1.67	2.06	1.50	4.81	2.63	6.30	3.73	1.77	4.01	2.52	4.13
$Al_2 \tilde{O}_3$	n.d.	n.d.	n.d.	0.26	0.32	0.48	0.25	0.23	n.d.	n.d.	n.d.
MnO	0.38	0.29	n.d.	n.d.	0.32	0.19	0.52	0.45	n.d.	n.d.	n.d.
FeO(t)	22.66	16.49	18.40	19.11	16.20	13.75	22.19	25.30	8.71	14.71	14.51
FeO	20.01	15.53	18.40	n.d.	16.20	13.75	22.19	24.39	8.71	14.71	14.51
Fe <sub>2</sub> O <sub>3</sub>	2.946	1.063	n.d.	n.d.	n.d.	n.d.	n.d.	1.01	n.d.	n.d.	n.d.
MgO	9.02	12.65	10.69	10.21	11.29	12.48	7.63	6.01	16.61	12.22	12.50
CaO	2.60	4.09	4.95	0.61	5.43	5.58	5.13	2.77	5.15	4.38	1.14
Na <sub>2</sub> O	5.29	4.62	6.67	7.12	4.14	3.94	4.29	5.16	3.89	4.44	6.92
K <sub>2</sub> Ō	5.05	5.07	4.97	0.45	5.08	5.09	4.89	4.91	5.34	5.16	5.08
F	0.64	n.d.	n.d.	n.d.	1.00	0.66	n.d.	n.d.	1.18	0.86	n.d.
Structural formula calcu	ulated on th	e basis of 2	3 cations								
Formula assignments	K-AF	K-RT	K-RT	Ti-Mg- Rieb	K-KAT	Ti-K-KAT	K-Fe-RT	K-AF	K-RT	K-RT	Ti-K-Mg AF
T site Si	7.917	7.878	7.924	7.876	7.904	7.517	7.655	7.978	7.745	7.954	7.946
Al	_	_	_	0.045	0.056	0.084	0.046	0.022	_	_	_
Ti	0.083	0.122	0.076	0.079	0.040	0.399	0.299		0.255	0.046	0.054
C site Fe <sup>3+</sup>	-	_	_	0.000	_	_	_	-	_	_	_
Ti	0.107	0.110	0.090	0.449	0.255	0.305	0.139	0.208	0.191	0.243	0.405
Al	-	_	_	_	_	_	_	0.020	_	_	_
Fe <sup>3+</sup>	0.335	0.119	_	0.859	_	_	_	0.121	_	_	_
Mn <sup>2+</sup>	0.004	0.013	_	_	0.040	0.024	0.069	0.060	_	_	-
Fe <sup>2+</sup>	2.525	1.941	2.268	1.472	2.020	1.708	2.901	3.187	1.077	1.872	1.791
Mg	2.029	2.817	2.349	2.220	2.509	2.763	1.778	1.401	3.663	2.772	2.751
Mn <sup>2+</sup>	0.045	0.024	_	-	_	_	_	_	_	_	_
B site Ca	0.420	0.655	0.782	0.095	0.867	0.888	0.859	0.464	0.816	0.714	0.180
Na Ca	1.535	1.322	1.218	1.905	1.133	1.112	1.141	1.536	1.116	1.286	1.820
A site Na	0.013	0.016	0.688	0.109	0.064	0.023	0.159	0.028	_	0.024	0.161
Κ	0.972	0.966	0.935	0.084	0.966	0.964	0.975	0.979	1.008	1.002	0.957
F	0.305	-	-	-	0.471	0.310	-	-	0.552	0.414	-

n.d. – not detected; Fe<sub>2</sub>O<sub>3</sub> and FeO calculated on a stoichiometric basis; C – Core; R – Rim. K-AF: potassic-arfvedsonite; K-RT: potassic-richterite; Ti-Mg-Rieb: Ti rich Mg riebeckite; K-KAT: potassic-katophorite; Ti-K-KAT: Ti-rich potassic-katophorite; K-Fe-RT: potassic-ferro-richterite; Ti-K-Mg AF: Ti-rich potassic-magnesio-arfvedsonite.



FIG. 8. Al<sub>2</sub>O<sub>3</sub> vs. FeO<sub>T</sub> compositional variation of phlogopite in Vattikod lamproites. Also shown is the field for phlogopites from other Ramadugu lamproites. Compositional fields and trends for kimberlites, lamproite, orangeite and minette micas from Mitchell (1995).

8.6–5.3 wt.%  $Al_2O_3$  and 20.9–15.8 wt.%  $FeO_T$  and are enriched in TiO<sub>2</sub> (6.1–4.4 wt.%). The  $Al_2O_3$ contents of Vattikod phlogopites compare well with the range of 5–11 wt.%  $Al_2O_3$  reported for other lamproite phlogopites (Jaques *et al.*, 1986; Mitchell, 1989; Mitchell and Bergman, 1991). The BaO contents of all micas are typically <2 wt. % and fluorine contents vary between 0.7–1.8 wt.% (Table 2).

The zoned phlogopite phenocrysts in VL5 are typical of lamproitic micas and the compositional zoning (core to rim) is a trend of decreasing Al<sub>2</sub>O<sub>3</sub> and MgO with increasing  $FeO_T$  (Table 2; Fig. 8). The tetraferriphlogopite rims are extremely enriched in FeO<sub>T</sub> and depleted in Al<sub>2</sub>O<sub>3</sub> (Table 2). Following Mitchell and Bergman (1991) the compositional evolution is considered to be from octahedral site-deficient Ti-rich phlogopite to tetraferriphlogopite. Mica compositional zonation trends are similar to those found in orangeites (also known as Kaapvaal lamproite; Mitchell, 2006) and lamproites (Mitchell and Bergman, 1991; Figs 8, 9). The Vattikod phlogopites are more evolved than those in other Ramadugu lamproites (Figs 8, 9; Chalapathi Rao et al., 2014).

#### Amphibole

The amphiboles exhibit a wide range in composition (Table 3 and Supplementary Table 2) and all have low  $Al_2O_3$  (<0.5 wt.%) contents typical of most lamproite amphiboles (Mitchell and Bergman, 1991). They contain (6.3 to 1.5 wt.%) TiO<sub>2</sub>, (7.1 to 3.9 wt.%) Na<sub>2</sub>O, (5.3 to 0.5 wt.%) K<sub>2</sub>O and (25.3 to 8.7 wt.%) FeO<sub>T</sub> with fluorine contents up to 1.1 wt.%.

The amphiboles show compositional evolution from Ti-rich potassic-magnesio-katophorite through Ti-rich potassic-katophorite, potassickatophorite, potassic-ferro-richterite and potassicrichterite to potassic-arfvedsonite (Table 3). Figures 10 and 11 show that the Vattikod amphiboles evolve to compositions that are far richer in FeO<sub>T</sub> than are typical of lamproites sensu lato, and are similar to evolved amphiboles from the Rice Hill lamproite (Mitchell and Bergman, 1991). The Vattikod amphiboles are also comparatively more evolved than those occurring in other Ramadugu lamproites (Figs 10, 11; Chalapathi Rao et al., 2014). Vattikod amphiboles are not compositionally equivalent to amphiboles in minettes, and other potassic rocks (Figs 10, 11; Mitchell and Bergman, 1991). The



FIG. 9. Al<sub>2</sub>O<sub>3</sub> vs. TiO<sub>2</sub> (wt.%) compositional variation of phlogopite in Vattikod lamproites. Also shown is the field for phlogopites from other Ramadugu lamproites. Compositional fields and trends for kimberlites, lamproite, orangeite and minette micas from Mitchell (1995).

extremely low  $Al_2O_3$  content of the amphiboles is attributed to the low alumina contents of their parental peralkaline magma (Wagner and Velde, 1986; Mitchell and Bergman, 1991).

# Clinopyroxene

Representative compositions of clinopyroxenes are given in Table 4. Two compositionally distinct clinopyroxenes are present; diopside and Na-Ferich pyroxene. Diopsides commonly occur as phenocrysts and a groundmass phase as reported in many lamproites (Table 4; Jacques *et al.*, 1986; Mitchell and Bergman, 1991). The sodic variety has been reported previously from Raniganj lamproites rimming diopside (Mitchell and Fareeduddin, 2009), although such iron-rich pyroxenes with (17.8–17.5 wt.%) FeO<sub>T</sub> and (1.3–0.4 wt.%) Na<sub>2</sub>O have not been reported from other lamproites (Table 3; Mitchell and Bergman, 1991). The Ti *vs*. Al diagram for all the varieties of clinopyroxenes clearly indicates their lamproitic affinity (Fig. 12).

#### K-feldspar

Representative compositions of K-feldspar, Ba-K-feldspar (hyalophane) and Na-feldspar are given in (Table 5). The pseudomorphic K-feldspars are comparable in composition to K-feldspars in other lamproites (Mitchell and Bergman, 1991), and are relatively poor in Na<sub>2</sub>O (n.d.–0.3 wt.%) and FeO<sub>T</sub> (0.4–2.1 wt.%; Table 5). Ba-K-feldspar and Na-feldspar occur in the ovoid aggregates (Table 5). Ba-K feldspars (hyalophane) have been reported from the Raniganj lamproites (Mitchell and



FIG. 10. Ti vs. Na/K (atoms per formula unit) compositional variation of amphiboles in Vattikode lamproites. The field for amphiboles from other Ramadugu lamproites is also shown. Compositional fields and trends for amphiboles in lamproites and other potassic rocks from Mitchell and Bergman (1991).



FIG. 11. FeO<sub>T</sub> vs. Na<sub>2</sub>O compositional variation of amphiboles from P-1 Vattikod lamproites. Also shown is the field for phlogopites from other Ramadugu lamproites. Compositional fields and trends for amphiboles in lamproites from Mitchell and Bergman (1991).

	1	2	2	4	5	6	7	0	0	10
Wt.%	VL2	VL3	VL3	VL3	VL3	VL3	VL3	vL3	VL7	VL7
SiO <sub>2</sub>	55.48	54.36	54.29	54.23	54.18	54.57	54.46	55.55	54.61	53.78
TiO <sub>2</sub>	0.44	1.23	1.18	1.31	1.04	1.16	0.49	0.53	2.39	2.00
$Al_2 \tilde{O}_3$	0.65	n.d.	n.d.	n.d.	n.d.	n.d.	0.53	0.56	0.49	0.45
FeO(t)	9.19	3.21	2.89	2.85	3.72	3.09	17.52	17.80	4.17	4.08
MgO	20.39	17.02	17.10	17.25	17.16	17.44	13.32	14.25	16.25	16.37
CaO	12.88	24.23	24.15	24.06	24.40	24.19	10.79	11.21	22.09	23.27
Na <sub>2</sub> O	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	1.31	0.42	n.d.	n.d.
Cr2O3	n.d.	0.57	0.57	0.66	n.d.	0.47	n.d.	n.d.	0.44	0.83
Total	99.03	100.62	100.18	100.36	100.50	100.92	98.42	100.32	100.44	100.78
Structural	formula calcula	ted on the basis	of 4 cations							
Si	2.046	1.981	1.985	1.979	1.974	1.979	2.093	2.105	2.009	1.970
Ti	0.012	0.034	0.032	0.036	0.028	0.032	0.014	0.015	0.066	0.055
Al	0.028	_	_	_	_	_	0.024	0.025	0.021	0.019
Fe	0.283	0.098	0.088	0.087	0.113	0.094	0.563	0.564	0.128	0.125
Mg	1.121	0.925	0.932	0.938	0.932	0.943	0.763	0.805	0.891	0.894
Ca	0.509	0.946	0.946	0.941	0.952	0.940	0.444	0.455	0.871	0.913
Na	-	-	-	-	-	-	0.098	0.031	-	_
Cr <sub>2</sub> O <sub>3</sub>	-	0.016	0.016	0.019	_	0.013	-	-	0.013	0.024

TABLE 4. Representative compositions (wt. %) of pyroxenes.

n.d. – not detected; FeO(t) – total Fe expressed as FeO.



FIG. 12. Compositional variation (Ti vs. Al in atoms per formula unit) of pyroxenes from Vattikod lamproites. Compositional fields and trends for lamproites, minettes, Roman province lavas and kamafugites from Mitchell and Bergman (1991).

Fareeduddin, 2009). Na-feldspar has not been reported in earlier studies of lamproites.

# Spinels

Representative compositions of spinel are given in (Table 6). The spinels contain Cr<sub>2</sub>O<sub>3</sub> (up to 53.0 wt.%), MgO (up to 4.2 wt.%),  $FeO_T$  (up to 41.7 wt.%), TiO<sub>2</sub> (up to 6.6 wt.%) and ZnO (0.42 to 5.6 wt.%). All spinels are chromium-rich and represent principally solid solutions between chromite and magnetite with minor amounts of gandilite, ulvöspinel and franklinite. Most of the groundmass spinels are not zoned. Minor continuous core-to-rim zoning is one of decreasing MgO and Cr2O3 and increasing total FeO and ZnO at nearly constant to higher Ti (Table 6). The Vattikod spinel compositions are shown in Fig. 13, projected onto the front face of the reduced spinel prism (Mitchell, 1986). The extreme Ti-enrichment which is common for lamproite spinels is not observed for Vattikod spinels (Mitchell, 1995; Mitchell and Fareeduddin, 2009). Figure 13 shows that these spinels are unlike all kimberlite spinels but are similar to relatively unevolved Ti-poor spinels in lamproites and orangeites (lamproite var. Kaapvaal). Similar trends of spinel compositions have been reported for Raniganj lamproites (Mitchell and Fareeduddin, 2009) and

also in ultramafic lamprophyres from Torngat (Tappe *et al.*, 2004). In comparison to spinels found in other Ramadugu lamproites, the Vattikod spinels are more evolved in terms of Fe and Ti (Fig. 13; Chalapathi Rao *et al.*, 2014). As most of the spinels are poor in alumina (<3.1 wt.%) and enriched in ZnO they indicate the peralkaline nature of the magma from which they crystallized, and their affinity to the lamproite clan.

## Apatite and monazite

Representative compositions of apatite are given in Table 7. The apatites are rich in SrO (up to 3.7 wt.%), and contain significant fluorine (up to 4.1 wt.%). They can be classified as fluorapatites, and are similar to those reported in many lamproites (Thy *et al.*, 1987; Edgar, 1989; Mitchell and Bergman, 1991). They contain no barium and are poor in light-rareearth elements. The sheaf-like quench apatites which are scattered throughout the groundmass of Vattikod lamproites are too small for quantitative analysis.

Representative compositions of monazite-(Ce) are given in Table 7. The monazite is enriched in  $Ce_2O_3$  (up to 34 wt.%), SrO (<2 wt.%) and with (up to 2 wt.%) ThO<sub>2</sub>. Monazites of similar composition have also been reported from the Raniganj lamproites (Mitchell and Fareeduddin, 2009).

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	1	2	3	4	5	6	7	8	9	10
Wt.%	VL1	VL2	VL3	VL4	VL5	VL6	VL7	VL8	VL3	VL4
SiO <sub>2</sub>	65.16	64.36	65.62	64.86	65.84	65.22	64.82	65.47	67.07	54.59
$Al_2O_3$	18.57	17.82	18.01	17.25	17.77	18.07	17.25	17.64	18.99	19.76
FeO(t)	0.39	1.35	0.61	2.10	0.44	0.82	1.58	0.46	0.24	n.d.
Na <sub>2</sub> O	n.d.	0.29	n.d.	0.20	n.d.	n.d.	0.18	n.d.	13.15	0.35
$K_2 \bar{O}$	16.08	15.67	16.08	15.11	15.91	15.54	16.10	16.04	n.d.	12.66
BaO	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	11.31
Total	100.20	99.49	100.32	99.52	99.96	99.65	99.93	99.61	99.45	98.67
Structural	formula calculate	ed on the basis o	f 8 atoms of oxy	gens						
Si	3.000	3.000	3.020	3.020	3.034	3.016	3.017	3.032	2.970	2.782
Al	1.008	0.979	0.977	0.947	0.965	0.985	0.946	0.963	0.991	1.187
Fe	0.015	0.053	0.023	0.082	0.017	0.032	0.061	0.018	0.009	_
Na	_	_	_	0.018	_	_	0.016	_	1.129	0.035
Κ	0.945	0.932	0.944	0.898	0.935	0.917	0.956	0.948	0.000	0.823
Ва	-	-	-	-	-	-	-	-	-	0.226

TABLE 5. Representative compositions (wt.%) of K-feldspar, Na-feldspar and Ba-K-feldspar.

n.d. - not detected; FeO(t) - total Fe expressed as FeO.

	1	2	3	4	5	7	9	10	11	12	13	14
Wt.%	VL1	VL2	VL3	VL4	VL5	VL6	VL7	VL10	VL7 C	VL7 R	VL8 C	VL8 R
TiO <sub>2</sub>	5.55	4.09	4.31	3.98	4.68	5.14	5.55	6.64	4.94	5.61	3.70	5.29
$Al_2\bar{O}_3$	2.00	2.88	1.94	1.88	2.19	2.08	2.07	1.40	1.97	2.01	3.07	2.16
FeO(t)	38.81	34.56	37.56	35.97	39.47	39.79	38.70	41.74	37.93	38.71	32.40	35.56
FeO	30.61	29.05	27.96	28.99	32.37	33.53	32.60	34.62	31.68	31.75	26.40	28.42
$Fe_2O_3$	9.11	6.12	10.67	7.76	7.89	6.96	6.78	7.92	6.94	7.73	6.67	7.93
MnO	2.07	2.02	1.83	1.90	2.10	2.14	2.29	1.69	1.93	2.05	1.97	2.29
MgO	2.77	0.61	0.30	n.d.	0.51	0.95	0.62	2.60	2.76	2.57	4.15	1.02
ZnO	0.60	3.83	5.56	5.33	3.63	2.24	4.60	0.42	1.41	2.02	1.23	4.83
$Cr_2O_3$	47.98	49.57	45.15	50.03	45.98	45.35	45.81	44.06	48.57	46.96	53.02	47.01
Total	100.69	98.17	97.72	99.87	99.35	98.39	100.32	99.34	100.21	100.70	100.21	98.95
Structura	l formula calc	ulated on the	basis of 32 a	atoms of oxy	gens							
Ti	1.203	0.923	0.988	0.894	1.047	1.155	1.230	1.461	1.075	1.218	0.793	1.187
Al	0.680	1.018	0.697	0.661	0.768	0.732	0.719	0.483	0.672	0.684	1.032	0.759
Fe2 <sup>+</sup>	7.380	7.288	7.128	7.238	8.054	8.377	8.033	8.473	7.666	7.665	6.295	7.091
Fe3 <sup>+</sup>	1.978	1.381	2.446	1.742	1.765	1.564	1.502	1.743	1.512	1.679	1.430	1.780
Mn	0.505	0.513	0.472	0.480	0.529	0.542	0.571	0.419	0.473	0.501	0.476	0.579
Mg	1.190	0.273	0.136	_	0.226	0.423	0.272	1.134	1.190	1.106	1.764	0.454
Zn	0.128	0.848	1.251	1.175	0.797	0.494	1.001	0.091	0.301	0.430	0.259	1.064
Cr	10.936	11.756	10.881	11.809	10.814	10.712	10.671	10.195	11.111	10.717	11.951	11.087

TABLE 6. Representative compositions (wt.%) of spinels.

n.d. - not detected; FeO(t) - total Fe expressed as FeO; C - core and R - Rim.



FIG. 13. Compositional variation of spinels from Vattikod lamproites projected onto the front face of the reduced iron spinel compositional prism (Mitchell, 1986). Compositional fields and trends for spinels from kimberlites (T1) and lamproites (T2) from Mitchell (1986, 1995).

Monazites have not been reported from Vattikod lamproite dykes by earlier workers (Kumar *et al.*, 2013*a*) or from other Ramadugu lamproites (Chalapathi Rao *et al.*, 2014).

#### **Discussion and conclusions**

We consider that bulk-rock geochemistry of these mineralogically complex and altered rocks cannot be used to characterize their parental magmas. The best way to characterize them is by consideration of their typomorphical mineralogy (Mitchell, 1991; Mitchell and Tappe, 2010) as given in Table 8.

The Vattikod mineral assemblage and their compositions are comparable to those of lamproites *sensu lato* in terms of the presence of pseudo-leucite, phlogopite-tetraferriphlogopite, K-Na-Ti amphibole, Al-poor clinopyroxene, apatite and spinels. The Vattikod lamproites do not contain fresh olivine but pseudomorphs after olivine are considered to be present on the basis of their morphology. Priderite and other Ti and K zirco-nium minerals (Mitchell and Bergman, 1991) have not been recognized in the Vattikod dykes. The dykes have also undergone diverse degrees of

deuteric alteration which is evident by the development of secondary phases such as titanite. allanite, hydro-zircon, calcite, chlorite, quartz and cryptocrystalline SiO<sub>2</sub>. The formation of titanite and allanite along the cleavages and margins of pseudomorphed phases is rarely observed in VL4 and VL5 dykes which also have very minor anhedral secondary titanite in comparison to other Vattikod dykes, thus indicating these to be the least altered dykes. In addition VL4 and VL5 dykes preserve fresh phlogopite and alteration to chlorite is very limited in comparison to other Vattikod dykes. All of the 'leucite' which was present as phenocrysts and groundmass phase, is now completely pseudomorphed by K-feldspar and secondary phases.

Taking into account the textural and mineralogical account of the various dykes we conclude that VL4 and VL5 are the least altered followed with increasing degrees of alteration by VL6, VL7, VL8, VL2, VL3, VL10 and V1. We classify the Vattikod dykes VL4 and VL5 as pseudoleucitephlogopite-lamproite; VL2 and VL3 as pseudoleuciteamphibole-lamproite; VL6, VL7 and VL8 as pseudoleucite-phlogopite-amphibole-lamproite.

Wt.%	1 VL 1	2 VL2	3 VL3	4 VL4	5 VL5	6 VL6	$_{\rm VL7}^{7}$	8 VL8	9 VL10	10 VL4	11 VL4	12 VL5	13 VL5
$P,O_{\varsigma}$	40.48	40.05	41.02	40.59	41.28	40.81	40.10	39.72	40.82	30.31	30.65	30.42	31.10
Si0,	0.64	0.67	1.02	1.99	0.93	1.43	1.22	2.19	1.16	1.12	1.82	2.02	2.38
FeO(t)	0.24	0.30	n.d.	n.d.	n.d.	n.d.	0.32	n.d.	n.d.	n.d.	n.d.	0.73	0.42
CaO	51.23	51.25	52.79	52.11	52.63	52.95	50.23	50.74	51.38	n.d.	n.d.	n.d.	n.d.
SrO	2.65	3.30	2.59	2.46	2.95	2.29	3.30	3.66	2.80	1.23	1.12	1.31	1.77
$La_2O_3$	n.d.	n.d.	n.d.	0.49	n.d.	0.44	0.60	n.d.	0.47	18.31	17.09	21.01	19.56
$Ce_{2}O_{3}$	n.d.	n.d.	0.56	1.12	0.68	0.93	1.19	0.94	0.83	34.16	33.34	32.85	32.87
Nd <sub>2</sub> O <sub>3</sub>	n.d.	n.d.	n.d.	0.36	0.39	0.37	0.47	0.57	0.48	8.26	8.78	6.63	8.00
$Pr_{20_3}$	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	3.13	3.09	2.72	2.95
$ThO_{2}$	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	0.86	1.70	0.65	0.25
н	4.05	3.93	2.89	3.20	3.36	3.61	3.30	3.26	3.18	n.d.	n.d.	n.d.	n.d.
Total	95.24	95.57	97.98	99.12	98.86	99.22	97.43	97.82	97.94	97.38	97.59	98.34	99.30

1-9: apatites; 10-13 monazites; n.d. - not detected; FeO(t) - total Fe expressed as FeO.

As VL1 is completely altered the precursor mineralogy cannot be identified. VL10 is also extensively altered but contains fresh euhedral apatite microphenocrysts together with pseudomorphs after leucite and is classified as a pseudoleucite-apatite-(phlogopite?) lamproite.

The amphiboles, phlogopites and spinels of the Vattikod dykes are more evolved in comparison to those in the Ramadugu, Yacharam and Somavarigudem lamproites of the Ramadugu lamproite field (Figs 8, 9, 10, 11 and 13; Chalapathi Rao *et al.*, 2014). The presence of monazite and allanite indicates enrichment in rareearth elements of the batch of magma from which these dykes rocks evolved.

Although these rocks have many of the mineralogical characteristics of lamproites (*sensu lato*), they are subtly different in terms of local mineralogical variation. Thus, the amphibole compositions are atypical of amphiboles from many lamproites in terms of their low  $TiO_2$  and high  $FeO_T$  contents, although they are similar to Raniganj and Ramadugu amphiboles (Mitchell and Bergman, 1991; Mitchell and Fareeduddin, 2009; Chalapathi Rao *et al.*, 2014). The overall compositional trend of spinels in these rocks is also similar to that found in lamproites, although it differs in that the Vattikod spinels have relatively low Ti/(Ti + Cr + Al) ratios of <0.6 and are enriched in Zn.

Notable differences between lamproites (sensu lato) and the Vattikod rocks include the presence of rutile, titanite, allanite, monazite, hydro-zircon, quartz and cryptocrystalline SiO<sub>2</sub> as a late-stage groundmass and alteration minerals. However, the textural and mineralogical data demonstrates that in terms of a mineralogical-genetic classification the Vattikod dykes are bona fide lamproites. It is suggested that the Vattikod lamproites represent a spectrum of modal variants of lamproite produced by the differentiation and crystallization of a common parental peralkaline potassic magma. The magma from which the dykes were formed is best regarded as the expression of a particular variety of cratonic potassic magmatism derived from a local metasomatized mantle source (Mitchell, 2006). Similar conclusions have been drawn for the Wajrakurur P2-West, P5, P12, P13, TK1 and TK4 intrusions in the Eastern Dharwar craton which are now reclassified as lamproite, and for the Raniganj dykes of the Gondwana coal fields (Mitchell, 2006; Mitchell and Fareeduddin, 2009; Gurmeet Kaur and Mitchell, 2013; Gurmeet Kaur et al., 2013; Gurmeet Kaur and Mitchell, 2016; Shaikh et al., 2016). Further study of the Ramadugu

TABLE 7. Representative compositions (wt.%) of apatites and monazites

	Minerals	VL1	VL2	VL3	VL4	VL5	VL6	VL7	VL8	VL10
1	Phlogopite									
2	Amphibole				•	•	$\dot{}$	√	v	
3	Pyroxene		V.	V.				V.		
4	K-feldspar		V.	V.				V.		
5	Spinel		V.	V.	V.		V.	V.	V.	V.
6	Titanite	V.	V.	V.				V.	V.	V.
7	Apatite	v	v	v	V.	V.	v	v	v	v
8	Monazite				V.	V.		•		
9	Allanite									
10	Rutile	V.	V.	V.	V.	V.	V.	V.	V.	
11	Calcite	V.	V.	V.	V.			V.		
12	Baryte	V.	V.	V.		V.	V.	V.	V.	V.
13	Hydro-zircon	V.	V.	V.	V.		V.	V.	V.	V.
14	Chlorite	V.	V.	V.	V.			V.		V.
15	SiO <sub>2</sub>									
16	Dolomite				V.	V.	•			
17	Magnetite				V.	V.				
18	Pyrite									
19	Co-Ni-sulfide		v	v	·	·	·	·		

TABLE 8. List of minerals present in Vattikod lamproite dykes.

 $\sqrt{-\text{present.}}$ 

lamproite field as a whole, and in conjunction with the Krishna and Cuddapah Basin lamproites and also other peralkaline rocks occurring in the Eastern Dharwar Craton and adjoining Eastern Ghats Mobile Belt is required to establish in greater detail the inter-field mineralogical variation and the evolution of the parental peralkaline magmas in this south-eastern segment of southern India.

We have previously proposed a link between the disposition of 'Deformed Alkaline Rocks and Carbonatites' commonly known as DARC's (Burke and Khan, 2006) of the Eastern Ghats Mobile Belt and the lamproites of the Eastern Dharwar Craton (Fig. 1; Leelanandam et al., 2006; Burke and Khan, 2006; Gurmeet Kaur and Mitchell, 2016). The near-linear disposition of DARC's and lamproites has been interpreted to imply a relationship with ancient subduction-related processes (Fig 1; Das Sharma and Ramesh, 2013; Gurmeet Kaur and Mitchell, 2016). Das Sharma and Ramesh (2013) have reported the presence of relict subducted oceanic slab material at depths of 160-220 km in the subcontinental lithospheric mantle. This subducted oceanic slab is considered to be a product of suturing of the Eastern Dharwar Craton and Eastern Ghats Mobile Belt at ~1600 Ma. This timing is appropriate for the later emplacement of all lamproites in the Eastern Dharwar Craton between 1100-1450 Ma (Gopalan and Kumar, 2008; Osborne et al., 2011; Chalapathi Rao et al., 2013; Chalapathi Rao et al., 2014). Although we have no geochronological data we see no reason why the Vattiokod dykes should not belong to this general period of lamproite magmatism. Clearly, any ancient subducted material, if metasomatized in Proterozoic times, could provide a source for the lamproitic magmatism. Recently, Dongre et al. (2015) have proposed a subduction-related origin for Archaean eclogite xenoliths from the Wajrakarur kimberlite field in the Eastern Dharwar Craton. We note that the extensive near-linear disposition of the east Indian lamproites is not in accord with the ascent of a mantle plume as a mechanism for causing partial melting of potential sources. In conclusion we propose that the Vattikod and other lamproites in eastern India emplaced at 1100-1450 Ma are possible manifestations of ancient subductionrelated alkaline magmatism along the Eastern Ghats Mobile Belt as also has been proposed for 1.2 Ga Krishna lamproites (Fig. 1) in the neighbourhood of Ramadugu lamproites by Chakrabarti et al. (2007) on the basis of Nd-Hf-Pb isotopic characteristics, low SiO<sub>2</sub>, high Mg-numbers, low Al<sub>2</sub>O<sub>3</sub>/TiO<sub>2</sub>, high CaO/Al<sub>2</sub>O<sub>3</sub>, high TiO<sub>2</sub>, high Ni,

Cr, Th/U and Nb/Th–Nb/U ratios. This proposition is in contrast to extension-related anorogenic lamproite magmatism related to supercontinent(s) break-up, as has been suggested for Ramadugu and other Dharwar Craton lamproites (Chalapathi Rao *et al.*, 2014). We do not consider that these rocks are Mediterranean-type lamproites or that they were formed in active subduction zones. We merely speculate that the material involved in the formation of their source regions was ancient subducted material in common with other lamproitic magmas (Mitchell and Bergman, 1991).

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## Supplementary material

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